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Identification of Widespread Pollution in the Southern Hemisphere Deduced from Satellite Analyses

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Vertical profiles of ozone obtained from ozonesondes in Brazzaville, Congo (4°S, 15°E), and Ascension Island (8°S, 15°W) show that large quantities of tropospheric ozone are present over southern Africa and the adjacent eastern tropical South Atlantic Ocean. The origin of this pollution is widespread biomass burning in Africa. These measurements support satellite-derived tropospheric ozone data that demonstrate that ozone originating from this region is transported throughout most of the Southern Hemisphere. Seasonally high levels of carbon monoxide and methane observed at middle- and high-latitude stations in Africa, Australia, and Antarctica likely reflect the effects of this distant biomass burning. These data suggest that even the most remote regions on this planet may be significantly more polluted than previously believed.

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troposphere (1) define distinct plumes emanating from North America, Asia, and Europe. When these data are compared with available ozonesonde measurements at sites with enough data to derive a climatology of free tropospheric O_3 , the differences in agreement between the satellite analysis and the ozonesonde measurements are generally smaller than 15% (1). The satellite measurements also indicate that a large amount of O_3 pollution comes from tropical southern Africa and that this source is most pronounced during the dry season from August to October. In addition, elevated concentrations of carbon monoxide (CO) and meth-

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ane (CH₄) have been observed on a seasonal basis in the Southern Hemisphere. Both CH₄ and CO are trace gases that would be present in the atmosphere even without sources from anthropogenic activity. Because industrialization is less in the Southern Hemisphere than in the Northern Hemisphere, it has generally been assumed that the concentrations of both of these trace gases in the remote regions of this hemisphere are not greatly influenced by pollution. But the source of the seasonality of these trace gases has been unclear. In this report, we compare the data from satellite and in situ measurements to investigate the origin of these trace gases, and we suggest that pollution originating from biomass burning in tropical and subtropical southern Africa has a pronounced influence on the seasonal cycles of CO and CH4 that have been observed throughout the Southern Hemisphere.

More than 32,000 concurrent satellite measurements have been made from the Total Ozone Mapping Spectrometer (TOMS) and the Stratospheric Aerosol Gas Experiments (SAGE) between 1979 and 1989 (Fig. 1) (2). Tropospheric O₃ levels can be obtained by subtracting the amount of O₃ in the stratosphere, derived from the SAGE measurements, from the concurrent



Fig. 2. Ozonesondes from Brazzaville (solid line) and Ascension Island (dashed line) for 6 and 7 August 1990, respectively.

measurement of total O_3 measured by TOMS at the same location. The difference is approximately 10% of the total signal and is referred to as the tropospheric residual (1).

The tropospheric data show a well-defined maximum in O_3 of greater than 45 Dobson units [1 Dobson unit (D.U.) = 2.69×10^{16} molecules of O_3 per square centimeter] off the west coast of southern Africa during September and October. To investigate the origin of these elevated O_3 concentrations, we used ozonesondes (3) at Brazzaville, Congo (4°S, 15°E), and at As-



Fig. 1. Distribution of the tropospheric O_3 residual derived from data between 1979 and 1989 for bimonthly periods.



Fig. 3. The seasonal cycles of CO at Cape Point, Africa (O), and Cape Grim, Australia (Δ), and the integrated amount of O₃ at the same two locations (\bullet and \blacktriangle). Data derived from the satellite technique used to construct Fig. 1.

cension Island (8°S, 15°W) to obtain more accurate O3 measurements (Fig. 2). Partial pressures of O3 greater than 50 nbar were found between 1.5 and 4.3 km in the Brazzaville sounding. The O3 concentration peaks at 66 nbar at an altitude of 2.2 km. The tropospheric O₃ distribution was generally similar at both localities, but the concentrations in the polluted layer at Ascension Island, where the highest tropospheric O₃ partial pressure was 56 nbar at 3.5 km, were typically 15 to 20% less than the concentrations in Brazzaville. These profiles substantiate our conclusion from satellite data that widespread air pollution is the reason for the enhancement observed over the western part of southern Africa. During the 1990 dry season, the average integrated amount of O_3 in the troposphere was 45 D.U., and the level at Brazzaville was ~8 D.U. higher than at Ascension Island.

The distribution of O_3 in the lower troposphere at Ascension Island supports the premise that the O₃ produced over southern Africa is transported by low-level easterlies to the eastern South Atlantic Ocean. Some of the O₃ in the boundary layer makes its way into the middle and upper troposphere, where the winds in the tropics and in the subtropics are generally westerlies (4). Because O₃ is long-lived in the free troposphere, it can be transported long distances. The satellite data for July-August and October-September (Fig. 1) show the long-range transport of this O3. The prevailing meteorology during November and December suggests that O₃ from both southern Africa and southeastern Brazil should feed into the region of elevated O₃ (>40 D.U.) over the South Atlantic Ocean near 25° to 30°S.

Examination of the annual cycles of CO and CH₄ at four locations in the Southern Hemisphere [Cape Point, South Africa (34°S, 18°E), Cape Grim, Australia (41°S, 145°W), Mawson, Antarctica (68°S, 63°E), and the South Pole] shows that the concentrations of these trace gases also exhibit a



Fig. 4. The seasonal cycles of surface tropospheric O_3 for Cape Point (34°S) (\bullet) and Cape Grim (41°S) (\bigcirc).

distinct maximum during September and October (5, 6). Comparison of the seasonal cycles of CO and integrated O₃ in the troposphere at Cape Grim and Cape Point (Fig. 3) shows that both were enhanced during austral spring, September to November. The CO data are monthly averages; the O₃ data have been averaged over 45-day periods for 10° latitude by 20° longitude boxes centered on Cape Grim and Cape Point. The orbit of the satellites that carry the SAGE instruments is such that the same approximate location is sampled at intervals of ~40 days. Therefore, meaningful monthly averages cannot be obtained for every month.

The detrended CH₄ seasonal cycle at Cape Point and Cape Grim (7) is virtually identical to the seasonal cycle of CO; both have an amplitude of ~ 30 ppb by volume (ppbv); the highest concentrations were in September to October and the lowest concentrations were in February. In contrast, in the Northern Hemisphere, the seasonal cycles of CH_4 and CO are quite different (8). Because the atmospheric lifetime of CH_4 is on the order of a decade, its seasonal cycle most likely reflects the seasonal variation of its sources rather than the seasonality of its removal mechanism, which is reaction with the hydroxyl (OH) radical. The primary sources of CH4 are biomass burning, rice paddies and marshes, and enteric fermentation from ruminants (9). Each of these sources is estimated to release on the order of 1×10^{14} g of CH₄ per year. Only biomass burning is significantly concentrated in southern low latitudes.

In contrast to the cycle of CH_4 , the seasonal cycle of CO at these two sites partially reflects the seasonal nature of its removal by the OH radical. The atmospheric residence time of CO is 1 to 2 months, and the lowest OH concentrations at southern mid-latitudes are June and July. Therefore, the highest CO concentrations at southern mid-latitudes should occur in July or August, rather than October as observed. Results from a three-dimensional photochemical transport model (10) show that biomass burning, rather than the seasonal

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variation of the OH radical at southern mid-latitudes, is the primary reason why the maximum observed at Cape Point and at Cape Grim occurs in October. Without seasonal input from biomass burning, the model yields a seasonal CO cycle that peaks in August and an average CO concentration in the Southern Hemisphere of only \sim 35 ppbv, lower by nearly a factor of 2 than the observations (Fig. 3). The amplitude of the CO seasonal cycle without biomass burning is only \sim 11 ppbv instead of the observed \sim 30 ppbv.

Satellite measurements of CO from two Space Shuttle flights in November 1981 and October 1984 (11) confirmed that large amounts of CO were coming from Africa as a result of widespread biomass burning. Subsequent analysis (4) of these data showed that CO over the eastern Indian Ocean was transported more than 7500 km to the southeast (almost to Australia). We propose that such long-range transport is not an occasional occurrence, but a common one that leads to the widespread dissemination of CO, CH₄, and tropospheric O₃ from biomass burning. The seasonal cycle of elemental carbon (C) at Cape Grim also maximizes in October and supports this premise (12)

At both Cape Point and Cape Grim (Fig. 4), the highest concentration of surface O_3 (13) is found in July and the lowest concentration in January. This pattern is significantly different from the seasonal cycles of CO and CH₄ at the surface and the seasonal cycle of the satellite-derived tropospheric O₃. To explain this apparent discrepancy, we compared the satellite-derived tropospheric O₃ data with the seasonal cycle of O_3 (at 500 mbar) derived from an analysis of 752 ozonesondes at Aspendale, Australia (38°S, 145°E), between 1965 and 1982 (Fig. 5) (14). The satellite observations are in good agreement with the free tropospheric O₃ measurements. Earlier workers had assumed that middle tropospheric O₃ reached its highest concentrations in austral



Fig. 5. The seasonal cycles of O_3 at 500 mbar (Δ) derived from ozonesonde data at Aspendale, Australia, between 1967 and 1982, and the tropospheric residual (from satellite data) (\blacktriangle) and surface O_3 measurements (\bigcirc) at Cape Grim, Australia. Surface and 500-mbar data reported in parts per billion by volume, satellite data in D.U.

spring, because this was when stratospheretroposphere exchange was most intense (15). The spatial continuity provided by the satellite analysis, on the other hand, clearly shows that the excess O_3 at southern middle latitudes at this time of the year is transported from lower latitudes, not higher latitudes, and is thus from biomass burning, not the stratosphere.

We propose that the observed differences between the seasonal behavior of surface O₃ and the other trace species are related to the relatively short atmospheric residence time of surface O₃ compared to those of these other gases. The dominant sink for both CO and CH₄ is reaction with OH. If we assume that these two trace gases originate in the southern tropical and subtropical regions of the Southern Hemisphere, then the atmospheric residence time of these trace gases can be obtained if the OH concentration is known. Using the model-derived OH distribution computed in (10), we determined the atmospheric lifetime of CO to be approximately 30 days in the boundary layer and 40 days in the middle troposphere at these latitudes. For CH4, the atmospheric lifetime is approximately 3 years in the boundary layer and 8 years in the middle troposphere. These lifetimes are sufficiently long that CO and CH₄ produced in these regions can be transported substantial distances.

Ozone, however, is a different story. The primary photochemical sink for tropospheric O_3 is the reaction of metastable atomic oxygen, $O(^{1}D)$, one of the products of O_{3} photolysis, with water vapor. Thus, the lifetime of tropospheric O_3 is most dependent on the distribution of water vapor and the amount of ultraviolet radiation available. In the tropical boundary layer, both of these are abundant, and we compute (16) a lifetime of only 2 to 5 days in the tropical planetary boundary layer. In the middle troposphere, however, we compute an atmospheric lifetime of approximately 90 days. Thus, the O3 that is so abundant in the lower atmosphere over Brazzaville and Ascension Island (Fig. 2) does not persist long enough to be transported long distances. If, however, the O3 can get out of the planetary boundary layer, the lifetime is sufficiently long that it can be transported significant distances. The seasonal cycle of O3 at the surface may thus be controlled directly by the strength of the sink at these latitudes.

The data do not rule out other factors that contribute to the seasonal variability of the trace gas measurements summarized in this report. The agreement between the seasonal cycles of the satellite-derived tropospheric O_3 data and the surface CO and CH₄ measurements is not perfect, and other sources, sinks, and transport mechanisms are likely responsible for these observed differences in seasonal cycles. Nonetheless, we believe that the enhancement of trace gas concentrations in the Southern Hemisphere that is due to emissions from biomass burning in Africa and, to a lesser extent, South America, is the mechanism that determines much of the seasonality of these trace species.

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Identity Elements for Specific Aminoacylation of Yeast tRNA^{Asp} by Cognate Aspartyl-tRNA Synthetase

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The nucleotides crucial for the specific aminoacylation of yeast tRNA^{Asp} by its cognate synthetase have been identified. Steady-state aminoacylation kinetics of unmodified tRNA transcripts indicate that G34, U35, C36, and G73 are important determinants of tRNA^{Asp} identity. Mutations at these positions result in a large decrease (19- to 530-fold) of the kinetic specificity constant (ratio of the catalytic rate constant k_{cat} and the Michaelis constant K_m) for aspartylation relative to wild-type tRNA^{Asp}. Mutation to G10-C25 within the D-stem reduced k_{cat}/K_m eightfold. This fifth mutation probably indirectly affects the presentation of the highly conserved G10 nucleotide to the synthetase. A yeast tRNA^{Phe} was converted into an efficient substrate for aspartyl-tRNA synthetase through introduction of the five identity elements. The identity nucleotides are located in regions of tight interaction between tRNA and synthetase as shown in the crystal structure of the complex and suggest sites of base-specific contacts.

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tRNAs by their cognate synthetase is crucial for accurate transmission of genetic information and is determined by certain structural features of the tRNA, which in certain systems include nucleotides in the anticodon loop, acceptor stem, and D-loop (1). Regions of contact between yeast tRNA^{Asp} and yeast aspartyl-tRNA synthetase (AspRS) have been previously characterized with chemical and enzymatic footprinting methods (2, 3). A high-resolution x-ray structure of this complex (4) has confirmed that the anticodon loop and stem as well as portions of the acceptor stem are sites of interaction with AspRS. We describe steady-state aminoacylation kinetics for a series of mutant transcripts of yeast tRNA^{Asp} in order to delineate the determinant nucleotides important for aminoacylation by yeast AspRS.

The aminoacylation kinetics of mutant tRNAs were compared to that of the unmodified transcript of yeast tRNAAsp in which U1-A72 base pair was changed to G1-C72 (Fig 1A); both transcripts have equivalent kinetic parameters for aspartylation as the fully modified molecule (5). Since the transcript of Escherichia coli tRNAAsp is an equivalent substrate for aspartylation by yeast AspRS as the yeast transcript (Table 1), a number of nucleotide positions could be eliminated as potential identity elements (Fig. 1B). Moreover, single-stranded nucleotides protected in footprinting experiments (2, 3), G-U base pairs (6), and nucleotides identified by computer sequence analysis (7) as specific for tRNA^{Asp} were tested explicitly. For simplicity, the effects of mutations (Fig. 1C) on the steady-state aminoacylation kinetics (Table 1) are described below in terms of the four structural domains of the tRNA molecule.

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