# Metabolic Activity of Subsurface Life in Deep-Sea Sediments

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Global maps of sulfate and methane in marine sediments reveal two provinces of subsurface metabolic activity: a sulfate-rich open-ocean province, and an ocean-margin province where sulfate is limited to shallow sediments. Methane is produced in both regions but is abundant only in sulfate-depleted sediments. Metabolic activity is greatest in narrow zones of sulfate-reducing methane oxidation along ocean margins. The metabolic rates of subseafloor life are orders of magnitude lower than those of life on Earth's surface. Most microorganisms in subseafloor sediments are either inactive or adapted for extraordinarily low metabolic activity.

During the past 15 years, studies of Ocean Drilling Program (ODP) cores have consistently identified abundant prokaryotes in deeply buried oceanic sediments (1). Microorganisms have been recovered from depths as great as 800 m below the seafloor (mbsf) (2). The potential for in situ activity of subseafloor microorganisms has been demonstrated by geochemical (3) and radiotracer (1, 4) experiments on sediments recovered from a range of burial depths. In recent contamination-tracer experiments, most of the microorganisms reported from ODP cores were inherent to the drilled sediments (5).

The number and mass of prokaryotes in subseafloor sediments have been estimated by extrapolation from direct counts of sedimentary microorganisms at a small number of ODP sites (6, 7). On the basis of that extrapolation, these prokaryotes constitute one-tenth (6) to one-third of Earth's biomass (7). In situ metabolic activity by at least some of these prokaryotes is spectacularly demonstrated by hydrates of methane  $(CH_{4})$  produced in deep-sea sediments. These hydrates contain four to eight times the amount of carbon in Earth's surface biomass and terrestrial soils combined (8). Porewater chemical studies (9) and recent microbiological discoveries (10, 11) suggest that, on an ongoing basis, the CH<sub>4</sub> produced in deep-sea sediments may be primarily destroyed by sulfatereducing activity of microorganisms in overlying sediments.

The activity of subseafloor microorganisms may directly affect the surface Earth. Intermittent release of  $CH_4$  from marine sedimentary hydrates may have greatly affected the global climate and/or oceans several times in Earth history [e.g., (12)]. And relatively small changes in subsurface sulfate  $(SO_4^{2-})$  reduction may have appreciably affected total oceanic alkalinity and, consequently, the partitioning of  $CO_2$  between atmosphere and ocean over geologic time (13). Despite the large apparent mass and possible biogeochemical effects of subseafloor life, the magnitude of its metabolic activity in situ remains largely unknown. To assess this activity, we have compiled and analyzed sedimentary porewater chemical data from Deep Sea Drilling Project (DSDP) and ODP sites throughout the world ocean (DSDP Leg 1 through ODP Leg 182) (14).

We can infer that  $SO_4^{2-}$  reduction, methanogenesis (CH<sub>4</sub> production), and fermentation are the principal degradative metabolic processes in subsurface (>1.5 mbsf) marine sedi-

ments, for three reasons: (i) At the sedimentwater interface, concentrations of dissolved  $SO_4^{2-}$  are more than 50 times as great as concentrations of all electron acceptors with higher standard free energies combined (15, 16). (ii) External electron acceptors that yield more energy than  $SO_4^{2-}$  typically disappear within the first few centimeters to tens of meters of sediment depth (15). (iii) Once all  $SO_4^{2-}$  has been reduced, methanogenesis and fermentation are the principal remaining avenues of metabolic activity.

The concentration of dissolved  $SO_4^{2-}$  in typical marine sediments results from the balance between diffusion of  $SO_4^{2-}$  from the overlying ocean and reduction of  $SO_4^{2-}$  by microbial activity in the sediments. Peak  $SO_4^{2-}$  concentrations typically occur at the sediment-water interface (Fig. 1). Sulfate concentrations are generally stable over stratigraphic intervals where little  $SO_4^{2-}$  reduction occurs (Fig. 1). At sites where subsurface microbial activity is consistently low,  $SO_4^{2-}$  concentrations are relatively high throughout the sediment column (Fig. 1, A and C). Where subsurface activity is relatively high, all dissolved  $SO_4^{2-}$  is consumed in the upper sediment column and microorganisms in deeper sediments produce CH<sub>4</sub>, which diffuses up toward the sulfate-rich sediments (Fig. 1B). Consequently, peak CH<sub>4</sub> concentrations usually occur in sulfate-depleted sediments well below the seafloor (Fig. 1B). At sites where upward-diffusing CH<sub>4</sub> is entirely consumed by sulfate-reducing methane oxidation (9),  $CH_4$  approaches laboratory-background concentrations within the sulfate reduction zone (Fig. 1, B and C).



**Fig. 1.** Representative profiles of  $SO_4^{2-}$  (open squares) and  $CH_4$  (solid circles) concentrations in deep-sea sediments: (**A**) open-ocean ODP Site 851, (**B**) ocean-margin ODP Site 798, (**C**) open-ocean ODP Site 846.

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To identify geographic patterns of  $SO_4^{2-}$ reduction and methanogenesis in deep-sea sediments, we created global maps of  $SO_4^{2-}$ and CH<sub>4</sub> concentration data from DSDP and ODP cores (Fig. 2). The maps exhibit two broad patterns in subsurface  $SO_4^{2-}$  and  $CH_4$ concentrations. First, in open-ocean sediments, dissolved  $SO_4^{2-}$  concentrations are high and CH<sub>4</sub> concentrations are low throughout the entire sediment column (Fig. 2). Second, along ocean margins, dissolved  $SO_4^{2-}$  concentrations are essentially reduced to zero within a few tens of mbsf (Fig. 2A). Below this depth, CH<sub>4</sub> concentrations are high because microbial activity is generally limited to fermentation and methanogenesis (Fig. 2B).

The continuously high  $SO_4^{2-}$  concentrations and low CH<sub>4</sub> concentrations of openocean sites indicate that rates of SO<sub>4</sub><sup>2-</sup> reduction and CH<sub>4</sub> production are very low in subsurface open-ocean sediments. In contrast, the general absence of  $SO_4^{2-}$  below a few tens of mbsf at ocean-margin sites indicates that the rate of subsurface  $SO_4^{2-}$  reduction is high relative to the rate of downward  $SO_4^{2-}$  diffusion in ocean-margin sediments. The absence of high CH<sub>4</sub> concentrations in the shallowly buried sulfate-rich zone of ocean-margin sites suggests that subsurface sulfate-reducing methane oxidation is the primary means of CH<sub>4</sub> destruction throughout the world ocean.

About one-sixth of all open-ocean sites (32 of 184 sites) contain abundant dissolved  $SO_4^{2-}$  and above-background  $CH_4$  concentrations (~10 to 1000 ppm) (Fig. 2, A and B). These are typically sites with cool (<20°C) organic-poor open-ocean sediments that lack strong subsurface flow and lie hundreds to thousands of km from the organic-rich methanogenic zones of the oceanic margins (17). In general,  $CH_4$  profiles mirror  $SO_4^{2-}$ profiles at these sites (Fig. 1C); CH<sub>4</sub> concentrations are stable and relatively high over intervals of stable SO<sub>4</sub><sup>2-</sup> concentrations, but they decline to background values in the shallowly buried sulfate reduction zone. This finding indicates that methanogenesis commonly occurs in sulfate-rich open-ocean sediments. This occurrence conflicts with the generally accepted redox sequence, based on standard-state free energies of reaction (18), which predicts that sulfate will be depleted before methanogenesis occurs.

The occurrence of methanogenic activity in sulfate-rich open-ocean sediments may result from one or more causes. The methanogens and the sulfate-reducing microorganisms may rely on different electron donors (19). The relative in situ free energies of  $SO_4^{2-}$  reduction and  $CO_2$  reduction may deviate from standard-state free energies. Ecological properties, such as relative predation susceptibility, may allow methanogens to



**Fig. 2.** (A) Global  $SO_4^{2-}$  map. Symbol color indicates the depth of the shallowest sample where subsurface  $SO_4^{2-}$  concentrations either stabilize at a nonzero value (circles) or reach zero (diamonds) (white,  $\leq 5$  mbsf). Circle size indicates the  $SO_4^{2-}$  concentration at which each subsurface  $SO_4^{2-}$  profile stabilizes. (B) Global CH<sub>4</sub> map. Circle color indicates the depth of the shallowest sample that exhibits a CH<sub>4</sub> concentration above the laboratory background level (white,  $\leq 5$  mbsf). Circle size indicates the peak subsurface concentration. Crosses mark sites where CH<sub>4</sub> never rises above laboratory background. Cross color marks the depth of the deepest sample analyzed (36).

persist in an environment energetically more favorable to sulfate reducers. Methanogens may be adapted to survive at lower rates of dissolved electron donor production.

In sedimentary ecosystems where  $SO_4^{2-}$ is the terminal electron acceptor (e.g.,  $3SO_4^{2-}$ +  $C_6H_{12}O_6 \rightarrow 3S^{2-} + 6CO_2 + 6H_2O$ ), the depth-integrated rate of  $SO_4^{2-}$  reduction is a direct measure of total dissimilatory activity. At open-ocean sites, where  $CH_4$  concentrations are low, the dissimilatory activity of subsurface microorganisms can be approximated by the flux of dissolved  $SO_4^{2-}$  into the sediment (20). Along ocean margins, where subsurface  $CH_4$  is destroyed by sulfate-reducing methane oxidation at the  $CH_4$ - $SO_4^{2-}$ interface, total subsurface dissimilation can also be approximated by the flux of dissolved

 $SO_4^{2-}$  into the sediment. In such regions, degradative metabolic activity in sediments deeper than the sulfate reduction zone is ultimately methanogenic (e.g.,  $C_6H_{12}O_6 \rightarrow$  $3CH_4 + 3CO_2$ ). Relative to  $CH_4$ , appreciable concentrations of hydrogen (21) and dissolved organic products of fermentation (22) do not occur at the  $CH_4$ - $SO_4^{2-}$  interface or in the upper portion of the methane-rich zone. In these subsurface environments,  $SO_4^{2-}$  serves as the terminal electron acceptor for the ecosystem as a whole, because the CH<sub>4</sub> is consumed by  $SO_4^{2-}$  reduction as it diffuses up into the sulfate-rich zone (SO<sub>4</sub><sup>2-+</sup> CH<sub>4</sub>  $\rightarrow$  $S^{2-} + CO_2 + 2H_2O$ ). The diffusive flux of CH<sub>4</sub> upward can be estimated by assuming that all of the  $SO_4^{2-}$  reduced in the  $CH_4$ - $SO_4^{2-}$  transition zone is used for methane

<b>Table 1.</b> Rates of biological activity at representative sites (23). n/a indicates that SO <sub>4</sub> <sup>2</sup>	<sup>2-</sup> is present throughout the drilled hole and, consequently, that there
is no base to the $SO_4^{2-}$ reduction zone in the sampled sediment column.	· · ·

Location	Base of SO₄ <sup>2−</sup> reduction zone (mbsf)	SO4 <sup>2-</sup> flux downward to subsurface (mol cm <sup>-2</sup> year <sup>-1</sup> )	$SO_4^{2-}$ flux due to CH <sub>4</sub> oxidation (% of total)	Total subsurface respiration (mol CO <sub>2</sub> cm <sup>-2</sup> year <sup>-1</sup> )	Photosynthesis of overlying ocean (mol CO <sub>2</sub> reduced cm <sup>-2</sup> year <sup>-1</sup> )
		Ocea	n-margin sites		· · · · · · · · · · · · · · · · · · ·
798B (Japan Sea)	10	4.2 ( $\pm$ 1.3) $ imes$ 10 $^{-6}$	80	8.4 (±2.6) × 10 <sup>-6</sup>	$2.3  imes 10^{-3}$
681C (Peru Margin)	30	8.1 (±2.5) × 10 <sup>-7</sup>	85	1.6 (±0.5)́ × 10 <sup>−6</sup>	$4.9 imes10^{-3}$
1175 (Nankai Trough)	11	1.3 (±0.6)́ × 10 <sup>−6</sup>	43	2.6 (±1.2)́ × 10 <sup>−6</sup>	$1.8  imes 10^{-3}$
		Ope	n-ocean sites		
851 (Equatorial Pacific)	n/a	2.8 (±2.5) × 10 <sup>−9</sup>	0	5.6 (±5.0) × 10 <sup>-9</sup>	9.3 × 10 <sup>−4</sup>
834 (Lau Basin)	n/a	2.0 (±2.2) × 10 <sup>-9</sup>	0	$4.0(\pm 4.4) \times 10^{-9}$	6.7 × 10 <sup>−</sup>
1149 (Izu-Bonin Trench)	n/a	1.3 (±0.6)́ × 10 <sup>-8</sup>	0	$2.6(\pm 1.2) \times 10^{-8}$	4.8 × 10 <sup>−4</sup>

**Table 2.** Sulfate reduction per subsurface cell at representative sites (27). n/a indicates that there is no discernible subsurface methane oxidation zone at the site.

Location	Mean SO4 <sup>2-</sup> reduction per cell in CH4-depleted SO4 <sup>2-</sup> -rich zone (mol cell <sup>-1</sup> year <sup>-1</sup> )	Mean SO4 <sup>2-,</sup> reduction per cell in anaerobic methane oxidation zone (mol cell <sup>-1</sup> year <sup>-1</sup> )
	Ocean-margin sites	
798B (Japan Sea)	$3.1(\pm 1.3) \times 10^{-17}$	2.6 (±1.1) × 10 <sup>-16</sup>
681C (Peru Margin)	1.4 (±0.6) × 10 <sup>-19</sup>	6.4 (±2.7)́ × 10 <sup>−18</sup>
1175 (Nankai Trough)	$3.3(\pm 1.4) \times 10^{-15}$	7.5 (±3.8) × 10 <sup>−15</sup>
	Open-ocean sites	
851 (Equatorial Pacific)	$1.2(\pm 1.1) \times 10^{-19}$	n/a
834 (Lau Basin)	$1.5(\pm 1.7) \times 10^{-20}$	n/a
1149 (Izu-Bonin Trench)	9.5 (±5.8)́ × 10 <sup>-20</sup>	n/a

oxidation. If the upward diffusion of  $CH_4$ into the  $CH_4$ - $SO_4^{2-}$  transition zone is in a steady-state balance with the rate of  $CH_4$ production below, then (i) the total rate of microbial activity within the methanogenic zone can be directly estimated from the rate of  $SO_4^{2-}$  reduction in the transition zone, and (ii) the total respiration of the entire sediment column is approximated by the downward flux of  $SO_4^{2-}$  into the sediment column.

flux of  $SO_4^{2^-}$  into the sediment column. We have determined the downward flux of dissolved  $SO_4^{2^-}$  at six representative ODP sites (Table 1) (23). Total rates of subsurface  $SO_4^{2^-}$  reduction are at least two to three orders of magnitude higher at the methane-rich ocean-margin sites than at the sulfate-rich open-ocean sites (Table 1). At the open-ocean sites, total rates of subseafloor  $SO_4^{2^-}$  reduction are so low as to be nearly indistinguishable from zero (Table 1). This finding of a large difference between  $SO_4^{2^-}$  reduction rates at ocean-margin sites and those at open-ocean sites is consistent with an earlier study of DSDP sites (24), which found that  $SO_4^{2^-}$  reduction can vary by a factor of more than 1000 from one marine environment to another.

On a more detailed level, these flux calculations suggest that subseafloor metabolic activity is greatly concentrated in relatively narrow zones of anaerobic methane oxidation along ocean margins. At the ocean-margin sites, most of the  $SO_4^{2-}$  flux downward past 1.5 mbsf is used to oxidize CH<sub>4</sub> created in the underlying sediments (Table 1). Recent studies have similarly shown that in sediments along the continental slope of Namibia (25) and in sediments of the Amazon Fan (26), nearly 100% of the downward  $SO_4^{2-}$  flux goes to anaerobic CH<sub>4</sub> oxidation. Comparison of these ocean-margin fluxes to the  $SO_4^{2-}$  fluxes at open-ocean sites (Table 1) suggests that anaerobic methane oxidation may be the dominant sink for  $SO_4^{2-}$  in marine sediments. Consequently, it may have a considerable long-term effect on oceanic alkalinity.

In terms of carbon content, the estimated mass of subsurface microorganisms in marine sediments (3  $\times$  10<sup>17</sup> g C) is two orders of magnitude greater than the mass of living organisms in the overlying ocean ( $4 \times 10^{15}$  g C) (7). Despite its large inferred size, the annual metabolic activity of this subsurface world is extremely low relative to the annual metabolic activity in the overlying ocean. At the ocean-margin sites, the annual rate of biomass production in the sunlit water of the ocean surface is two or more orders of magnitude greater than the annual rate of subsurface respiration in the underlying sediments (Table 1). And at the open-ocean sites, the rate of oceanic biomass production is four or more orders of magnitude greater than the rate of respiration in the underlying sediments (Table 1).

The mean respiration per enumerated subsurface cell can be calculated from the flux estimates of Table 1 and previously published cell counts (27) (Table 2). Mean respiration per cell is highest in the anaerobic methaneoxidation zones of the ocean-margin sites and lowest in the methane-poor sulfate reduction zones of the open-ocean sites. Most rates of  $SO_4^{2-}$  reduction in Table 2 are orders of magnitude lower than per-cell rates of  $SO_{4}^{2-}$ reduction exhibited by pure sulfate-reducing bacteria cultures or radiotracer experiments with coastal marine sediments. Per-cell rates measured for pure cultures range from 7  $\times$  $10^{-14}$  to  $1 \times 10^{-11}$  mol SO<sub>4</sub><sup>2-</sup> cell<sup>-1</sup> year<sup>-1</sup> (28, 29). Per-cell rates calculated from absolute cell numbers and radiotracer-based SO42- reduction rates of marine coastal nearsurface ( $\leq 10$  cm below seafloor) sediments (30) are in the range of 1  $\times$  10  $^{-15}$  to 2  $\times$  $10^{-15} \text{ mol SO}_4^{2-} \text{ cell}^{-1} \text{ year}^{-1} (31).$ 

These comparisons suggest that very little adaptation to low metabolic activity is necessary for most enumerated subsurface microorganisms to be active in the anaerobic methane-oxidation zone at the sites of highest activity (1175 and 798B). They also indicate that most of the subseafloor microorganisms enumerated in most sediment at most sites must be either inactive or adapted for extraordinarily low metabolic activity.

#### **References and Notes**

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results in the production of two equivalents of alkalinity per mole.

- 14. Sulfate is routinely accurately measured on ODP cruises. In contrast, because sediments outgas as cores are brought up through the water column, concentrations of CH, in deep-sea porewaters cannot routinely be accurately estimated from ODP CH<sub>4</sub> concentration data (headspace analysis). This quality of the  $CH_4$  data does not preclude using these  $SO_4^{2-}$  and  $CH_4$  data (i) to map downhole profiles of relative  $CH_4$  abundance, or (ii), as described in the text, to estimate steady-state rates of sulfate-reducing methane oxidation and (by inference)  $CH_4$  production. The DSDP and ODP ship-board data used for these profiles and maps were edited to remove samples affected by seawater contamination and sites where spot coring, poor core recovery, or intermittent porewater sampling caused large gaps that rendered profiles difficult to interpret. Sites with fewer than three samples in the zone of stable  $SO_4^{2-}$  concentrations were excluded from the  $SO_4^{2-}$  map. Because we limited our study to the diffusional realm typical of open-ocean sediments, this analysis omitted stratigraphic intervals from downhole  $SO_4^{2-}$  and  $CH_4$ records in cases where porewater chemical profiles were unambiguously affected by hydrologic flow or pronounced lithologic breaks, subsurface anhydrite deposits, or sulfate-enriched brines. Sites identified as probably affected by CaSO<sub>4</sub> precipitation in the underlying basement were also deleted from the SO<sup>2-</sup> map.
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## Demographic Characteristics and Population Dynamical Patterns of Solitary Birds

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In birds and many other animals, there are large interspecific differences in the magnitude of annual variation in population size. Using time-series data on populations of solitary bird species, we found that fluctuations in population size of solitary birds were affected by the deterministic characteristics of the population dynamics as well as the stochastic factors. In species with highly variable populations, annual variation in recruitment was positively related to the return rate of adults between successive breeding seasons. In stable populations, more recruits were found in years with low return rates of breeding adults. This identifies a gradient, associated with the position of the species along a "slow-fast" continuum of life history variation, from highly variable populations with a recruitment-driven demography to stable, strongly density-regulated populations with a survival-restricted demography. These results suggest that patterns in avian population fluctuations can be predicted from a knowledge of life-history characteristics and/or temporal variation in certain demographic traits.

One of the challenges in ecology is to identify characteristics that can be used to predict interspecific differences in patterns of population fluctuations (1). Comparisons covering a wide range of taxa have shown a strong pattern of covariation of life history traits that divide species along a "slow-fast continuum" (2-6). Life history characteristics such as early onset of reproduction, rapid ontogenetic development, and large litter sizes are typical for species at one end of this continuum, whereas species with low reproductive rates, but longer life expectancies, are found at the other end. Several hypotheses have been proposed to explain this covariation among life

\*To whom correspondence should be addressed. Email: bernt-erik.sather@chembio.ntnu.no history traits, e.g., density-dependent r-K selection (3), adaptive life history responses to differences in extrinsic mortality (5), or adaptations to variation in predictability or variability of the habitats (7). Few studies have, however, quantitatively examined how characteristics of the population dynamics are related to the species' position along this continuum of life history variation; an exception was Fowler (8, 9), who showed that the pattern of density regulation was related to the rate of increase per generation. The presence of such patterns will enable characterization of patterns in population fluctuations from knowledge of basic demography or life history characteristics. Here we used data on the fluctuations of solitary bird populations to examine stochastic effects on population fluctuations. This enables us to quantitatively relate patterns in population dynamics to the position of the species along the slow-fast continuum of life history variation.

To examine how interspecific variations in population dynamical characteristics are affect-

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