to the uptake of soluble protein antigens, antibodies present in the mucus layer may combine with antigens to form complexes and thereby prevent the molecules from reaching the surface of the enterocyte from whence uptake by pinocytosis is initiated (2). With respect to the uptake of preformed antigen-antibody complexes or those forming in the lumen of the intestine, mucus may serve as a mechanical barrier to the diffusion of the complexes, thereby again limiting the access of complexes to the surface of the enterocyte (1). Our experiments suggest an additional relation between immune complexes and mucus, that is, the ability of the former to stimulate release of mucus by the intact small intestine of the rat. Release of mucus may in turn serve to clear the surface of the gut of adherent immune complexes.

The mechanisms by which immune complexes stimulate the release of goblet cell mucus is still to be determined. Certain antigen-antibody complexes might interact with receptors on the surface of goblet cells and thereby stimulate release of mucus, or interaction of complexes with epithelial cells might stimulate such cells to release unknown mediators which in turn activate goblet cells. Alternatively, it is known that the first component of complement and possibly others are produced by the intestine (16), and that rat IgG-1 antibodies are capable of activating the complement sequence (17). Therefore, it is possible that components of the complement system might be involved in mediating the release of mucus.

W. Allan Walker

MARGARET WU

Pediatric Gastrointestinal Unit, Massachusetts General Hospital, Boston 02114

KURT J. BLOCH Arthritis, Clinical Immunology and Allergy Units, Massachusetts General Hospital

References and Notes

- 1. W. A. Walker, M. Wu, S. Abel, K. J. Bloch, J.
- W. A. Walker, M. Wu, S. Aber, K. J. Bloch, J. Immunol. 117, 1028 (1976).
 W. A. Walker, M. Wu, K. J. Isselbacher, K. J.
- 3.
- W. A. Walker, M. M. S. (1975).
 W. A. Walker, K. J. Isselbacher, K. J. Bloch, *ibid.*, 111, 221 (1973). **W**. A P. Minden and R. S. Farr, Handbook on Experimental Immunology (Blackwell, Oxford, 1973),
- 5. The term ABC-33 refers to the reciprocal dilu-
- tion of antiserum which will precipitate 33 per-cent of the labeled antigen added to the test sys-
- . Cuatrecasas and C. B. Anfinsen, Annu. Rev. Biochem. 40, 259 (1971).
- (Thomas, Springfield, Ill., 1961), p. 319. A. L. Wu and W. A. Walker, Infect. Immun. 14, 7. 8.
- A. L. Wu and W. A. Walker, Infect. Immun. 14, 1034 (1976). R. D. Lillie, Histopathologic Techniques and Practical Histochemistry (McGraw-Hill, New York, 1965), pp. 38 and 510.
- 10. M. A. Jennings and H. W. Florey, Q. J. 372

Exp. Physiol. 41, 131 (1956); N. L. Lane, L. R. Otero-Vitardebo, G. C. Godman, J. Cell. Biol. 21, 339 (1964).

- D. H. Lowry, N. J. Rosebrough, A. L. Farr, R. J. Randal, J. Biol. Chem. 193, 265 (1951).
 H. L. Elliott, C. C. J. Carpenter, R. B. Sack, J. H. Yardby, Lab. Invest. 22, 112 (1970).
- 13. R. J. Gibbons and J. van Houte, Annu. Rev. Mi-
- K. J. Gibbons and J. van Houte, Annu. Rev. Microbiol. 29, 19 (1975).
 R. J. Gibbons, D. M. Spinell, Z. Skobe, Infect. Immun. 13, 238 (1975).
 D. R. Strombeck and D. Harrold, *ibid.* 10, 1266 (1977).
- 16. H. R. Colten, Adv. Immunol. 22, 67 (1976).
- H. C. Morse III, K. J. Bloch, K. F. Austen, J. Immunol. 101, 658 (1968); *ibid*, 102, 327 (1969).
- Supported by grants from NIH (AM-16269, AM-5067, AM-3564, and AI-10129), a contract from the U.S. Army Medical Research and Devel-opment command (DADA-17-70-C-0113), and 18. grants from the Massachusetts chapter of the Arthritis Foundation and L. H. Bendit Foundation. W.A.W. is a recipient of a Research Aca-demic Career Development Award (KO4-AM00025) from the NIAMDD.

30 November 1976; revised 8 February 1977

Circulation of H⁺ and K⁺ Across the Plasma **Membrane Is Not Obligatory for Bacterial Growth**

Abstract. Streptococcus faecalis grows normally in the presence of gramicidin and other ionophores under conditions such that there is no gradient of pH or of electrical potential across the plasma membrane and that currents of H^+ , K^+ , and Na^+ are short-circuited. Growth requires a rich medium, a slightly alkaline pH, and a high concentration of external K^+ . The proton circulation maintains the cytoplasmic pH and pools of ions and other metabolites but is not obligatory for biosynthetic functions including DNA replication, cell division, or assembly of the structural framework of the cell.

It is now generally accepted that a central feature of energy metabolism in bacteria is the circulation of protons across the plasma membrane (1). Several major metabolic pathways, including the respiratory chain and the membrane-bound adenosine triphosphatase (E.C. 3.6.1.3) complex, mediate electrogenic transport of protons outward; the current loop is completed by an array of molecular devices that allow protons to return to the cytoplasm while performing useful work. Examples of proton-linked processes are oxidative and photosynthetic phosphorylation, transhydrogenation of pyridine nucleotides, motility, and the transport of many metabolites (1, 2). In order to determine whether ion currents are involved in other essential functions, particularly in the construction and replication of the fabric of bacterial cells, we have examined the growth of Streptococcus faecalis in the presence of ionophores. This fermentative organism normally maintains a circulation of protons across the membrane by means of a proton-translocating adenosine triphosphatase (2, 3); gramicidin and other ionophores were used to short-circuit the proton current and to dissipate gradients of K⁺ and Na⁺. The results confirm that the proton circulation is required to maintain cytoplasmic pools of ions and metabolites as well as a neutral pH. However, it is not obligatory for the synthesis of macromolecules and organelles (cell wall, membranes, nucleoid and other structural elements), for DNA replication, or for cell division.

The basic observation is illustrated in Fig. 1A. Growth of Streptococcus fae-

calis (faecium) (American Type Culture Collection 9790) in the standard complex medium NaTY (4) was completely blocked by gramicidin D, an ionophore that renders the cytoplasmic membrane permeable to K^+ , Na⁺, and H⁺ (5, 6). By contrast, in the medium designated KTY2XH (4) exponential growth continued at a rate not much less than that of control cells. This particular medium is buffered at pH 7.7 and contains 0.28N K⁺ and less than 0.01N Na⁺. Similar results were obtained with complex media containing as much as $1N K^+$ and buffered with bicine, tricine, or Hepes (7) between pH 7.5 and 8.2. Other ionophores, including gramicidin A, valinomycin plus nigericin (1 μ g/ml each), and the proton conductors tetrachlorosalicylanilide and carbonylcyanide mchlorophenylhydrazone $(10^{-5}M)$ also block growth in NaTY but not in media enriched with K⁺ (data not shown).

Although the properties of the ionophores are well documented (5, 6), it is important to establish that cells growing in the presence of gramicidin on medium KTY2XH maintain neither an electrical potential across the cytoplasmic membrane nor concentration gradients of H⁺, K^+ , or Na⁺. With methods presently available (8) these measurements cannot be made directly on growing cells. We therefore compared cells that had been grown for at least four generations in KTY2XH in the absence or presence of gramicidin under the conditions of Fig. 1A; the cells were then washed and resuspended in buffer $(0.14M \text{ K}_2\text{HPO}_4)$, 0.05M Hepes, pH 7.5). Cells grown with gramicidin retain the antibiotic through

washing and resuspension. With glucose as an energy source, control cells generated a pH gradient of about 0.6 unit, with the interior alkaline, as measured by the distribution of 14C-dimethyloxazolidinedione (8); in gramicidin-grown cells there was no detectable pH gradient. Attempts to measure the membrane potential on the basis of the accumulation of lipid-soluble cations or by fluorescence quenching (8, 9) were vitiated by the high concentration of external K+; we therefore used the accumulation of ¹⁴C-methyl-β-D-thiogalactoside as an index of the total proton motive force (10). In the presence of glucose, control cells achieved a concentration gradient of about 24 (average of three experiments), corresponding to a total proton motive force of at least -83 mv; there was no measurable accumulation by gramicidingrown cells. In other experiments we found that gramicidin-grown cells exhibit neither glucose-stimulated uptake of threonine (11) nor valinomycin-induced quenching of dye fluorescence (9); these results support our expectation that gramicidin-grown cells are permeable to cations and cannot sustain a metabolic proton circulation. Donnan effects could still give rise to a membrane potential or pH gradients, but this should not be the case here as the external K⁺ concentration (0.28N to 1N) brackets the cytoplasmic concentration (0.4N to 0.8N)(3). We conclude that the cells are indeed permeable to H^+ , K^+ , and Na^+ and that they are growing in the absence of a significant membrane potential or cation concentration gradients.

We now turn to the capacities and limitations of cells short-circuited by gramicidin. Exponential growth continued for at least 15 generations when portions of the culture were periodically diluted with fresh medium containing gramicidin; the doubling time was generally somewhat longer than that of control cells (on the average, 50 and 40 minutes, respectively). Cell division was essentially normal. After 15 doublings in the presence of gramicidin the viable count was 45 percent of that of a control culture at the same turbidity; the organisms tend to form somewhat longer chains in the presence of the antibiotic. Normal and gramicidin-grown cells did not differ markedly with respect to their adenosine triphosphate content, their rates of glycolysis and adenosine triphosphate turnover, or their ability to adapt to arginine fermentation with induction of the arginine dihydrolase (E.C. 3.5.3.6) pathway. Streptococcus faecalis (faecium) 9790 generates adenosine triphosphate entirely by substrate-level reactions 22 JULY 1977



Fig. 1. Effect of gramicidin D on the growth of Streptococcus faecalis. (A) The role of K+ Organisms were grown at 34°C in either the Na⁺-rich medium NaTY (pH 7.5; curves 1 and 2) or in the K+-rich medium KTY2XH (curves 3 and 4) at pH 7.7. The arrow indicates the addition of gramicidin D (1 μ g/ml final, in ethanol) to tubes 2 and 4; control cultures received ethanol only. (B) The role of pH. Cultures were grown in KTY2XH at pH 7.7, 7.1, 6.6, or 6.0. Control cultures received ethanol at the arrow; all grew at the same rate (curve). Experimental cultures received gramicidin D (1 μ g/ml final); growth was increasingly inhibited at the lower pH (curves 2 through 5). Growth was monitored on the basis of the turbidity at 600 nm.

rather than by phosphorylation linked to electron transport. Most important, electron micrographs revealed no obvious morphological abnormalities; thin sections of control and gramicidin-grown cells were indistinguishable at a magnification of \times 100,000 (data not shown).

However, gramicidin restricts growth to a narrow range of conditions. Growth in KTY2XH required a pH above 7 (Fig. 1B), presumably because in the presence of gramicidin the cytoplasmic pH equals that of the medium. A very high concentration of K^+ was also required (at pH7.8, the doubling time was approximately 50 minutes in 0.3N K⁺ and 90 minutes in 0.1N K⁺; the cells did not grow in 0.02N K⁺), and growth was inhibited by Na⁺ above 0.1N. By contrast, control cells grow well on 0.001N K+, even in the presence of 0.3N Na⁺, at a pH as low as 5. The concentrations of tryptone and yeast extract were not critical, but, when the experiments were repeated in defined medium (12), the limitations of short-circuited cells became apparent. In the standard medium (12), containing most of the amino acids at 1 to 2 mM, the cells grew almost as well in the presence of gramicidin as in its absence. However, when the amino acid concentrations were reduced to 0.1 mM, gramicidin severely inhibited both the rate and

the extent of growth, by comparison with control cells. Presumably these stringent nutritional conditions require the cells to accumulate amino acids against the concentration gradient, a process that depends upon the proton circulation (2, 6, 10, 11).

The disabilities of gramicidin-grown cells are consistent with the known roles of the proton circulation in the transport of metabolites and the maintenance of the cytoplasmic pH(1-3). Under conditions that compensate for these disabilities, S. faecalis can grow normally in the presence of ionophores. The proton circulation is thus dispensable for the construction and replication of a bacterial cell. The finding of Kopecky et al. (13) that Escherichia coli "killed" by colicin K grow when supplied with excess K⁺ and Mg²⁺ suggests that this conclusion may be generally applicable. By contrast, there is mounting evidence (14) that many eukaryotic microorganisms rely upon ion currents localized in space to guide their growth and development.

> FRANKLIN M. HAROLD JENNIFER VAN BRUNT

Division of Molecular and Cellular Biology, National Jewish Hospital, Denver, Colorado 80206, and Department of Microbiology, University of Colorado Medical Center, Denver 80220

References and Notes

- P. Mitchell, Biol. Rev. Cambridge Philos. Soc. 41, 445 (1966).
 B. A. Haddock and C. W. Jones, Bacteriol. Rev. 41, 47 (1977); F. M. Harold, Curr. Top. Bioenerg. 6, 83 (1977); P. Mitchell, Biochem. Soc. Trans. 4, 399 (1976); R. D. Simoni and P. W. Postma Annu Rev. Biochem. 43, 533 ос. Trans. 4, 399 (1976); К. D. Simoni and W. Postma, Annu. Rev. Biochem. 43, 523 (1975)
- 3.
- (1975). F. M. Harold and D. Papineau, J. Membr. Biol. 8, 27 (1972); *ibid.*, p. 45. Composition of NaTY (per liter): 8.5 g of Na₂HPO₄, 10 g of tryptone, 5 g of yeast extract, 10 g of glucose; pH 7.5. Composition of KTY2XH (per liter): 20 g of K_2 HPO₄, 20 g of truncione 10 g of Heppes 4. K112AR (per mer): 20 g of K₂HrO₃, 20 g of tryptone, 10 g of yeast extract, 12 g of Hepes, and 10 g of glucose.
 5. D. A. Haydon and S. B. Hladky, *Q. Rev. Biophys.* 5, 187 (1972); F. M. Harold and K. Altordo Gurra Course To Morek T. Terdori, C. 1970.
- tendorf, Curr. Top. Membr. Transp. 5, (1974 6. F. M. Harold and E. Spitz, J. Bacteriol. 122, 266
- 7. Buffers used: bicine, N,N'-bis(2-hydroxyethyl)glycine; tricine, *N*-tris(hydroxymethyl)-methylglycine; Hepes, *N*-2-hydroxyethylpiper-
- azine-N'-2-ethanesulfonic acid. P. C. Maloney, E. R. Kashket, T. H. Wilson, in Methods in Membrane Biology, E. D. Korn, Ed. 8. (Plenum, New York, 1975), vol. 5, p. 1; H. Rot-tenberg, J. Bioenerg. 7, 63 (1975). enberg, J. Bioenerg. 7, 63 (1975). P. C. Laris and H. A. Pershadsingh, Biochem.
- P. C. Laris and H. A. Pershadsingh, Biochem. Biophys. Res. Commun. 57, 620 (1974).
 E. R. Kashket and T. H. Wilson, Proc. Natl. Acad. Sci. U.S.A. 70, 2866 (1973); Biochem. Biophys. Res. Commun. 59, 879 (1974).
 S. S. Asghar, E. Levin, F. M. Harold, J. Biol. Chem. 248, 5225 (1963).
 G. Toennies and G. D. Shockman, Arch. Bio-chem. Biophys. 45, 447 (1983).
- 13.
- G. Ioennies and G. D. Snockman, Arcn. bio-chem. Biophys. 45, 447 (1953). A. L. Kopecky, D. P. Copeland, J. L. Lusk, Proc. Natl. Acad. Sci. U.S.A. 72, 4631 (1975). R. Nucitelli and L. F. Jaffe, Annu. Rev. Biophys. Bioeng. 6, 445 (1977); F. M. Harold, Annu. 14.
- *Rev. Microbiol.* 31, 181 (1977). Supported by PHS grant AI 03568 to F.M.H. We thank Dr. A. Staehelin for preparing the 15. electron micrographs.
- 18 February 1977; revised 29 March 1977