where α_n and α_s are proportional to electronic (kinetic) energies in the normal and SC states. In our experiments, ρ_s and $[N_n - N_s]$ are obtained independently (28). Therefore, the inequality $\rho_s > [N_n - N_s]$ (Fig. 3) indicates that kinetic energy change associated with the SC transition may account for the discrepancy in spectral weight.

The change of the electronic kinetic energy at $T < T_{\rm c}$ suggested by our data should be contrasted with the behavior of conventional superconductors where this effect is negligibly small. Moreover, in metallic superconductors $\rho_{\rm s} - [N_{\rm p} - N_{\rm s}]$ ought to be negative, consistent with the experimental data for lead films (29). At least two models proposed for high- T_c superconductors (17, 18) predicted the correct sign of the effect but expected it to be dominant in the response of the CuO₂ planes. The interlayer tunneling (ILT) theory (5, 8, 9) predicted the $\rho_{\rm s} > [N_{\rm n} - N_{\rm s}]$ inequality found in the c-axis transport, but the absolute value of ρ_s in Tl2201 is smaller (7, 20) than is expected within the ILT model (5, 8). Because change in the interlayer kinetic energy has been detected in several classes of high- T_c superconductors, we believe that this unusual effect will be instrumental in narrowing the field of plausible theoretical models of high- $T_{\rm c}$ superconductivity.

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minishing of $\sigma_1(\omega)$ below T_c can be attributed at least partially to the same process that leads to transfer of the spectral weight to higher energies in the pseudogap state at $\mathit{T} > \mathit{T}_{\rm c}.$ Alternatively, one would be forced to assume that development of the pseudogap is suddenly interrupted at $T = T_c$; this conflicts with studies of thermodynamic properties and of nuclear magnetic resonance, both of which display continuous behavior in the underdoped cuprates across T_c [J. W. Loram et al., Phys. Rev. Lett. 71, 1740 (1993); W. W. Warren Jr. et al., ibid. 62, 1193 (1989); R. E. Walstedt et al., Phys. Rev. B 41, 9574 (1990)]. Therefore, a 50% contribution of mid-IR frequencies to the spectral weight of the SC δ function should be regarded as a lower limit because arguments based on the continuity of the pseudogap development at T_c appear to indicate even stronger discrepancy between $\rho_{\rm s}$ and $[N_{\rm n}$ – $N_{\rm s}]$, at least in the La214 and Y123 compounds. Our room temperature data for Tl2201 do not extend below 60 cm⁻¹, and the evolution of the electronic spectral weight at $T > T_c$ in this material requires further study.

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Imaging Electron Wave Functions of Quantized Energy Levels in Carbon Nanotubes

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Carbon nanotubes provide a unique system for studying one-dimensional quantization phenomena. Scanning tunneling microscopy was used to observe the electronic wave functions that correspond to quantized energy levels in short metallic carbon nanotubes. Discrete electron waves were apparent from periodic oscillations in the differential conductance as a function of the position along the tube axis, with a period that differed from that of the atomic lattice. Wave functions could be observed for several electron states at adjacent discrete energies. The measured wavelengths are in good agreement with the calculated Fermi wavelength for armchair nanotubes.

Carbon nanotubes are molecular wires that exhibit fascinating electronic properties (1). Electrons in these cylindrical fullerenes are confined in the radial and circumferential

directions and can only propagate in the direction of the tube axis. Nanotubes are therefore interesting systems for studying the quantum behavior of electrons in one dimen-

sion (1D). Limiting the length of a carbon nanotube leads to a "particle-in-a-box" quantization of the energy levels. Such discrete energy levels have been observed in transport experiments on individual nanotubes and ropes (2, 3). The electron wave functions corresponding to these discrete states can in principle be imaged by scanning tunneling microscopy (STM). The well-known STM work on quantum corrals demonstrated that wave patterns could be directly imaged in the local density of states of a 2D metal surface (4). Here, we applied this technique to map out the wave functions of single molecular orbitals in short metallic carbon nanotubes. Electronic wave functions were apparent from periodic oscillations in the low-bias differential conductance along the tube axis.

Previous STM spectroscopy studies were done at a large (~ 2 eV) energy scale to investigate the band structure of nanotubes (5, 6). These experiments confirmed the prediction (7) that carbon nanotubes can be semiconducting or metallic, depending on the tube diameter and the chiral angle between the tube axis and hexagon rows in the atomic lattice. Here, our focus is on the low-energy (~0.1 eV) features of short metallic nanotubes that exhibit quantum size effects. Single-wall nanotubes with a diameter of ~ 1.4 nm were deposited on Au(111) substrates (5, 8). On most tubes, we were able to obtain STM images with atomic resolution (5), which allowed us to determine the chiral angle and diameter of the tubes (9). The nanotube in Fig. 1B is identified as an armchair tube by the good fit between the observed hexagon structure and the overlay of the graphene lattice. Armchair tubes have a nonchiral structure because the hexagon rows are parallel to the tube axis. This type of tube has metallic properties (7). Current-voltage (I-V) characteristics measured up to ± 0.5 V on the armchair tube in Fig. 1B indeed demonstrate the simple linear behavior expected for a metallic tube. Such I-V measurements are done by keeping the STM tip stationary above the nanotube, switching off the feedback, and recording the current as a function of the voltage applied to the sample. In all our experiments, the STM was operated at 4.2 K (10).

The armchair tube in Fig. 1B was shortened to a length of \sim 30 nm by locally cutting the tube; this was done by applying a voltage pulse of +5 V to the STM tip at a position 30 nm from the end of the tube (11). STM spectroscopy was then carried out near the middle of the short tube. *I-V* curves for the shortened nanotube show a step-like behavior (Fig. 2A), which we ascribe to quantum size effects. Steps in I-V correspond to quantized energy levels entering the bias window when the voltage is increased. Current steps at a voltage V thus correspond to discrete electron states at energy $E = E_F + \alpha eV$, where E_F is the Fermi energy, e is the electron charge, and $\alpha \simeq 1$ (12). The experimentally observed width of the current plateaus between the steps ranges from 0.05 to 0.09 V. The plateau width is determined by the total energy required to add an electron to the tube. This addition energy consists of a combination of finite-size level splitting and the Coulomb charging energy that is due to the small capacitance of the tube. A simple estimate for the energy level splitting for a tube of length L = 30 nm is given by $\Delta E = hv_{\rm F}/2L = 0.06$ eV, where $v_{\rm F} = 8.1 \times 10^5$ m/s is the Fermi velocity and h is Planck's constant. The capacitance C of a nanotube lying on a metallic substrate can be approximated by the formula for a metallic wire parallel to a conducting plane, $C = 2\pi\varepsilon_0 L/\ln\{[d + (d^2 - R^2)^{1/2}]/R\}$ (13), where $\varepsilon_0 = 8.85 \times 10^{-12}$ F/m, d is the distance from the wire axis to the plane, and R = 0.65 nm is the wire radius. Estimating $d \approx 0.9$ nm gives $C \approx 2.0$ aF, which yields a charging energy $E_{\rm c} = e^2/C = 0.08$ eV. Both numbers are in the same range as the observed plateau width. Because the charging



Fig. 1. STM topographic images of individual single-wall carbon nanotubes. (**A**) Example of a nanotube shortened by applying a voltage pulse to the STM tip above the tube (11). (**B**) Atomically resolved image of an armchair nanotube. The arrow denotes the direction of the tube axis. This nanotube can be identified as armchair-type because the hexagon rows run parallel to the direction of the tube axis (see overlay of the graphene lattice). The tube diameter is 1.3 nm. This image was taken before the tube was shortened to 30 nm. Feedback parameters are V = 0.1 V, I = 20 pA. Images were taken in constant-current mode.

energy and level splitting are of about equal magnitude, an irregular step spacing in the I-V curve is expected (14). Here, the relevant point is that each step corresponds to a discrete energy level entering the bias window.

The central result of our experiments is that the tunneling conductance measured for such discrete states is found to oscillate along the length direction of the nanotube with a period different from the atomic lattice constant. I-V spectroscopy curves were obtained at different locations on top of the nanotube along a line parallel to the tube axis in constant-current mode. At every point, spaced 23 pm apart, the feedback was switched off to take an I-V curve, starting at the bias voltage used for feedback in the constant-current mode. Figure 2A shows several I-V curves obtained in this way at different positions. The current displays a clear variation between maximum (dashed curves) and minimum values (solid curves) for negative bias voltage. Peaks in the differential conductance dI/dV (Fig. 2B) appear at the voltage positions of current steps in the I-V curves. The



Fig. 2. STM spectroscopy measurements on a 30-nm-long armchair tube. (A) I-V characteristics of the tube shown in Fig. 1B, taken at positions about 0.18 nm apart [data points 1 to 4 in (C)] on a straight line along the tube axis. Current steps correspond to discrete energy states entering the bias window. Dashed curves, maximum values; solid curves, minimum values. (B) Differential conductance dl/dV versus V, as calculated from the I-V curves. Peaks appear at the voltage positions of current steps in the I-V curves. (C) Differential conductance dl/dV as a function of position along the tube. Data were taken at a bias voltage of -0.08 V. Data points 1 to 4 indicate the positions at which the four I-V curves plotted in (A) and dI/dV curves in (B) were obtained.

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height of the dI/dV peaks varies periodically with position x along the tube axis (Fig. 2C). The period of these oscillations in the differential conductance is ~0.4 nm, which clearly differs from the lattice constant of 0.25 nm. The periodic variation of dI/dV versus x can be attributed to the electronic wave functions in the nanotube, as discussed below.

The wave functions of several adjacent energy levels can be displayed simultaneously by plotting the differential conductance dI/dV as a function of the voltage and the position x along the tube (Fig. 3A). Wave patterns can be observed for four different energy levels appearing at bias voltages of 0.11, 0.04, 0.00, and -0.05 V (15). At each level, a horizontal row of about seven maxima is resolved in dI/dV as a function of position x along the tube (see Fig. 3B for the 1D spatial profile of the wave functions belonging to these states). The experimental quantity dI/dV is a measure of the squared amplitude of the quantized electron wave function $|\Psi(E,x)|^2$ (16). The curves in Fig. 3B are fitted with a function of the form $dI/dV = A \sin^2(2\pi x/\lambda + \phi) + B$, which represents a simple trial function for

Fig. 3. Spectroscopy and topography line scans along the nanotube, showing electron wave functions of discrete electron states as well as the atomic lattice. (A) Differential conductance dl/dV (in color scale) against the bias voltage (y axis) and the position on a straight line along the tube (x axis). This plot results from about 100 I-V curves taken at positions \sim 23 pm apart along the tube axis. Electron wave functions of four different energy levels are observed as periodic variations in dl/dV along the tube at voltages of 0.11, 0.04, 0.00, and -0.05 V. A horizontal row of about seven dl/ dV maxima is observed at each energy level. Note that the exact voltages at which the peaks in dl/dV appear in Figs. 2 and 3A are different (15). (B) dI/dV profiles at the four resolved energy levels. Fits of the function dl/

 $|\Psi(E,x)|^2$. The separation of ~0.4 nm between peaks in dI/dV corresponds to half the wavelength λ because dI/dV measures the square of the wave function. The wavelengths obtained from the fitting procedure vary from 0.66 to 0.76 nm (Fig. 3B). Other measurements on the same tube reproduced values for λ in the range of 0.65 to 0.8 nm. From repeated spectroscopy measurements (such as Fig. 3A) on the same tube, we estimate the error in the wavelength to be ~0.02 nm.

Note that the dI/dV maxima in Fig. 3A occur at different positions x for the different horizontal rows. This excludes many experimental artifacts (such as, for example, oscillations in the STM) and provides compelling evidence for the interpretation in terms of standing electron waves. Typically, only a small number of discrete levels (about four) were observed around zero bias. At larger bias voltages beyond the images shown here, peaks in dI/dV could no longer be discerned clearly. At these voltages, the broadening of energy states apparently exceeds their separation. Similar electron waves with a wavelength of ~0.7 nm were also observed in a



number of other shortened metallic nanotubes. On shortened semiconducting nanotubes, the level splitting could not be resolved, and attempts to measure electron waves were unsuccessful. A small energy level splitting is indeed expected for semiconducting tubes, because here the Fermi energy is located at the top of a band (7).

Figure 3C shows the topographic height profile from the constant-current measurement at +0.3 V, which clearly has a different periodicity from that observed in dl/dV (Fig. 3B). The period of 0.25 nm is in agreement with the atomic lattice constant $a_0 = 0.246$ nm for an armchair nanotube. Apparently we are imaging the atomic corrugation at high bias voltage. Simultaneously, I-V spectroscopy curves are measured at every point (Fig. 3, B and C), starting at the set point used for feedback (100 pA and +0.3 V). As a result of maintaining feedback at this voltage, the lattice periodicity is largely compensated because the STM tip follows the atomic corrugation, which makes it possible to resolve the quantized electron waves in Fig. 3, B and C (16).

Discrete levels are probed at energies near the Fermi energy $E_{\rm F}$, and therefore the wavelength of the electron waves is close to the Fermi wavelength λ_F . Electronic band structure calculations (1, 7) for armchair tubes yield two bands near $E_{\rm F}$ with a linear energy dispersion $E(k) = E_F \pm \hbar v_F(k - k_F)$, where \hbar = $h/2\pi$, $k = 2\pi/\lambda$ is the wave vector, and $k_{\rm F}$ = $2\pi/\lambda_{\rm F}$ is the Fermi wave vector. In undoped nanotubes, the two bands cross at the Fermi energy where $k = k_{\rm F} = 2\pi/3a_0$. This yields $\lambda_F = 3a_0 = 0.74$ nm, independent of the length of the tube. For nanotubes on Au(111), however, $E_{\rm F}$ is shifted away from the crossing point to lower energy by $\delta E =$ 0.3 eV. This is due to charge transfer as a



Fig. 4. Pairing of conductance peaks. (A) Spectroscopy line scan where pairing of dl/dV maxima can be observed. Neighboring peaks are nonequidistant, indicating a nonsinusoidal wave function. The distance between nextnearest-neighbor peaks is ~0.75 nm, which agrees with the Fermi wavelength. (B) Schematic of a possible arrangement of lobes of the wave function of a single molecular orbital. In a line scan along the blue line, peaks in dl/dV will be equidistant, whereas pairing will occur if a line scan is carried out along the red line.

 $dV = A \sin^2(2\pi x/\lambda + \phi) + B$ are plotted as dashed curves. The resulting wavelengths λ are given above the curves on the right. Curves are vertically offset for clarity. (C) Topographic height profile z(x) of the nanotube. STM topographic imaging and spectroscopy were performed simultaneously by scanning the tip along the tube and recording both the tip height (with feedback on) and *I*-V spectroscopy curves (feedback off). Feedback parameters are V = 0.3 V, I = 100 pA.

result of a difference in work function with the underlying substrate (5). This shifts $k_{\rm F}$ to $k_{\rm F} \pm \delta k$ with $\delta k = \delta E/\hbar v_{\rm F}$, and $\lambda_{\rm F}$ thus becomes $(2\pi)/(k_{\rm F} \pm \delta k) = 0.69$ nm (+) or 0.79 nm (–). The experimentally observed wavelengths (Fig. 3B) correspond well to the theoretical values, confirming the predicted band structure with two linear bands crossing near $E_{\rm F}$. This result provides quantitative evidence for our interpretation of the oscillations in dI/dV in terms of wave functions of discrete electron states.

A short metallic nanotube resembles the textbook model for a particle in a 1D box. For a discrete energy state with quantum number n, the corresponding wavelength $\lambda_n = 2L/n$. The observed, wavelength is much smaller than the tube length, in accordance with the fact that the number of electrons within one nanotube band is large $(n \sim 10^2)$. The wavelength will therefore vary only slightly $(\Delta \lambda_n = \lambda_n/n \sim 0.01 \text{ nm})$ for adjacent discrete energy levels in one band.

The measurements reported here are technically challenging because they require a large series of reproducible I-V curves. Occasionally, we were able to resolve some of the spatial structure in the wave function at a length scale smaller than the Fermi wavelength (Fig. 4A). In this scan the peak spacing is nonequidistant, leading to an apparent pairing of peaks. This feature indicates that the wave function does not conform to a simple sinusoidal form. Recent calculations by Rubio et al. (17) indicate a nontrivial spatial variation of the nodes in the wave function of discrete electron states in the direction perpendicular to the tube axis (Fig. 4B). Line profiles can show either pairing or an equidistant peak spacing, depending on the exact position of the line scan. The observation of pairing confirms that the relevant period in the line scans is the distance between next-nearestneighbor peaks.

Our experiments demonstrate that individual wave functions corresponding to the quantized energy levels in a short metallic nanotube can be resolved because of the large energy level splitting. The technique for recording the wave periodicity at different energy states provides a tool for further exploration of the dispersion relation in nanotubes. Future work should include similar experiments on nanotubes with various chiral angles. The methodology presented here also opens up the possibility of obtaining full 2D spatial maps of the electron wave functions in carbon nanotubes.

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- 16. The total wave function is in fact defined by the atomic lattice potential modulated with a standing wave profile resulting from the confinement in the length direction. Because the STM tip follows the atomic corrugation by scanning in constant-current mode at a high bias voltage, the lattice periodicity is largely compensated so that the standing waves can be resolved in the spectroscopy measurements for several discrete states at low bias.
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DOAS Measurements of Tropospheric Bromine Oxide in Mid-Latitudes

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Episodes of elevated bromine oxide (BrO) concentration are known to occur at high latitudes in the Arctic boundary layer and to lead to catalytic destruction of ozone at those latitudes; these events have not been observed at lower latitudes. With the use of differential optical absorption spectroscopy (DOAS), locally high BrO concentrations were observed at mid-latitudes at the Dead Sea, Israel, during spring 1997. Mixing ratios peaked daily at around 80 parts per trillion around noon and were correlated with low boundary-layer ozone mixing ratios.

Sudden boundary-layer ozone (O_3) depletion events during Arctic spring (1, 2) are likely caused by the presence of reactive halogen

*To whom correspondence should be addressed. Email: stu@uphys1.uphys.uni-heidelberg.de species, particularly BrO. Bromine or BrO (or other halogens) catalytically convert two ozone molecules to three oxygen molecules without a loss of the halogen (3). Although the chemical cycles destroying ozone are well understood, the source of the reactive bromine compounds remains unclear, but they are most likely formed by oxidation of sea salt halogenides (4). Bromine oxide has been observed by DOAS in several places in the Arctic (5, 6) and Antarctica (7). Satellite-

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