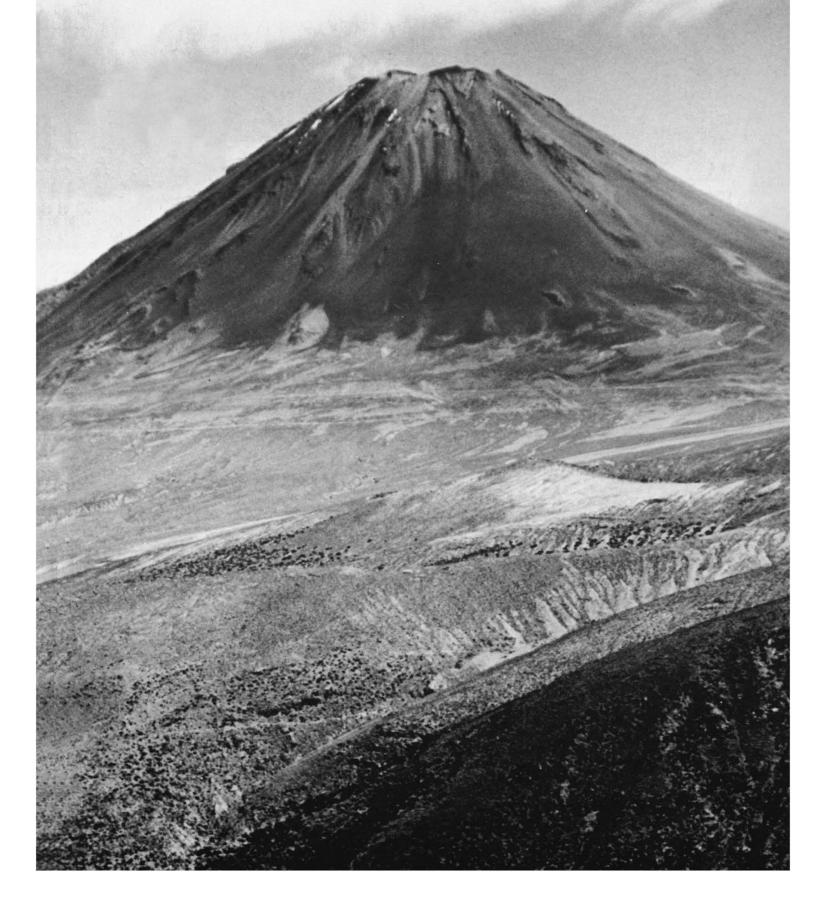


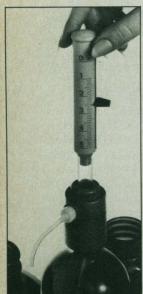
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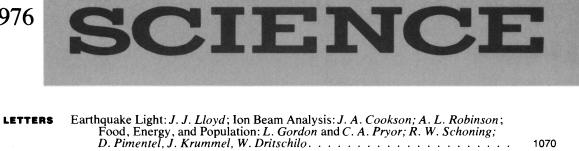
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## 17 September 1976

Volume 193, No. 4258



EDITORIAL	Federal R & D and the Economy: T. D. Long	1079
ARTICLES	Opioid Peptides (Endorphins) in Pituitary and Brain: A. Goldstein	1081
	Ancient Lithosphere: Its Role in Young Continental Volcanism: C. Brooks, D. E. James, S. R. Hart	1086
	Hyperlipidemia and Atherosclerosis: R. Ross and L. Harker	1094
NEWS AND COMMENT	NIH Budget: Senate Committee Holds History's Quietest Inquiry	1100
NEWS AND COMMENT	R & D and Economic Growth: Renewed Interest in Federal Role	1101
	Environmental Research: EPA Plan Termed Myopic.	1103
	Helping the Dying Die: Two Harvard Hospitals Go Public with Policies	1105
RESEARCH NEWS	Rerefined Oil: An Option That Saves Oil, Minimizes Pollution	1108
BOOK REVIEWS	The Conquest of Will, <i>reviewed by K. C. Laudon</i> ; Population Genetics and Ecology, B. R. Levin; The Nile, T. M. Zaret; The New World Primates, K. E. Glander; Organ Culture in Biomedical Research, M. H. Hardy; Books Received; Book Order Service	1111
REPORTS	Solar Neutrinos: Proposal for a New Test: M. S. Freedman et al	1117
	Aluminum-26 in Deep-Sea Sediment: J. L. Reyss, Y. Yokoyama, S. Tanaka	1119
	Density Maxima in High-Pressure Supercooled Water and Liquid Silicon Dioxide: C. A. Angell and H. Kanno	1121

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New Radar Image of Venus: D. B. Campbell, R. B. Dyce, G. H. Pettengill	1123
Uptake and Continued Metabolic Activity of <i>Azotobacter</i> Within Fungal Protoplasts: K. L. Giles and H. Whitehead	1125
Role of <i>Erythronium americanum</i> Ker. in Energy Flow and Nutrient Dynamics of a Northern Hardwood Forest Ecosystem: <i>R. N. Muller</i> and <i>F. H. Bormann</i>	1126
Dinoflagellates: Fossil Motile-Stage Tests from the Upper Cretaceous of the Northern New Jersey Coastal Plain: F. E. May	1128
Calcium Release from Skeletal Muscle Sarcoplasmic Reticulum: Site of Action of Dantrolene Sodium?: W. B. Van Winkle	1130
Photochemotherapy: Identification of a Metabolite of 4,5',8-Trimethylpsoralen: B. B. Mandula, M. A. Pathak, G. Dudek	1131
H-Y (Male) Antigen: Detection on Eight-Cell Mouse Embryos: C. J. Krco and E. H. Goldberg	1134
Coronary Arterial Smooth Muscle Contraction by a Substance Released from Platelets: Evidence That It Is Thromboxane A <sub>2</sub> : <i>E. F. Ellis</i> et al	1135
Microbial Degradation of Condensed Tannins: W. D. Grant	1137
Behavioral Fever in Newborn Rabbits: E. Satinoff, G. N. McEwen, Jr., B. A. Williams	1139
Hardy-Weinberg Law: Asymptotic Approach to a Generalized Form: A. E. Stark	1141
The Effect of Stimulus Sequence on the Waveform of the Cortical Event-Related Potential: K. C. Squires et al.	1142
Drag Reduction by Formation Movement in Spiny Lobsters: R. G. Bill and W. F. Herrnkind	1146
Handedness in a Chinese Population: Biological, Social, and Pathological Factors: <i>E. L. Teng</i> et al	1148
Malaria: Successful Immunization Against the Sexual Stages of <i>Plasmodium</i> gallinaceum: R. W. Gwadz	1150
Myosin Synthesis Increased by Electrical Stimulation of Skeletal Muscle Cell Cultures: A. Brevet et al.	1152
Technical Comments: Need for a Better Solar Radiation Data Base: R. W. Durrenberger and A. J. Brazel	1154

#### PRODUCTS AND Materials

	CHAUNCEY STARR CHEN NING YANG	WILLIAM T. GOLDEI Treasurer	N	WILLIAM D. CAREY Executive Officer	
GEOLOGY AND GEOGRAPHY Helen L. Cannon Ramon E. Bisque	(E) BIOLOGICAL SCIEL Edwin L. Cooper Jane C. Kaltenbach			POLOGY (H) Mandelbaum Ish	COVER
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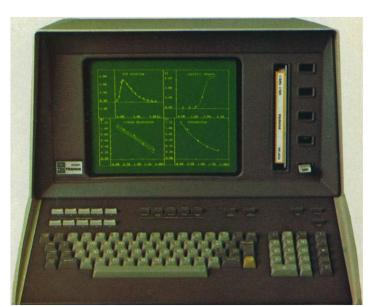


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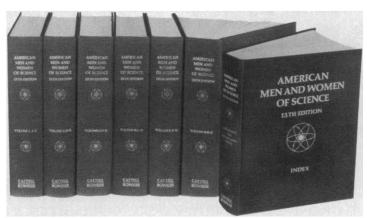
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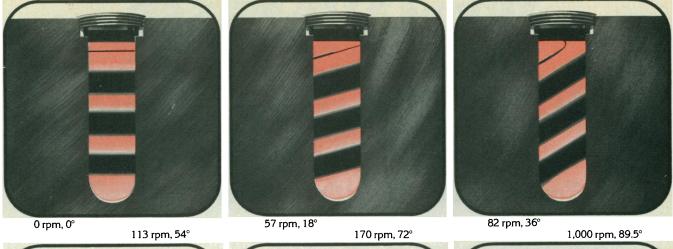
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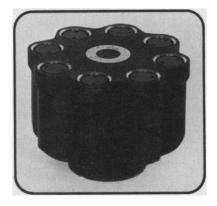
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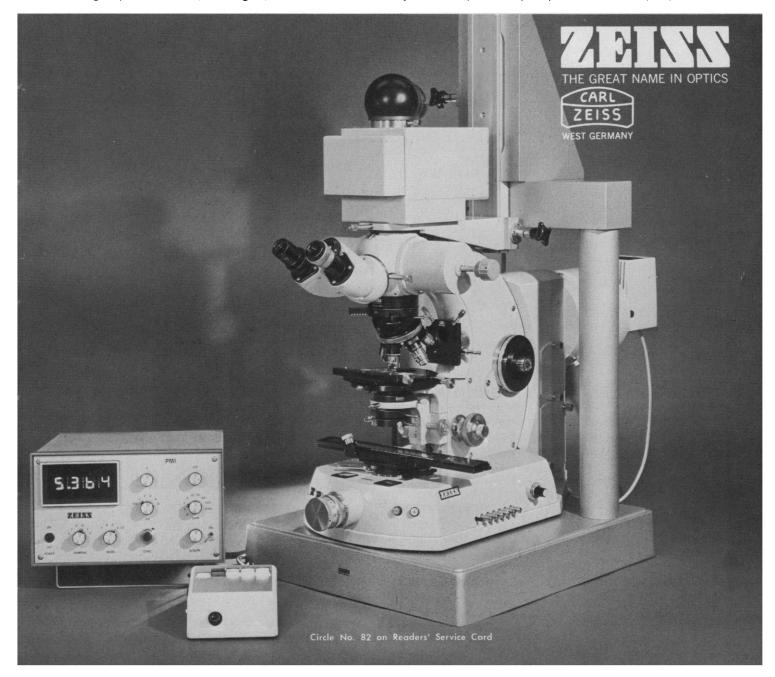
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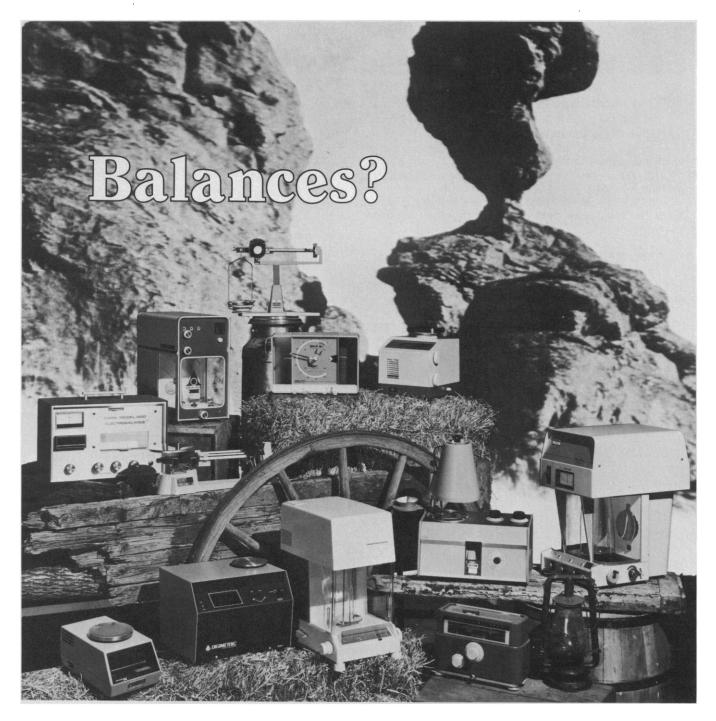
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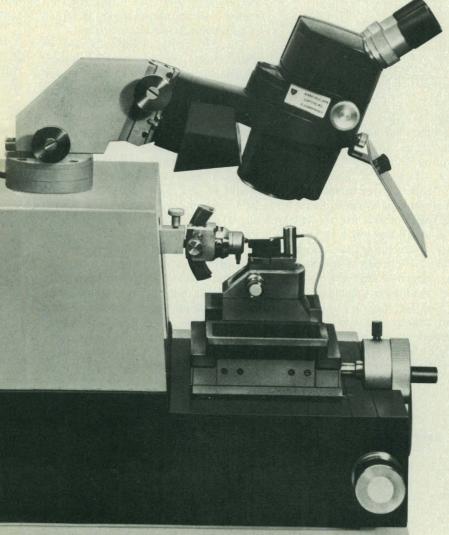
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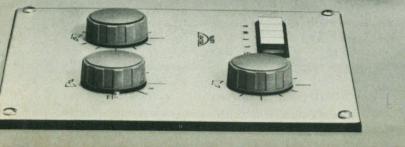
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preservation, and storage. Elimination of waste in processing can also increase yields substantially. While serious technological and economic problems make it difficult to predict the future for aquaculture, the possibility of significant increases by the year 2000 (or later) cannot be discounted.

In other words, a 200-million-ton catch appears quite possible by the year 2000; it could provide at least 28 million tons of protein. A 300-million-ton catch could provide at least 42 million tons of protein, about equal to the amount projected for livestock in 2000.

A last thought relates to the fossil energy needs for catching fish. Estimates given by Pimentel et al. are at odds with Steinhart and Steinhart's estimates (2) that coastal fishing breaks even on calories expended versus the calorie yield, and that even relatively inefficient distant water fishing consumes less than 20 calories for each calorie of yield; they also conflict with Hirst's conclusion (3)that fish is one of the most efficient of several major food groups in terms of the ratio of primary energy use to protein content

In sum, Pimentel et al. have probably seriously underestimated the contribution that aquatic foods can make to the world food supply.

**ROBERT W. SCHONING** National Marine Fisheries Service, National Oceanic and Atmospheric Administration, Department of Commerce, Washington, D.C. 20235

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p. 116.
2. J. S. Steinhart and C. E. Steinhart, *Science* 184, 307 (1974).
3. E. Hirst, *ibid.*, p. 134.

The projection by Schoning of a total annual fish yield of 200 million metric tons for the year 2000 is about double that of other projections (1). We hope Schoning's projection is correct.

Schoning questions the 20 calories energy input per calorie of fish-protein output. In spite of "Hirst's conclusion," Hirst (2) reported an input of 27 calories energy input per calorie of fish-protein output. Hirst's estimate is actually higher than the estimate we quoted (3) of 20 calories input per fish-protein calorie and the estimate of Edwardson (4) of about 21 calories input per fish-protein calorie. A recent analysis by Rochereau and Pimentel (5) concerning the Northeast fisheries of the United States suggests that Northeast fisheries are far more efficient than other fisheries. For the Northeast

SCIENCE, VOL. 193

fisheries which use small fishing vessels, we calculated 5 calories input per calorie of fish-protein produced or at least  $\frac{1}{4}$  the energy expenditure of the other fisheries mentioned.

Gordon and Pryor are concerned about our extrapolations. We emphasized in our 1975 article that extrapolations which involve the interdependencies of food, population, energy, land, water, and other environmental resources are fraught with many difficulties. This is especially so when no one set of acceptable assumptions exists for any of the factors. Our assumptions were stated and the resulting extrapolations were made to provide some perspective about energy and land constraints as they relate to feeding a rapidly growing world population. Based on our selected set of assumptions, we projected that "If petroleum were the only source of energy for food production and if we used all the petroleum reserves solely to feed the world population [4 billion], the 66,053 billion liter reserve would last a mere 13 years.'

With another set of assumptions or coefficients, the 13 years might have been doubled to 26 years. Extending the supply of a resource from 13 years to 26 years might be consoling to our generation, but neither projection is consoling for future generations. As we stated in our articles, "Numerous estimates . . . are possible with the use of various combinations of population size, dietary standards, and production technology," but the one selected sufficed "as an *example* of the limitations" that face mankind.

We fully recognized the fact in both our articles that about half of the energy budget for corn replaces labor, and the other half is for productivity (see also  $\delta$ ). This is documented and was the reason we focused much of our attention on labor inputs and alternatives.

Petroleum and natural gas are the two important energy resources utilized by agriculture. The reserves of both of these resources are in short supply. Petroleum was often used because it is a convenient, familiar measure. Coal, if used for agriculture in the future, will compete with agriculture for a vital agricultural resource—land (6).

Gordon and Pryor question our using corn as an average crop for an energy input-output analysis and extrapolation. They base their argument strictly on the fact that corn requires 2.5 times more nitrogen than the average of other crops. They apparently did not examine any of the data in our table 1 of our 1975 article. If they had, Gordon and Pryor would 17 SEPTEMBER 1976

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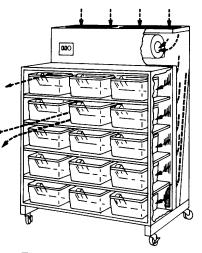
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have found that corn requires "roughly 1 gallon of gasoline per bushel of corn"; oats, 0.6; wheat, 1.2; and soybeans, 1.6 gallons per bushel. More crops could be mentioned, but the point is that when all the inputs are taken into consideration and not just nitrogen, corn is a good average crop.

Gordon and Pryor attribute to us the conclusion that "population control in densely populated, low-income regions is of the highest priority." This is not our conclusion. We stated that overpopulation is a problem for the world as a whole including the developed regions.

Then turning to energy used in the U.S. food system, Gordon and Pryor claim that only "one-quarter" of the energy in the "food system" is used in agricultural production. They give as their source an estimate that was published in 1974 (7), but this estimate is based on no better data than our estimate of "one-third." In addition, the U.S. Senate Committee on Agriculture and Forestry (8) reported that food production is more energy-intensive than either food processing or food distribution based on the percentage of energy costs compared with total costs of the activity.

Our extrapolations. although "simple," were based on "realistic coefficients" that were selected to call attention to the fact that high energyintensive U.S. agricultural technology is generally inappropriate for "green revolution agriculture." We hope that others will examine the relationships that exist among food, energy, and population, using the "numerous estimates" or coefficients that are available. Unfortunately, no matter how the coefficients are manipulated either by Gordon and Pryor, by us, or by others, the "real problems of the real world" will not disappear.

> DAVID PIMENTEL JOHN KRUMMEL WILLIAM DRITSCHILO

College of Agriculture and Life Sciences, Cornell University, Ithaca, New York 14853

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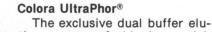
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### Federal R & D and the Economy

Hearings cross the congressional firmament like shooting stars. Most are only a momentary glimmer. A few signal the existence of larger issues. The hearings held in April and May by the Thornton subcommittee of the House Committee on Science and Technology were of the latter type. The subject was "Federal R & D Expenditure and the National Economy."

The issue revolves around the unanswered question whether the federal government should assure stable, long-term support for commercially oriented R & D. Some think such intervention is essential to the health and international competitiveness of the national economy. Yet most of our understanding of the relationship between R & D and the economy, with the important exception of agriculture, comes from 35 years of experience with defense- and space-oriented projects-of which the Manhattan and Apollo projects were spectacular examples-in which the federal government was both the funder and principal consumer of the R & D output.

There were four persistent themes in these hearings. One was that government and business operate in a climate of distrust, if not of hostility. Another was that R & D expenditures seem a weak and imperfect tool in comparison with tax, subsidy, antitrust regulation, education, and procurement policies. A third theme was the perceived negative effect of federal organization and management of its own R & D enterprises on several dependent sectors of the economy. Finally, the need for federal support of fundamental research was a clear and consistent theme. Every witness, regardless of philosophy, expressed deep concern for the health of the fundamental research effort. Apparently the case still has not been made that investment in fundamental research is not a discretionary societal luxury but an essential component of continued economic viability.

The sense of these hearings was that specific government intervention through R & D expenditure to stimulate the private sector is liable to be of limited effectiveness. There is little understanding or guidance available regarding what to do to achieve a particular desired result. Some opportunities stood out for analysis and perhaps experiment:

• Firms that perform R & D exclusively appear to play a critical role in the process of innovation. Differences in performance between firms created through federally supported R & D need to be explored.

• Small, highly innovative firms are important contributors of new products and processes, but are also most subject to failure in the course of normal economic fluctuations. Some alternatives need to be considered to cushion this impact.

• Policies and programs of other industrial countries are admired, but none has been tested in this country. Some selective, controlled experiments ought to be possible, at least on a limited basis.

• Innovation in the private, profit-making sector is a reasonably wellunderstood process. Similarities and differences in public sector innovation are not nearly as well described or explained.

What seems to be foreshadowed by the kinds of questions raised in these hearings is the broad issue of a national technology policy. Given our penchant in the United States for pluralistic, decentralized approaches to complex and important policy issues, clear-cut institutional structures or guidelines to action are not likely to appear. The least we should expect, however, is that existing centers of authority and initiative begin to review and refine their objectives. This would include the Departments of Agriculture, Commerce, and Defense. The mission agencies-ERDA, NASA, and NIH-also are important constituents. Where these lead, others will follow.-T. DIXON LONG, Provost, Western Reserve College, Case Western Reserve University, Cleveland, Ohio 44106

## SCIENCE

This editorial has been abstracted from a review of the hearings prepared by a subcommittee of the AAAS Science and Public Policy Committee. In addition to Mr. Long, its chairman, the subcommittee included John M. Logsdon and Edward E. David, Jr.

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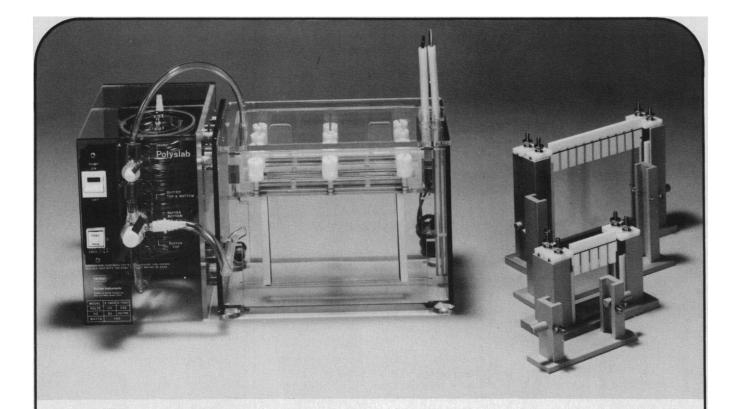
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## Now, one P.A.G.E. unit that can handle all four investigative techniques.

Analytical

Preparative

**Two Dimensional** 

Gradient

## The Buchler Polyslab for gel electrophoresis.

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For several months Buchler engineers analyzed every facet of P.A.G.E. technology. They listened and learned as feedback kept coming in from independent investigators like yourself. They discovered what your P.A.G.E. resolution problems were, and they worked to resolve them. The result is a gel electrophoresis instrument that outperforms anything in the field today.

Buchler's Polyslab can actually handle all four polyacrylamide electrophoresis techniques: analytical, preparative, two dimensional and polyacrylamide gradient electrophoresis. And its mounting and polymerization apparatus has been precision engineered for easier gel preparation. Just ask your Buchler dealer to give you a close look at the Polyslab today. You'll find Buchler put a lot of good thinking into the Polyslab, so you can get better results out of it.

Features include: Choice of two gel sizes; precision tooled mounting and polymerization stands; 3-way buffer circulation; optional tube adapters; heat transfer coil; and interlocking electrode plugs.

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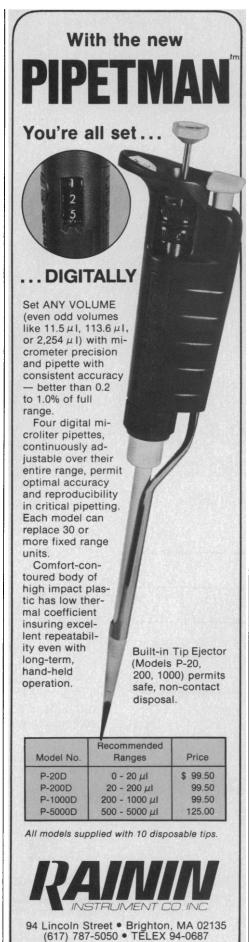
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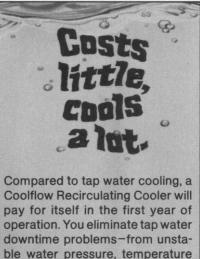


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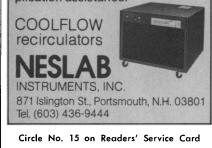
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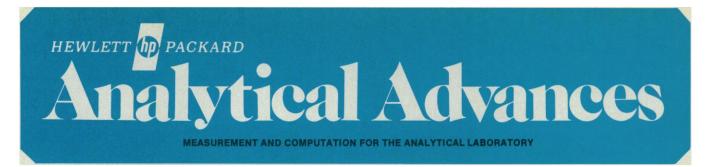


## ENERGY: Use, Conservation, and Supply

Interested in understanding the energy problems we face, in finding practical solutions to these problems? Then you're sure to want a personal copy of this AAAS compendium. It presents a collection of enduring and readable articles that originally appeared in *Science* during 1973–1974 articles providing a wealth of information for everyone concerned with the profound and far-reaching effects of the energy problems that continue to alter our traditional modes of supply and patterns of usage.

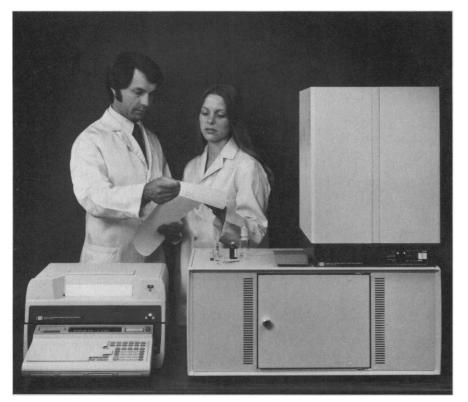
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## **New Benchtop GC/Mass Spectrometer System**

with calculator control, hyperbolic quadrupole filter, and automatic tuning



Hewlett-Packard announces the 5992A Gas Chromatograph/Mass Spectrometer system, the first such system that can be conveniently added to a laboratory just by placing it on a benchtop, yet with performance features comparable to (and often exceeding) those of much larger floor-mounted systems.

These features include a microprocessor-controlled gas chromatograph as an integral part of the system, an efficient hyperbolic rod quadrupole mass analyzer, and automatic tuning of the mass spectrometer components.

The analyzer module (right, in the photograph) contains the gas chromatograph and mass spectrometer, including all necessary pumping systems. It has none of the usual knobs or switches to set, but is controlled by a 9825A Programmable Calculator, which uses a combination of keystroke instructions and stored programs for efficient operation of the entire system. For example, after admitting a reference compound, the program AUTO-TUNE automatically finds and sets the optimum tuning parameters for the ion source, the mass filter, and the ion detector, then produces a standard mass spectrum with a calibrated mass scale. The use of AUTOTUNE ensures consistent analytical results.

The 5992A's quadrupole mass filter uses hyperbolic rods, the theoretically ideal shape for this type of filter. Improved peak shapes and higher sensitivity, compared to round-rod filters, are direct consequences of this geometry. For 1 nanogram of methyl stearate, scanned at 600 atomic mass units (amu) per second, the system produces a signal to noise ratio of 10:1. Total mass range is 10 to 800 amu.

The entire mass spectrometer assembly is mounted on a single flange and suspended in the core of an HP designed diffusion pump. Only one gasket, maintained at room temperature, is required. Troublesome flange bolts have been completely eliminated. Pumpdown and vacuum venting cycles are completely automated to preclude inadvertent damage to spectrometer components. Indicator lights show when the system is ready to be operated or disassembled.

A membrane (standard) or jet (optional) separator interfaces the gas chromatograph to the mass spectrometer. The dual-filament electron-impact ion source incorporates a Turner-Kruger lens to improve the transfer of higher mass ions into the mass filter.

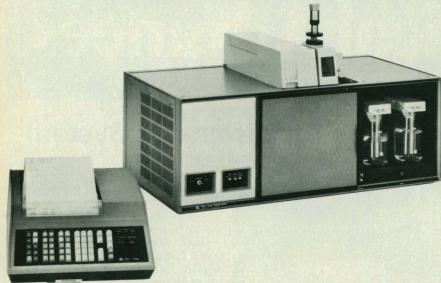
Standard data acquisition and display software provides normal spectrum generation, total ion chromatograms, tape cartridge storage and later display of individual spectra, and selected ion monitoring. Standard data output is to a thermal printer/plotter mounted on top of the 9825A calculator. A flatbed X-Y plotter is available as an option. The 9825A calculator may be used as a free-standing computation device when it is not functioning as the system controller.

A new product bulletin describes the benchtop GC/Mass Spectrometer in greater detail. Please check 5992A GC/MS System on the Reply Card for your copy.

#### IN THIS ISSUE Fall 1976

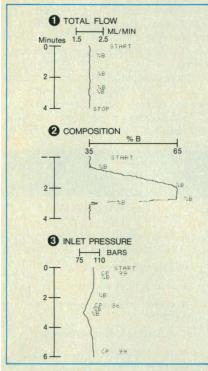
- New Bench-Top GC/MS System
- Plotting with 1080 Liquid Chromatographs
- Expanding Your Data
- Managing Workloads
- Glass Systems Improve Peaks
- Analyzing Organics in Water
- Troubleshooting Transformers

## See What's Happening with HP 1080 Liquid Chromatographs



A lot goes on inside a liquid chromatograph, particularly during gradient operation, but very little of it shows. Yet a close look at some of those internal variables can be of great help when creating and optimizing an analytical procedure. That's why the 1080's have a key labeled PLOT; it lets you display on the printer/plotter any of these quantities

- Total flow, the sum of the measured outputs of the two pump heads
- Individual pump head deliveries
- Mobile phase compositon, expressed as the percentage of the B solvent.



HP 1080 liquid chromatographs automatically plot critical internal variables. Again this is calculated from the measured flows

- Column inlet pressure
- Solvent reservoir temperatures
- Detector output, either the differential signal or the individual cell outputs

and several others.

The figures show some of the possibilities. The gradient program calls for an initial isocratic period (at 35% B) followed by a fairly steep rise to a second isocratic period at 65% B; composition then returns to the starting value to regenerate the column for the next run. The figures are instrumentproduced plots of the total measured flow (1), the mobile phase composition (2), and the pressure in bars at the column inlet (3) (solvent B is of lower viscosity than solvent A so the pressure profile is inverted relative to the composition profile). The time offset between the composition and pressure profiles is due to the hold-up volume of the column and the chemical equilibration time.

Note that the column pressure returns to its original value at the end of the plot (the three pressure printouts were produced with the LIST COL P keys). This indicates that the regeneration time chosen is adequate for this column/solvent combination.

The utility of the PLOT function in troubleshooting is obvious. Obstructions in the capillary flow lines, plugging or deterioration of the column, accumulated deposits in the detector cells; these are all readily detected by plotting the appropriate variable. A data sheet containing further information on these and other capabilities of the 1080 Series Liquid Chromatographs may be obtained using the Reply Card.

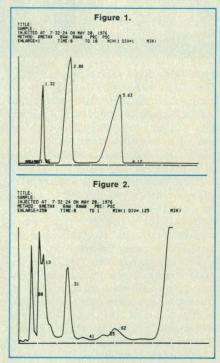
## Blow Up Your Chromatogram!

There's no substitute for a picture when you develop an analytical method or encounter problems with an established one. But the picture need not be on a piece of paper; Figure 1 shows a chromatogram displayed on a 3354 Lab Automation System terminal.

It's done using a program called CGRAM and a Tektronix 4006-1 graphics display. CGRAM examines the raw data file for the analysis (which contains every reading taken from the analyzing instrument) and reconstructs the chromatogram, scaled to the largest peak in the run. You can examine the entire analysis or any selected portion; scaling remains the same so that valid comparisons can be made.

But notice the blurred print in the lower left part of Figure 1. The system attempted to plot several very small peaks, too small to be seen, which are very close together. By expanding the time axis (plotting only the first minute) and the vertical axis (entering an enlargement factor of 350) we obtain Figure 2, which clearly shows these previously invisible small peaks. This can be repeated as often as desired. The effect is much like running the analysis at many chart speeds and attenuations, with the important difference that you only make one actual run!

For more information on CGRAM and the 3354 System which uses it, please check 3354 Lab Automation System on the Reply Card.



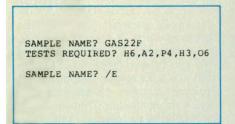
Graphics display adds flexibility to data examination.

## Managing Lab Work Loads with the 3354 Lab Automation System

A sample comes into the lab, all required tests are performed, and the report goes back out. That's the ideal situation; the real one is seldom so simple. Generally there are many samples in progress, often with different sets of tests to be run, and all at different stages of completion. Keeping track of sample progress can be a very difficult but necessary task in effectively running a laboratory.

However, it can be done quite simply using LAB BASIC II programming. Here's how:

**1.** As samples arrive in the lab, run LOGGER and enter the pertinent data. Terminate with /E.

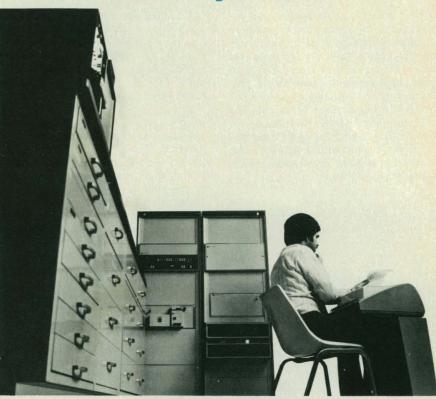


2. Whenever you need to know sample progress, run REPORT. You'll receive this three-part report.

	SA	MPLE PROGRES	S REP	ORT	
	SAMPLE:	GAS22F			
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	TESTS	PENDING:	н3	06	
1					
	SAMPLE:				
1		COMPLETED: PENDING:	A4		
1	IESIS	PENDING	44	8.3	К9
	SAMPLE:	REM906D			
ļ	TESTS	COMPLETED:	86	H6	
1	TESTS	PENDING:	L8	D7	U7
	SAMFLE:	REM906E			
	TESTS	COMPLETED:	F6		
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	Н3		S22F LV558	1	
1	06	GA	S22F		
	A4	SO	LV558	1	
l	К9	50	LV558	1	
	L8	RE	M906D		
1	D7		M906D		
	U7	RE	M906D		
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		COMP	LETED	SAM	PLES
	WORK SINC SAMP	ON THE FOLL E THE LAST S LES HAS BEEN	OWING TATUS REMO	SAM REP VED	PLES WAS COMPLETED ORT. DATA ON THESE FROM SYSTEM MEMORY.
	SAMPLE NA	ME TESTS	PERFO	RMED	
	REM906E	86			
			1		

The first portion gives the current status of every sample logged into the system. Part two, the backlog report, shows the sample load and can give advance warning of possible trouble

**17 SEPTEMBER 1976** 



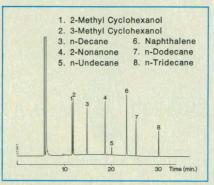
areas. The completed samples report "logs off" a sample when all tests are finished.

The difficult work is performed by a third program called LOGUP (for update). This runs at the end of each analysis to determine the sample name and the test performed from the run data itself, then updates the information in system memory to reflect this status change. Notice that this is completely automatic; there is no need for you to inform the system that the test has been run.

This is just one example of the laboratory management aspect of the 3354 Lab Automation System. To learn more, please use the Reply Card.

## No Volume+No Surface=Perfect Peaks

That's hardly a practical way to approach chromatography, but if we modify the requirements a bit we can come surprisingly close. Instead of "no volume" we'll use "no unswept volume", and we'll substitute "low surface activity" for "no surface". The chroma-



Glass system yields symmetry and resolution. togram shows what can be achieved under these somewhat relaxed (but more realistic) rules.

Note the symmetry of the hydrocarbon peaks (3,5,7,9). This demonstrates that there is little or no unswept volume in this all-glass system. Low surface activity is shown by the very small degree of tailing on the alcohol (1,2), ketone (4), and aromatic (6) compounds.

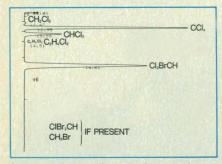
This separation was performed using the all-glass capillary injector designed for the HP 5700 and 5800 Gas Chromatographs. A glass capillary column was essential, of course, so that the benefits of this component would not be negated by the effects of a metal capillary wall.

For additional information on the injector and instruments in which they are used check All-Glass Systems on the Reply Card.

## **Solving Problems with the 5840A:**

#### **1. Volatile Organics in Water**

Recent evidence of the widespread occurrence of chlorinated and brominated organics in finished drinking water supplies and the Safe Water Drinking Act of 1974 have created a need for large scale, high sensitivity monitoring procedures. Direct injection of water samples is injurious to many of the columns used and may interfere with the detection process. Using the 5840A Reporting Gas Chromatograph, an automatic method has been developed which avoids these problems.



Automated procedure analyzes halocarbons in water. Volatiles are transferred from the water sample to a room temperature trap by a stream of carrier gas. When transfer is complete the trap is heated and backflushed into the analyzing column. Trap regeneration occurs automatically while the analysis proceeds.

A total of 8 valve operations and 4 temperature changes (in addition to any oven temperature program) are required. This would be a formidable task by manual operation, but with the 5840A all of these events are controlled by the built-in microprocessor. With a 10 ml sample and electron capture detection, a part-per-million analysis of the seven compounds of major interest is obtained in 30 minutes.

An Application Note is now available which discusses this procedure in detail. For your copy please check Water Analysis on the Reply Card.

## 2. Gas in Transformer Oils

Large oil-filled power transformers have very low failure rates. However, if failure should occur the repair is likely to be both expensive and time consuming. Most incipient failures (overheating, sparking, arcing, etc.) produce decomposition gases which are characteristic of the type of fault. Analysis of the gases can detect a problem long before serious damage occurs, and can indicate the extent of disassembly necessary to correct the fault.

The Model 5840 Reporting Gas Chromatograph with Option 835 is fully equipped with the appropriate columns, valves, reactors and other components necessary for the gas analysis. The sample may be either the headspace gas above the oil or dissolved gases extracted from it. The latter type of sample is preferred since the gases of interest are highly soluble in oil.

As can be seen from the chromatogram, the same system is applicable to a number of other gas analysis problems. A check mark by Transformer Oil Gas will bring you a recently published Application Note on this analysis, plus information on the 5840A itself.

	Hydrogen	
Oxygen		- 1.0
5	C-10310615 5 1	Nitrogen
1		Methane
-	Carbon Mor	noxide
£ *	- Carbon Dioxide	Ethylene
Ē		Ethane
1		Acetylene
5		Propylene

Dissolved gases in transformer oils help diagnose possible problems.



HP 5840 A Reporting Gas Chromatographs with microprocessor control of internal and external events permit ready automation of difficult analytical procedures.



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For more information call your local HP Sales Office or East (301) 948-6370 • Midwest (312) 677-0400 • South (404) 434-4000 • West (213) 877-1282 • Canada (416) 678-9430. Or write Hewlett-Packard, 1501 Page Mill Road, Palo Alto, CA 94304. In Europe: Hewlett-Packard S.A., P.O. Box 85, CH-1217 Meyrin 2, Geneva, Switzerland. In Japan: Yokogawa-Hewlett-Packard, 1-59-1, Yoyogi, Shibuya-Ku, Tokyo, 151.



## **NEW!** RAPID MOLECULAR WEIGHT OF BIO-POLYMERS · Microsamples · Easy Clarification · Absolute Calibration

Chromatix, world leader in tunable lasers, announces the KMX-6, a unique instrument with absolute calibration that permits rapid direct measurement of molecular weight. The KMX-6 is a low-angle light scattering photometer which may be used with a variety of flow, mixing, or microsample cells. The rapid direct measurement capability of the KMX-6 not only permits determination of molecular weight on pure liquids and solutions, but also allows real-time measurement of molecular weight changes in kinetic systems as well.

Revolutionary in design, the Chromatix KMX-6 capitalizes on the spatial characteristics of laser illumination to overcome the classical problems encountered with light scattering measurements. In particular, the excellent sensitivity of the KMX-6 allows the use of very small and dilute samples. Also, requirements for sample clarification are greatly relaxed due to the extremely small volume illuminated by the laser.

Capability of the KMX-6 covers molecular weights from 1000 to 3 x 10<sup>9</sup>. Applications extend to a wide variety of areas, including biochemistry, photochemistry, immunology, pharmacology, polymers, elastomers, and foods.

## chromatix

1145 Terra Bella Avenue Mountain View, CA 94043 Phone: (415) 969-1070 Telex: 910-379-6440 Central: 5109 Brown St. Skokie, IL 60076 Phone: (312) 679-4006 East: Tarragon Building – Rm. 122 811 Church Road Cherry Hill, NJ 08002

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R-30

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S-2

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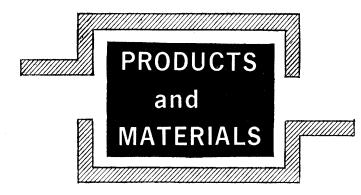
There's no set-up fuss. No dials. Simply turn it on, touch the tare bar, and weigh. Answer appears in 3 seconds on readout — large  $\frac{1}{2}$  digits, easy to read in any kind of light.

For the folks who want specifications, consider these: accuracy ±0.01g; reproducibility 0.005g; sensitivity and readability 0.01g; and capacity 200g.

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#### Syringe Needle Cleaning Kit

Each kit contains five wire sizes in tubes of ten wires each. The wires range in diameter from 0.00350 to 0.0120 inch. Also included are 75 milliliters of cleaning solution concentrate, flat-nosed tweezers, and a booklet of instructions. The cleaning solution contains no alkaline substance or harmful detergents, is nonflammable, and will not etch glass or metal. Unimetrics. Circle 678.

#### **Analog-Specific Ion Meter**

Model 407A formerly was available only as a portable unit for field use. Now it has been redesigned for bench-top operation. The two-decade logarithmic scale permits precise direct measurement of concentrations of ions or dissolved gases. Known addition and known subtraction scales simplify the use of increment methods. Titrations and measurements of oxidation-reduction potential may be performed with the millivolt scales. The meter also has a scale for routine pH measurements and an adjustable recorder output. Orion Research. Circle 680.

#### Fluid Dispenser

The Sequential fluid dispenser is for filling scintillation vessels with counting solution. The device is rapid at 25 cycles per minute and accurate to within  $\pm 1.0$  percent. It dispenses from 2 to 10 milliliters per vessel into counting containers. It is suitable for use with any scintillation medium. It features hand operation. Isolab. Circle 682.

#### **Melting Point Instrument**

The model FP61 reads to within 0.1°C over a range from 0° to 300°C. After a capillary tube has been prepared with a sample, the operator selects a starting temperature and one of five heating rates. The capillary is then inserted in the furnace and the "start" button is pressed. The determination is then made. Linear temperature increase is continuously displayed. Heating stops when the sample is melted. The final result is displayed until the unit is turned off or another run is started. A recorder may be attached to produce graphic results of the process. Heating or cooling may be selected from  $0.2^{\circ}$ ,  $1^{\circ}$ ,  $2^{\circ}$ ,  $3^{\circ}$ , or  $10^{\circ}$ C per minute. With the 0.2°C per minute rate, accuracy is  $\pm 5^{\circ}$ C up to 200°C and decreases to  $\pm 0.8^{\circ}$ C at 300°C. Mettler Instrument. Circle 679.

#### **Concentrator-Evaporator**

The Speed Vac Concentrator is designed for rapid solvent, phase elimination. The solid residue or solute is held by centrifugation at the bottom of the tube while the process of evaporation takes place without the bumping of the sample. It is suitable for both aqueous and organic solvents including water, hydrochloric acid, trichloroacetic acid, or ammonium hydroxide. Anhydrous samples may be obtained quickly without the loss of solute. Applications include hydrolysis, silylation, and peptide analysis. Savant Instruments. Circle 681.

#### Liquid Chromatograph

A new gradient system is available for high-performance liquid chromatography. It utilizes two metering pumps with electronic pulse correction for precise flow rates up to 10 milliliters per minute at 10,000 pounds per square inch. An optional version offers up to 28 milliliters per minute at 3,000 pounds per square inch. Two modes of

operation permit constant flow for maximum chromatogram reproducibility or constant pressure where column or packing pressure limits override flow considerations. The solvent programmer can be used to program flow rate or gradient profiles up to 18 hours long. Operating mode selections include manual solvent scouting, up to a 60-minute hold at final gradient concentration, and automatic column equilibration. An ultraviolet-visible detector with an 8-microliter flow cell and wavelength range from 254 to 660 nanometers completes the system. Altex Scientific. Circle 690.

#### **Dialysis** Tubing

Spectrapor dialysis tubing is available with a cutoff at a low molecular weight. It comes in a range of flat-width sizes of 10 to 45 millimeters. It is packaged in rolls 10 meters long. Molecular weight cutoff is 2000. Spectrum Medical Industries. Circle 683.

#### Literature

Severe Environment Products describes computer subsystems and peripherals for use in hostile environments. Electronic Memories and Magnetics. Circle 684.

*Continuous Water Still* depicts a unit that provides more than 2 liters per hour. Included are a redistillation option and a 5-gallon, automatically controlled reservoir. Kontes. Circle 685.

Paired-Ion Chromatography is an alternative to ion exchange. It is described in a 16-page publication that gives details for separating and analyzing ionic compounds and mixtures of ionized and nonionized materials. Waters Associates. Circle 686.

*Photrex Solvents for Spectrophotometry* includes 20 new solvents that brings the total available in this line to 35. J. T. Baker Chemical. Circle 687.

*Standards Catalog* features standards and publications for a wide variety of materials and techniques. American National Standards Institute. Circle 688.

*Exploring Molybdenum and Tungsten* includes physical and mechanical properties and illustrates a number of applications for these materials. Amax Specialty Metals. Circle 689.

SelectaSol Solvent Selector System describes a chromatographic system. It includes hardware, instructions, and suggested applications. Schleicher & Schuell. Circle 691.

Newly offered instrumentation, apparatus, and laboratory materials of interest to researchers in all disciplines in academic, industrial, and government organizations are featured in this space. Emphasis is given to purpose, chief characteristics, and availability of products and materials. Endorsement by *Science* or AAAS is not implied. Additional information may be obtained from the manufacturers or suppliers named by circling the appropriate number on the Readers' Service Card (on pages 1066A and 1162A) and placing it in the mailbox. Postage is free. —RICHARD G. SOMMER

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## PEAK-11...designed to help you automate your laboratory without telling you how to run it.

Some laboratory systems only work one way. But if you need a system that handles a variety of tasks, the Digital PEAK-11 system is designed for you.

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SCIENCE, VOL. 193

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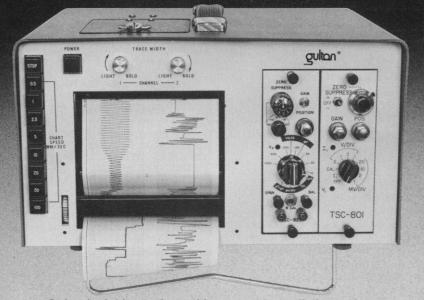
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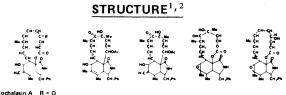
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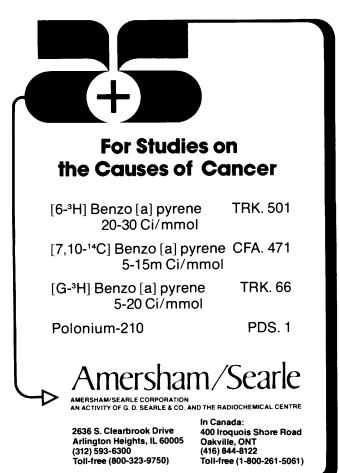
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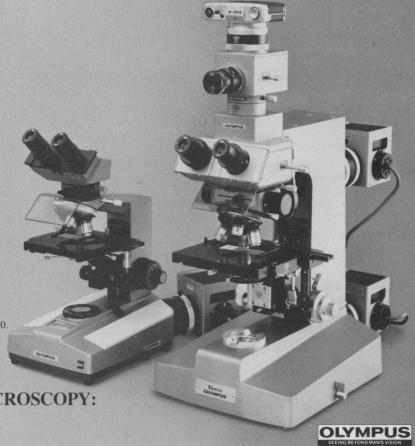
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