

Publisher: Richard S. Nicholson Editor-in-Chief: Daniel F. Koshland, Jr.

Editor: Ellis Rubinstein

Managing Editor: Monica M. Bradford

Deputy Editors: Philip H. Abelson (Engineering and Applied Sciences); John I. Brauman (Physical Sciences); Thomas R. Cech (Biological Sciences)

Editorial Staff

Assistant Managing Editor: Dawn Bennett

Senior Editors: Eleanore Butz, R. Brooks Hanson, Barbara Jasny, Katrina L. Kelner, David Lindley, Linda J. Miller, Phillip D. Szuromi, David F. Voss

Associate Editors: Gilbert J. Chin, Pamela J. Hines, Paula A Kiberstis Suki Parks I Brvan Bay

Letters: Christine Gilbert, Editor; Steven S. Lapham Book Reviews: Katherine Livingston, Editor; Annette Theuring, Assistant Editor; Susan Randolph, Editorial Assistan

Contributing Editor: Lawrence I. Grossman

Editing: Lois Schmitt (training), Valerie Jablow, Senior Copy Editor; Douglas B. Casey, Harry Jach, Erik G. Morris, Christine M Pearce

Copy Desk: Ellen E. Murphy, *Supervisor;* Joi S. Granger, Beverly Shields, Melissa M. Quackenbos, Kirsten L. Wall; Kameaka Williams, *Assistant*

Editorial Support: Sherryf Farmer, Supervisor; Linda Dienavs, Carolyn Kyle, Michele Listisard, Diane Long, Jen-nifer Mattson, Patricia M. Moore

Administrative Support: Leslie Blizard, Sylvia Kihara, Jeanette Prastein

Telephone: 202-326-6501; FAX: 202-289-7562; TDD: 202-408-7770

News Staff

News Editor: Colin Norman

Features Editor: John M. Benditt Deputy News Editors: Tim Appenzeller, Joshua Fischman, Jean Marx Jeffrey Mervis

News & Comment/Research News Writers: Ivan Amato, Christopher Anderson, Jon Cohen, Faye Flam, Troy Gately, copy, Constance Holden, Richard A. Kerr, Eliot Marshall, Richard Stone, Karen Fox, Intern

U.S. Bureaus: Marcia Barinaga (Berkeley), Elizabeth Culotta (Durham, NC), Anne Simon Moffat (Chicago), John Travis

Contributing Correspondents: Joseph Alper, Barry A. Cipra, Robert Crease, Ann Gibbons, Virginia Morell, Robert Pool, Leslie Roberts, Gary Taubes, M. Mitchell Waldrop Administrative Support: Fannie Groom, Jennifer Hodgin Telephone: 202-326-6500; FAX: 202-371-9227

Art & Production Staff

Production: James Landry, Director; Wendy K. Shank, Manager; Laura A. Creveling, Scherraine B. Mack, Linda C. Owens, Associates

Art: Amy Decker Henry, *Director*; C. Faber Smith, *Asso-ciate Director*; Diana DeFrancesco, *Technical Illustrator*; Holly Bishop, *Graphics Associate*; Elizabeth Carroll, *Graph*ics Assistant

Europe Office

Senior Editor: Richard B. Gallagher Associate Editor: Jeffrey Williams News Editor: Daniel Clery Correspondent: Peter Aldhous Editorial Associate: Catherine S. Siskos Business Manager: Julie Eastland Address: Thomas House, George IV Street, Cambridge, UK CB2 1HH Telephone: (44) 0223 302067; FAX: (44) 0223 302068

Science Editorial Board John J. Hopfield

Yasutomi Nishizuka

Bengt Samuelsson

Charles J. Arntzen	John J. Hopfield
Elizabeth E. Bailey	F. Clark Howell
David Baltimore	Paul A. Marks
J. Michael Bishop	Yasutomi Nishizuk
William F. Brinkman	Helen M. Ranney
E. Margaret Burbidge	Bengt Samuelssor
Pierre-Gilles de Gennes	Robert M. Solow
Joseph L. Goldstein	Edward C. Stone
Mary L. Good	James D. Watson
Harry B. Gray	Richard N. Zare

EDITORIAL

Bioinorganic Chemistry

Some scientific activity is long-lived, but much of it changes with time. Older areas change and newer areas develop. One of the exciting changes in chemistry has been the renaissance of inorganic chemistry, especially in the area of bioinorganic chemistry. New molecules, new concepts, an understanding of important biology, and the application of new principles to nonbiological problems are among the hallmarks of this field. The remarkable growth of bioinorganic chemistry and its insights that have led to new research in chemistry present an important lesson in the development of science. In this issue, we present a Perspective by Lippard that provides a thoughtful overview of this field and four general articles that delineate some recent developments.

Metals play many important roles in biological systems. Aside from their properties as independent ions and charge carriers, they can act as structural components-holding complex structures together with very specific geometries—and as catalytic centers. Metals have a profound effect in accomplishing many chemical transformations. Almost everyone recognizes the role of iron in the oxygen carrier, hemoglobin, and the role of cobalt in vitamin B_{12} is well known, but the tale is infinitely richer even than these remarkable examples would suggest

Karlin discusses metalloenzymes—metal-containing proteins that act as catalysts. Basic recurring structures have been found that have been fine-tuned in different proteins in order to carry out specific functions. Understanding these important compounds involves structure elucidation, spectroscopy, mechanistic studies, and biomimetic modeling. Each of these aspects of chemistry are themselves important areas of activity. When all of the pieces are put together, an extremely rich picture results.

In one of the most startling recent discoveries in chemistry and biology, we have learned that RNA can itself be a catalyst. Pyle describes the surprising and extraordinarily interesting phenomena of RNA as a metalloenzyme. Ribozymes require and depend on divalent metal cations for their activity. The metals are crucial for structure by holding the catalyst in the optimum geometry for activity. They also participate in the catalytic reactions of RNA phosphodiester linkages.

Metals play a special role in gene expression. O'Halloran discusses metalloregulatory proteins-how they function and why. Zinc finger proteins are now recognized as among the most pervasive and important structural features in biology. Some pertinent examples of regulation, including the control mechanism for ferritin production and its relation to aconitase activity, are analyzed.

Finally, Abrams and Murrer describe the use of metals in diagnostic reagents and in drugs. Metals can have spectroscopic emission or absorption in regions which are otherwise transparent, so metal-containing diagnostic reagents would appear to be ideal, provided their toxicity can be controlled. Specific biological molecules and organs often show great specificity and binding affinity for certain metals. Consequently, the idea of incorporating metals into diagnostic reagents and into drugs is an important and active area of research. Currently, many imaging methods, such as radiodiagnosis or paramagnetic-enhanced proton relaxation, depend specifically on metals. Similarly, the combination of affinity and chemical reactivity makes cisplatin and related compounds particularly powerful drugs.

The ultimate application of many of the compounds and the principles now being uncovered is sure to be significant. Oxygen carriers modeled on hemoglobin, oxidizing reagents modeled on cytochrome P-450s, reducing reagents modeled on nitrogenase, new diagnostic reagents, and new drugs will benefit all of us in the long run. The perceptive reader will recognize the value of research in new areas such as bioinorganic chemistry and will recognize that directing this research too closely can only be detrimental. Nature still has many secrets for us to discover and unlock, but we have not yet learned enough to know exactly where and how to look.

John I. Brauman