

frontalin and racemic *exo*-brevicomin were used, can now be ascribed to the (-) enantiomer of frontalin and the (+) enantiomer of *exo*-brevicomin. Recently, Borden (21) has observed a related phenomenon for the ambrosia beetle, *Gnathotrichus sulcatus*, but in this case both enantiomers of the same attractant pheromone compound [6-methyl-5-hepten-2-ol (22)] have low activity and together they evoke a synergistic response. With both bark beetle species, only the enantiomers that occur in nature are active, at least at the concentrations tested. Therefore we expect both a chiral synthesis and chiral olfactory receptor system in these species.

*Note added in proof:* Recently, (S)-(-)-ipsenol has been reported to be much more attractive to *Ips grandicollis* than its antipode (23).

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## Characterization of the Androgen Receptor from a Syrian Hamster Ductus Deferens Tumor Cell Line (DDT<sub>1</sub>)

**Abstract.** *The hamster ductus deferens cloned tumor cell line (DDT<sub>1</sub>) contains a complex steroid receptor protein that binds <sup>3</sup>H-labeled 5 $\alpha$ -dihydrotestosterone. Diethylaminoethyl (DEAE) chromatography of cytosol from these cells yields two major receptor peaks of activity. Identification of this steroid binding protein as a cytoplasmic receptor was confirmed by salt dissociation on sucrose gradients, stability of the hormone-receptor complex at 0°C, and the retention patterns on phosphocellulose and DEAE cellulose. Multiple forms of the receptor exist in a single homogeneous cell type. The data support the theory that steroid hormones bind to a cytoplasmic protein receptor composed of dissimilar subunits as the initial step in steroid hormone action.*

Steroid hormones are believed to exert their effects on target tissues by a series of events beginning with the combination of the steroid with a specific cytoplasmic receptor in the cell. This hormone-receptor complex is then translocated into the cell nucleus where it binds to the genetic material and initiates the appropriate responses to the steroid such as cell growth (1).

Model systems for studying growth responses to steroid hormones in cell culture have proved exceedingly difficult to

establish. We have developed the DDT<sub>1</sub> cell line (2), from a leiomyosarcoma induced in the ductus deferens of a male Syrian hamster by long-term administration of testosterone propionate and diethylstilbestrol (3). The DDT<sub>1</sub> cell line contains androgen receptors that translocate to the nucleus and exhibits a growth response to androgens. The order of binding affinity is 5 $\alpha$ -dihydrotestosterone (DHT) > testosterone (T) > 17 $\beta$ -estradiol > progesterone. Cortisol and the synthetic glucocorticoids dexamethasone and triamcinolone acetonide do not inhibit <sup>3</sup>H-labeled DHT binding to cytoplasmic receptor. We now report further characterization of the cytoplasmic androgen receptor from this homogeneous cloned cell population.

DDT<sub>1</sub> cells, clone MF-2, were grown at 37.5°C in 6-liter suspension culture vessels to a density of 1.5 × 10<sup>5</sup> cell/ml in a medium consisting of 93 percent Hams-F12, 5 percent fetal bovine serum, penicillin and streptomycin (100  $\mu$ g/ml), and Fungizone (2.5  $\mu$ g/ml) (all from Gib-

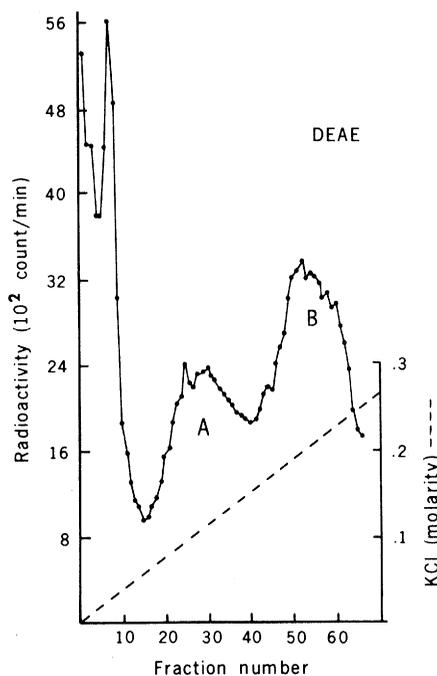


Fig. 1. DEAE-cellulose chromatography of 2 ml of DDT<sub>1</sub> cytosol labeled with [<sup>3</sup>H]DHT and eluted with a 0 to 0.3M KCl gradient. Fractions (2 ml) were collected and the radioactivity in 0.5-ml samples were counted in 4 ml of Aquasol (New England Nuclear) at 25 percent efficiency (Beckman LS-233). Conductivities were determined with a radiometer conductivity cell and converted to KCl molarity. Two peaks of radioactivity were eluted: at 0.12M KCl (peak A) and at 0.23M KCl (peak B).

co). Cells were collected at 4°C, and all further procedures were carried out at 4°C; after collection, the cells were concentrated by centrifugation at 600g. The pellet of cells was then 1:1 in buffer A (tris, 0.01M; Na<sub>2</sub> EDTA, 0.0015M, 1-thioglycerol, 0.012M, pH 7.45) made up in 10 percent glycerol. The cells were allowed to swell for 15 minutes, and then sufficient buffer C (buffer A with 1.0M KCl) was added to reach a final KCl concentration of 0.15M. Homogenization was accomplished by 15 strokes of a Ten Broecke ground-glass homogenizer.

The homogenate was centrifuged at 600g for 10 minutes, and the lipid layer was removed. The remaining cytosol was centrifuged at 150,000g for 1 hour. Tritiated 5 $\alpha$ -dihydrotestosterone (1,2,3,5,6,7-<sup>3</sup>H<sub>6</sub>) (5 $\alpha$ DHT; Amersham/Searle; specific activity 175 c/mmole) was added to the cytosol at this time at concentrations specified in the text, and the solution was allowed to stand for a minimum of 1 hour.

Diethylaminoethyl (DEAE)-cellulose (Whatman DE-52, preswollen) chromatography was performed on cytosol that had been labeled with 5  $\times$  10<sup>-9</sup>M [<sup>3</sup>H]DHT. A column (1.3 by 7 cm) with an exchanger bed height of 3 to 4 cm was washed extensively with 30 to 50 ml of buffer B (buffer A without glycerol) until the conductivity of the eluate remained constant. The cytosol sample (2 ml) in buffer A was diluted with three volumes of buffer A, applied to the column, and washed with buffer A. Elution was accomplished by a 160-ml linear gradient of 0 to 0.3M KCl made up in buffer B.

As is the case for the chick oviduct progesterone receptor (4) and the rat ventral prostate androgen receptor (5), the DDT<sub>1</sub> cytosol eluted from the DEAE-cellulose column to yield two major peaks of radioactivity (Fig. 1). First, peak A eluted at 0.12M KCl; the second, peak B, eluted at 0.23M KCl. This chromatographic behavior indicated the heterogeneous nature of the cytosol androgen receptor in this cell line.

To ensure that both peaks of radioactivity were specific for <sup>3</sup>H-labeled 5 $\alpha$ DHT, we performed competition studies by simultaneously adding a 250-fold excess of unlabeled 5 $\alpha$ DHT, cyproterone acetate (CA), progesterone, 17 $\beta$ -estradiol (E<sub>2</sub>), or triamcinolone acetate (TA) and performed the DEAE chromatography as above. Binding was reduced by the following: unlabeled 5 $\alpha$ DHT by 95.9 percent, CA by 83.8 percent, progesterone by 77.1 percent, E<sub>2</sub> by 86.5 percent, and TA not at all. Reduction of specific binding in the A and B peaks was always equivalent.

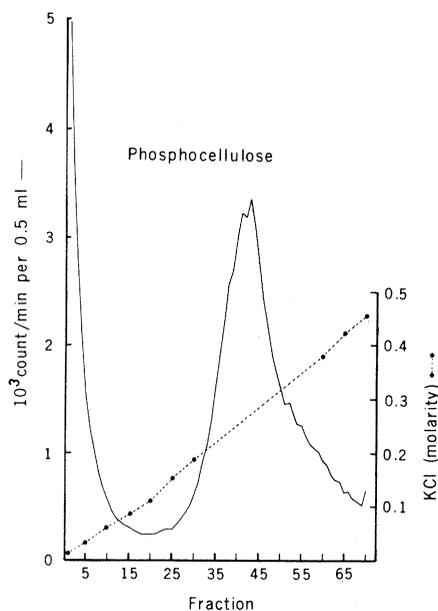


Fig. 2. Phosphocellulose chromatography of 5 ml of DDT<sub>1</sub> cytosol prepared in buffer A and labeled with [<sup>3</sup>H]DHT. Radioactivity and KCl molarity were determined as in Fig. 1. A single peak of radioactivity was eluted at 0.26M KCl with a shoulder at 0.25M KCl.

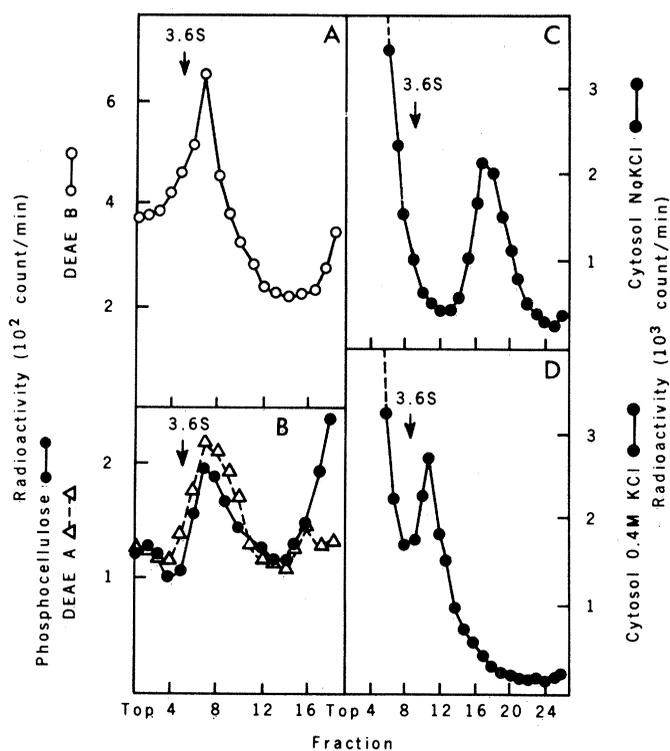
phosphocellulose (8), as is the case with cytoplasmic receptors.

To further characterize the DDT<sub>1</sub> cytoplasmic receptor, we performed phosphocellulose chromatography (9) (Fig. 2), using a column (1.3 by 7 cm) with an exchanger bed height of 5 cm. Cytosol (5 ml) labeled with 5  $\times$  10<sup>-9</sup>M [<sup>3</sup>H]DHT was applied and eluted as described for DEAE chromatography, except that the salt gradient concentration was increased to 0.5M KCl. Under these conditions cytosol labeled with [<sup>3</sup>H]DHT also revealed a receptor peak that was retained on the phosphocellulose and eluted at 0.26M KCl.

In the case of the progesterone receptor the A and B peaks elute separately on phosphocellulose (9); therefore in order to determine whether the DDT<sub>1</sub> A and B receptors coeluted from phosphocellulose producing the single peak of radioactivity, we separated the DDT<sub>1</sub> A and B peak on DEAE as in Fig. 1 and chromatographed these individually on phosphocellulose. Both peaks eluted at the same 0.26M KCl under these conditions.

We next examined the dissociation of

Fig. 3. Sucrose gradient analysis of DDT<sub>1</sub> cytosol receptors. The DEAE and phosphocellulose peaks were concentrated on hydroxylapatite and eluted with buffer A containing 0.5M phosphate (12). Samples (200  $\mu$ l) of the concentrated receptors or cytosol receptors were centrifuged for 18 hours at 257,000g on linear sucrose gradients (5 to 20 percent) prepared in buffer A with or without 0.4M KCl (Beckman SW 65Ti rotor). Fractions of 0.3 ml (DEAE A, DEAE B, and phosphocellulose) or 0.2 ml (untreated and cytosol treated with 0.4M KCl) were diluted to 0.5 ml with H<sub>2</sub>O and counted as described (Fig. 1).



<sup>14</sup>C-labeled ovalbumin (3.6S) was used as an internal marker. (A) Hydroxylapatite-concentrated DEAE peak B migrates in the presence of 0.4M KCl at 4.7S. (B) Hydroxylapatite-concentrated DEAE peak A and the phosphocellulose peak migrate as broad peaks at 4.8S in the presence of 0.4M KCl. (C) Untreated cytosol migrates at 7S in the absence of KCl. (D) Cytosol in the presence of 0.4M KCl migrates at 4.8S.

receptor-hormone complex. Excess (1000-fold) unlabeled hormone was added to prevent detectable recombinations of [<sup>3</sup>H]DHT with receptor. Using the charcoal-dextran method of Korenman (10) we measured the half-life of the complexes under pseudo first-order conditions. The half-life was 74.5 hours at 0°C, with a rate constant of dissociation of  $1.55 \times 10^{-4} \text{ min}^{-1}$ ; at 0°C, all of the known extracellular androgen binding proteins, including the above-mentioned ABP and testosterone-estradiol binding globulin, have half-lives of dissociation in minutes (6). Using sucrose gradient centrifugation to separate bound and free steroid we measured the equilibrium constant ( $K_d$ ) to be  $2.99 \times 10^{-10} M$  (2).

In order to examine further the subunit structure of the DDT<sub>1</sub> cytoplasmic receptor, we utilized sucrose gradient analysis (2). Gradients (containing 0.4M KCl) analyzing the partially purified receptor peaks for DEAE and phosphocellulose or from cytosol treated with 0.4M KCl yielded receptor peaks sedimenting at 4.6 to 5S (Fig. 3). Similar gradients (without KCl) in similar cytosol preparations not treated with KCl revealed a peak sedimenting at 7S (Fig. 3). These data indicate that the 7S cytosol receptor in these cells has salt-dissociable 4.8S subunits that each bind androgen and are separable by DEAE chromatography. These results, in addition to our previous demonstration that this DDT<sub>1</sub> receptor enters the nucleus (2), exclude the possibility that the molecules are ABP's or other protein resembling the testosterone-estradiol binding globulin.

Our data indicate that the DDT<sub>1</sub> cytoplasmic receptor has several physical-chemical homologies with the highly purified chick oviduct progesterone receptor described by Schrader and O'Malley (4). The chick oviduct progesterone receptor has been suggested to have a subunit structure in which two 4S monomers combine to form a 6S dimer and also higher aggregates. The monomers appear to have different functions in the nucleus, the A subunit binding preferentially to DNA and the B monomer binding to chromatin. This has led to the proposal (11) that the actual cytoplasmic form of the receptor is a 6S dimer, consisting of an A and B monomer, which moves into the nucleus and is bound to the chromatin acceptor site by the B subunit. The A subunit then dissociates from the B monomer and searches along adjacent chromatin regions until it locates and binds to a specific regulatory site on naked DNA. This model would explain the presence of subunits with different

functions, but both binding steroid hormone.

We are not aware of a previous demonstration that the A and B monomers exist in only one cell type. Using DDT<sub>1</sub> cells, we have now shown that putative receptor subunits are present in a cloned cell line. This finding excludes the possibility that the multiple receptor forms from steroid target tissues are the result of mixing cytoplasm from different cell types during homogenization. These findings substantiate the hypothesis that the A and B monomers are subunits of a larger dimer existing within a single cell.

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## Blue-Green Algae: Their Excretion of Iron-Selective Chelators Enables Them to Dominate Other Algae

**Abstract.** *During blue-green algal blooms, other algae can be completely suppressed. This ability of blue-green algae to suppress other algae may be determined by the availability of iron. Iron deprivation induces the production of hydroxamate chelators, which appear to be the agent suppressing other algae.*

The availability of iron for microbial uptake appears to be an important variable in determining the stability and composition of aquatic ecosystems. Twice in a eutrophic lake the sudden dominance of blue-green algae coincided with the production of strong iron chelators and the rapid uptake of iron. The increased demand for iron appears to result from high rates of nitrogen fixation. We believe that the *Anabaena* species, the dominant alga during these periods, excreted chelators that enhanced the growth of blue-green algae or directly restricted the growth of competing species (or both).

This type of competition between aerobic microbes has been intensively studied by terrestrial ecologists and pharmacologists. In microbes, the demand for iron often induces the excretion of siderochromes, which are trihydroxamates (or catechols) of low molecular weight that can selectively chelate ferric iron. Ferric iron is chelated by acetohydroxamic acid several orders of magnitude greater than are other cations (1). Neilands (2) believes that many of these chelators act as carrier molecules transporting iron across membranes. Species with

stronger iron uptake systems would thus be able to preferentially utilize the available iron. This has been best documented in mammalian systems where the competition for iron between the microbial siderochromes and mammalian proteins, such as transferrins and lactoferrins, determines the virulence of the microbe (3). Some microbes produce antibiotics that are structural homologs of a competitor's siderochromes. These antibiotics bind iron and are inducible by low concentrations of iron (3). Because these antibiotics are actively taken up by the siderochrome uptake system and this toxicity results from intracellular reactions, the toxicity can be reduced or eliminated by the addition of another siderochrome. Thus, the ability of a species to survive a period of low iron availability may depend both on its ability to take up iron and its capacity to antagonize competing species.

Although the processes controlling the uptake of iron in aquatic ecosystems are not as well understood, there are many reports that are consistent with the results obtained in terrestrial systems. Iron often stimulates photosynthesis in many diverse aquatic ecosystems (4, 5). Also,