

evoked consistent ocular following and, as Fig. 2A shows for one animal, once again the response varied inversely with the viewing distance. The maximum compensatory eye speeds achieved within 100 ms of the onset of stimulus motion are plotted in Fig. 2B, from which it is evident that OFR responses were a linear function of the inverse of the viewing distance. Similar data were obtained from all four animals. When the data from each animal were expressed as a percentage of that same animal's highest mean response and plotted against the inverse of the viewing distance, there was surprisingly little variation: the range of slopes was only 14% to 16% per diopter (mean, 15% per diopter), and the range of intercepts was 22% to 34% (mean, 28%).

That the OFR shares the TVOR's dependence on proximity leads us to suggest that the two reflexes share a pathway whose efficacy is modulated by absolute distance cues (9). Further, we suggest that these two systems are synergistic, functioning to compensate selectively for translational disturbances of the observer (10). In our proposed scheme (see Fig. 3), the TVOR and OFR share two gain elements: a variable one (k_1/d , where k_1 is a constant and d is the target distance), which gives the dependence on proximity, and a fixed one (k_2), which accounts for the offset in our data. The variable gain element allows the TVOR to receive inputs encoded in Cartesian coordinates [translational velocity of the head (\dot{H}_T)] and to respond with outputs coded in polar coordinates [rotational velocity of the eyes (\dot{E}_R)]. That the visual contribution enters the system upstream of the variable gain element might seem less than optimal since negative feedback systems such as this function best when their gain is fixed at some maximum limited only by stability considerations. However, we suggest that the variable element helps to offset velocity saturation, which is known to be present in the OFR (2) and has been incorporated into Fig. 3. Retinal slip speeds experienced by the moving observer will tend to vary inversely with viewing distance; hence ocular following will tend to show increasing saturation with near viewing, an effect that the gain element, k_1/d , will counteract. Thus, the observed dependence on proximity meets the geometric needs of the TVOR and offsets the intrinsic limitations of the OFR.

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1. The compensatory eye rotation, θ , required to keep the eyes aligned on an object at distance, d , when the observer is moved sideways in a straight line ("lateral translation" along the interaural axis) over a distance, M , is $\arctan(M/d)$. Taking derivatives, the velocity of the compensatory eye rotation, $\dot{\theta}$, is given

by the expression:

$$\left[\frac{1}{1 + \left(\frac{M}{d}\right)^2} \right] \left(\frac{\dot{M}}{d} \right)$$

In our experiments and most comparable everyday situations, the time scale of interest is brief and $d \gg M$, so that the first term approaches 1, hence the required compensatory eye movement effectively becomes \dot{M}/d .

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um pentobarbital to implant devices for the recording of eye movements; sterile procedures were used (2). Animals were trained to fixate target lights by previously described methods (2).

8. Our previous studies had shown that ocular following responses are transiently augmented by a prior saccade across a textured scene (2), and we made use of this in the present study by having the animal make a 10-degree rightward saccade into the center of the pattern 30 ms before the stimulus motion began. This was achieved by reinforcing the animals for transferring fixation between appropriately positioned target spots projected onto the scene.
9. Additional experiments indicated that the linear dependence of both reflexes on the inverse of the viewing distance was preserved over a range of stimulus motion parameters: sled jerk amplitude was varied from 305 to 940 cm/s² per second, and the velocity of the visual scene was varied from 10 to 160 degrees/s. Further data from three of the monkeys indicated that the TVOR responses could be increased by selectively increasing either vergence (by means of base-out prisms with the most distant target) or accommodation (by means of base-in prisms with the nearest target) and these increases in response were similar in the two cases. These findings indicate that the TVOR uses some internal measure of both the vergence and the accommodative states to modulate its gain in accordance with the viewing distance.
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Sequence-Specific Isotope Effects on the Cleavage of DNA by Bleomycin

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Bleomycin is a metal- and oxygen-dependent DNA cleaver. The chemistry of DNA damage has been proposed to involve rate-limiting abstraction of the 4'-hydrogen. A DNA fragment has been prepared that contains [4'-²H]thymidine residues of high isotopic content. Primary kinetic isotope effects have been directly observed at individual thymidine residues with DNA sequencing technology.

THE ELUCIDATION OF THE MECHANISMS of DNA cleavage by bleomycin (BLM) (1), the neocarzinostatin cofactor (2), calicheamicin (3), esperamicin (4), and related compounds (5) has been extensively investigated. High sensitivity and precision are required to evaluate the mechanistic changes that may accompany alterations in local DNA conformation or modifications in drug structure or both. We report a new technique that makes use of specifically deuterated ³²P end-labeled DNAs in combination with gel electrophoresis to detect and quantitate potential-

ly rate-limiting carbon-hydrogen bond cleavages by DNA-cleaving drugs at individual sequence sites. We use BLM as an example.

The activity of BLM in vitro depends on Fe(II) and O₂ or Fe(III) and H₂O₂ (6). The initial BLM-Fe(II)-O₂ complex (Fig. 1) undergoes one-electron reduction to ultimately yield "activated BLM," which can initiate DNA damage (7). Two types of DNA damage are observed with "activated BLM" (Fig. 2, A and B). Pathway A results in the formation of nucleic acid base propenal and a DNA strand scission that yields 3'-phosphoglycolate and 5'-phosphate termini. Pathway B results in the liberation of nucleic acid base plus an alkali-labile site that cleaves at pH 12 with piperidine to afford a 3'-phosphate and a 5'-phosphate terminus. On the basis of the identification of the propenal (7, 8), Giloni *et al.* (8) inferred that

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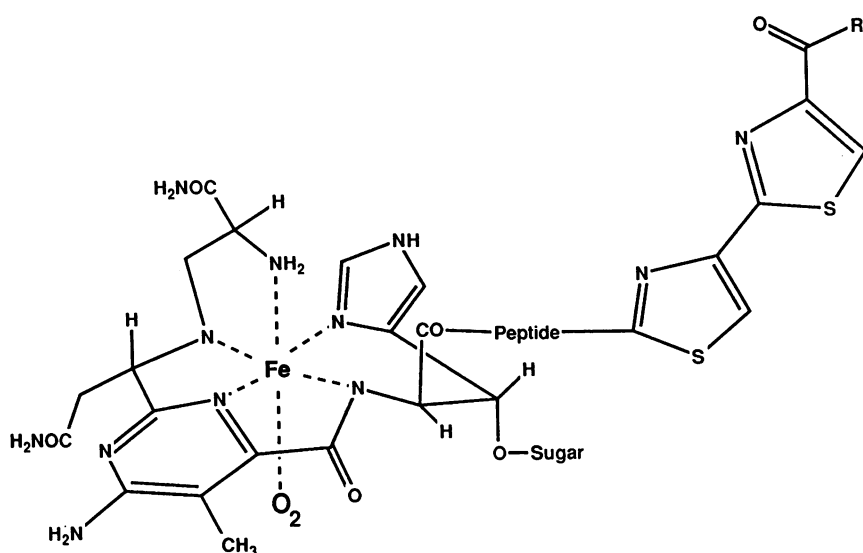
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We proposed a unified mechanism for BLM action (Fig. 2) (10–14) based on our studies of BLM reaction with simple DNA polymers [such as poly(dA·dU)] tritiated at specific positions in the deoxyribose ring, which suggested that “activated BLM” effected a 4'-C-H bond cleavage in B-form DNA that was subject to a surprisingly large tritium selection effect ($k_H/k_T = 7$ to 11). The putative 4'-radical intermediate could be intercepted by O_2 (pathway A) or under-

Our proposal has not gone without criticism (15). Several factors limit the sensitivity of the approach. First, tritium is necessarily used as a tracer isotope, so that extensive DNA damage is required (upwards of 50%) for the accurate quantitation of tritiated products. Second, such large isotope effects can be subject to relatively large errors (11) and are global in nature without specific sequence information. Finally, the analysis of the chemistry at minor damage sites is hampered by the lack of sensitivity. We noted, however, that the tritium selection

$$k_H/k_D = 1.44 \sqrt{k_H/k_T} = 4 \text{ to } 5 \quad (1)$$

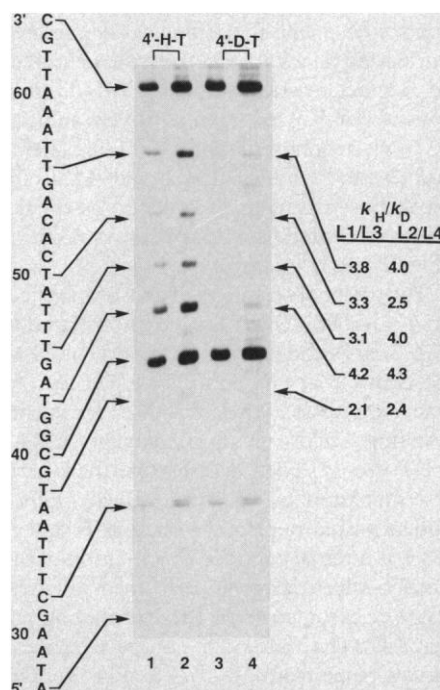
The chemical synthesis of [$4',2\text{-}^2\text{H}$]dTTP ($>95\%$ ^2H) followed published procedures (17). The deuteriated nucleotide was incorporated into the (+)-strand of the Eco RI-Bam HI restriction fragment (375 bp) of



The diagram illustrates the chemical reaction scheme for the formation of a nucleic acid base from a nucleoside monophosphate derivative. The scheme shows the following steps:

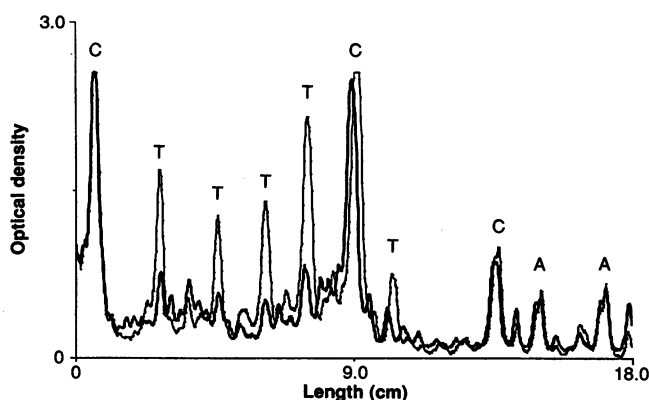
- Starting Material:** A nucleoside monophosphate derivative (NMP) is shown on the left. It consists of a ribose sugar ring with a phosphate group (ROPO₃⁻) attached to the 5' carbon and a nucleobase (N) attached to the 1' carbon. The phosphate group is shown as O=P(OR')₂.
- Reaction Conditions:** The reaction proceeds under the following conditions:
 - 4'-H RDS (Rate-Determining Step)
 - $k_H/k_T = 7 - 11$
 - B-form DNA
- Intermediate A:** The reaction proceeds via a 4'-H RDS to form an intermediate A. Intermediate A is a nucleoside monophosphate derivative where the phosphate group is now a cyclic diphosphate (pyrophosphate) structure, O=P(OR')₂, and the sugar ring is in a B-form DNA conformation.
- Intermediate B:** Intermediate A can follow two pathways:
 - Pathway A:** Intermediate A can undergo a reaction with O₂ to form a nucleic acid base (N) and a phosphate group (O=P(OR')₂).
 - Pathway B:** Intermediate A can undergo a reaction with OH⁻ (Piperidine) to form a nucleic acid base (N) and a phosphate group (O=P(OR')₂).
- Final Products:** The final products are a nucleic acid base (N) and a phosphate group (O=P(OR')₂).

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Fig. 4. Densitometry scan of the autoradiogram in Fig. 3. Total strand damage of [4'-³H]T-containing DNA fragment (lane 2, light line) by "activated BLM" is compared to total strand damage of [4'-²H]T-containing DNA fragment (lane 4, heavy line). The ratio of the integrated peak areas is a direct measure of the kinetic isotope effect at that position (L2/L4, Fig. 3). Similar scans were performed on lanes 1 and 3 (L1/L3, Fig. 3). The scans shown are the raw data without normalization.



pBR322 and then the 5' end of this strand was labeled with ³²P (18–21). The autoradiogram of the cleavage pattern (22) resulting from treatment of this fragment with limiting "activated BLM" is shown in Fig. 3. Identical experiments were performed on a control fragment containing no deuterated nucleotide (lanes 1 and 2) and the deuterated fragment (lanes 3 and 4). In addition, control and labeled fragments were subjected to electrophoresis both without (lanes 1 and 3) and with (lanes 2 and 4) alkali-piperidine treatment in order to assess the effects of deuteration on pathways A and B (Fig. 2) (23).

The direct observation of an isotope effect on 4'-C-H bond cleavage is demonstrated in Fig. 3, as is the known preference of BLM for cleavage at GC and GT sequences (1). The suppression of [³²P]DNA fragments resulting exclusively from damage at [4'-²H]T sites is strong evidence for the kinetic discrimination by "activated BLM." Since undeuterated nucleotides such as C and A serve as internal controls, quantitation of the isotope effects may be performed by scanning densitometry of the autoradiogram (Fig. 4). The calculated isotope effects exhibit a range from ~2 to 4.5 (Fig. 3). The differences in the magnitude of the effect at different sites is reproducible and suggests that local sequence variability may be important. Dissociation of the "activated BLM" from DNA must also be faster than bond cleavage to permit discrimination between labeled and unlabeled cleavage sites.

The isotope effects on pathways A and B (Fig. 2) are essentially the same for a particular damage site. The isotope effects on pathway A (L1/L3; Fig. 3) were determined by quantitation of neutral strand scission, whereas those on pathways A plus B (L2/L4; Fig. 3) were determined by quantitation of total alkali-induced scission. The effect on pathway B is similar to that on pathway A, which affirms the partitioning of a common intermediate at individual damage sites.

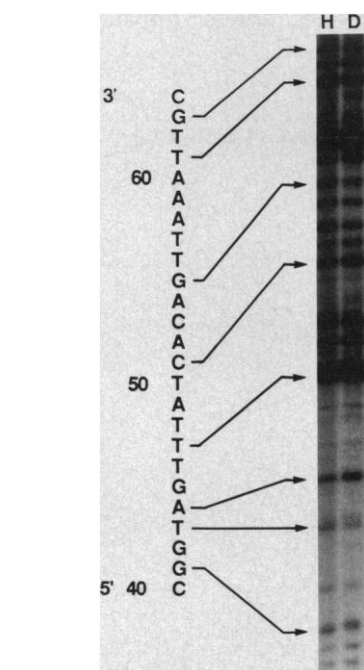


Fig. 5. Autoradiogram of a high-resolution denaturing polyacrylamide gel of the reaction of esperamicin A₁ with the pBR322 fragment containing [4'-³H]T (lane H) or [4'-²H]T (lane D). Each reaction (25 μl) contained 50 mM tris-HCl, pH 7.5, 0.1 mM EDTA, sonicated salmon sperm DNA (0.2 μg/μl), and ~164,000 cpm of a fragment 5' end-labeled with ³²P. Esperamicin A₁ (50 μM; 1 μl) was added and the solution was incubated at room temperature for 20 min. Dithiothreitol (25 mM; 1 μl) was then added and the reaction was incubated at 37° for 10 min. The reaction was terminated by addition of 0.1 mM EDTA, 2.5M sodium acetate and salmon sperm DNA (0.2 μg/μl) (50 μl final volume). Samples were precipitated with ethanol and subjected to gel electrophoresis (23).

Moreover, preliminary experiments varying the O₂ concentration to alter the partition ratio corroborate this proposal.

The isotope effects are clearly dependent on the nature of the DNA cleaver. Experiments performed with esperamicin A₁ (4) revealed, in addition to a substantially different sequence specificity, no significant iso-

tope effect on 4'-C-H bond cleavage (Fig. 5). While this result does not exclude this cleavage as a mode of action of esperamicin, it does rule out this step as a rate-determinant. The results, nevertheless, constitute a convincing control.

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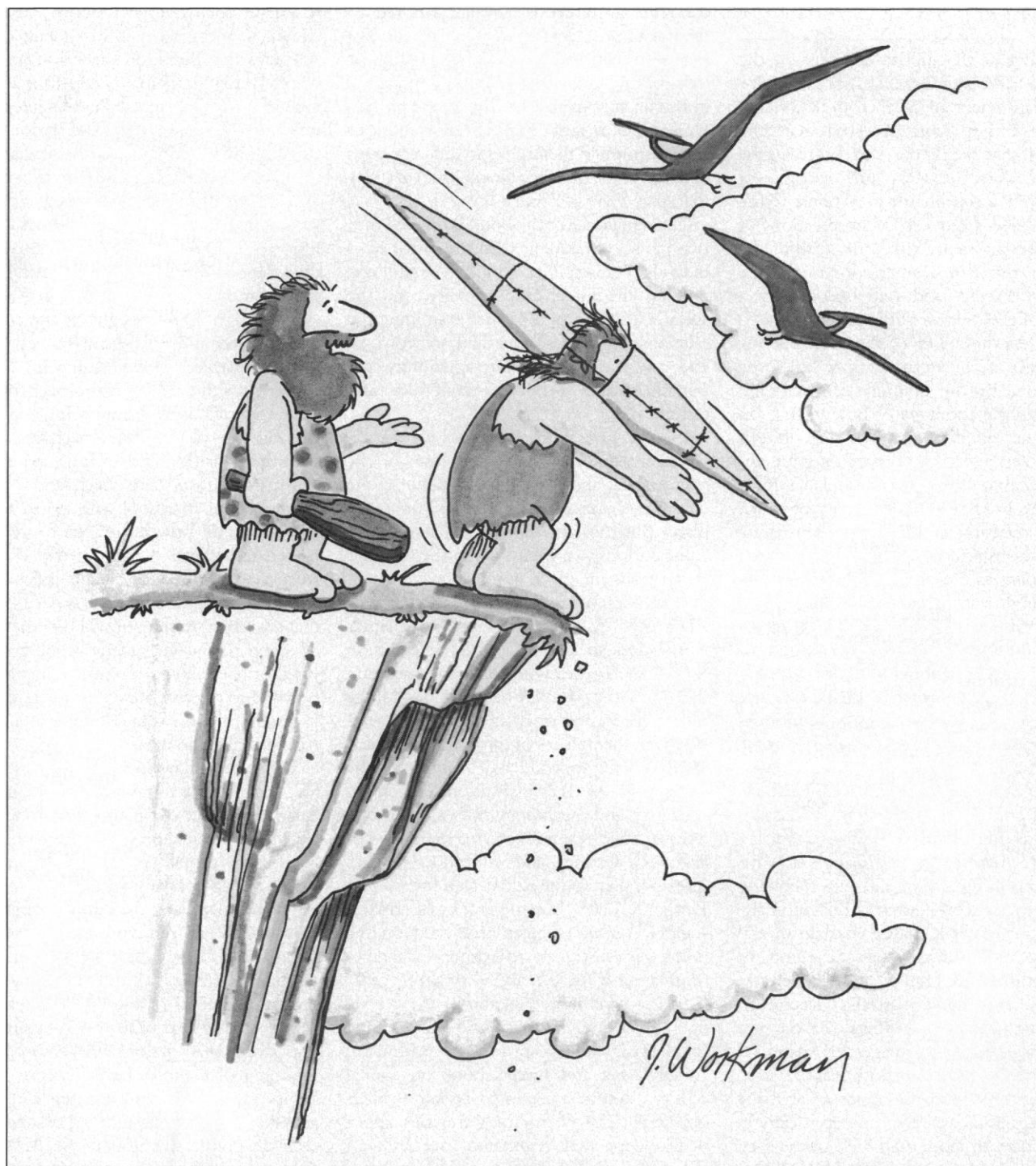
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18. The Eco RI–Bam HI restriction fragment (375 bp) from pBR322 was cloned into M13mp19, and the single-stranded DNA was isolated from infected *Escherichia coli* JM101 (19, 20). Synthesis of the complementary strand was accomplished by annealing kinase-treated primer to the single-stranded template and by performing the polymerization with Sequenase (U.S. Biochemicals) and with dGTP, dATP, dCTP, and either dTTP or [4'-²H]dTTP. The polymerization and terminal ligation was monitored by comparison to replicative form on agarose gels. Samples were purified on Sephadex G-50 (21). The prepared replicative form containing either [4'-³H]- or [4'-²H]T was treated with Eco RI (1 hour, 37°) and 5' end-labeled by standard procedures (21). The desired 375-bp fragment was produced by Bam HI digestion, resolved by electrophoresis on a 5% acrylamide gel and purified on an NACS Prepac (BRL) column.
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23. Since pathway A leads to 3'-phosphoglycolate termini and pathway B to 3'-phosphate termini, the corresponding oligomeric fragments are separable on high-resolution gels under certain conditions, which is especially apparent in Fig. 3 for the smallest fragments (see A and C damage sites at bottom of autoradiogram). Thus lanes 1 and 2 yield a predominant band corresponding to the 3'-phosphoglycolate ends generated under neutral conditions, and

lanes 2 and 4 reveal an additional band of slower mobility corresponding to the 3'-phosphate ends produced by the alkali-piperidine treatment. The resolution is lost as fragment size increases, but this has no effect on the subsequent interpretation of the

experiment.
24. We thank G. Freeman and T. D. Tullius for helpful advice and for use of the scanning densitometer. A gift of esperamicin A₁ was provided by B. Krishnan and T. W. Doyle. Supported by an American Cancer

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"Are you sure about this, Stan? It seems odd that a pointy head and long beak is what makes them fly."