LETTERS

Tryptophan-Poor Diets

The introductory paragraphs of the report by Lytle et al. (14 Nov., p. 692) tentatively relate their findings of "reduced levels of brain serotonin" and "increased responsiveness to electric shock" in "[r]ats fed tryptophan-poor corn diets" to the behavioral sequels of "protein-calorie malnutrition experienced early in life," or, more specifically, to those of "extreme kwashiorkor or marasmus." Indeed, they may be so related.

What seems remarkable to this reader is the absence of any mention of the "dementia" that, every spring for many years, filled virtually every available bed in the mental institutions of the American South with adults afflicted with pellagra. This disease was most surely the consequence of ingestion of a "tryptophan-poor corn diet." All symptoms, including the psychosis, were rapidly alleviated by the administration of nicotinic acid. The entire disease syndrome is prevented by dietary measures that increase the intake of all amino acids including tryptophan, as well as that of nicotinic acid. Tryptophan is the biological percursor of both nicotinic acid and serotonin.

PHILIP HANDLER National Academy of Sciences, Washington, D.C. 20418

We welcome Handler's reminder regarding the "dementia" and pellagra observed in humans consuming diets that are poor in tryptophan, and that lack supplementary nicotinic acid, for long periods of time. However, we respectfully disagree with his suggestion that the increased electroshock sensitivity that we observed in rats consuming tryptophan-deficient, cornbased diets was caused by nicotinic acid deficiency.

1) Both of our test diets (that is, the casein control and corn diets) were supplemented with 45 milligrams of niacin per kilogram (dry weight), a concentration well within the range present in most standard lab chows. Hence, it seems unlikely that our corn-fed rats were deficient in this vitamin. Of course, only its direct measurement in plasma of corn-fed and casein-fed animals will suffice to establish the absence of a deficiency state.

2) We have shown that the diet-induced changes in both brain serotonin and electroshock sensitivity are reversed within 1 hour after the injection of tryptophan. We know of no evidence that the dementia of pellagra is reversed so rapidly after a single dose of tryptophan.

3) The injection of fluoxetine hydro-

chloride (Lilly 110140), a highly specific inhibitor of serotonin reuptake into neurons, reverses the changes in electroshock sensitivity among corn-fed rats (1). There seems little basis for believing that this drug also restores normal concentrations of nicotinic acid.

4) p-Chlorophenylalanine, a drug that decreases brain serotonin by inhibiting the enzyme tryptophan hydroxylase, exacerbates the diet-induced increases in electroshock sensitivity. We also know of no evidence that this drug decreases nicotinic acid.

5) Injections of the amino acids leucine, valine, or phenylalanine (which acutely lower brain tryptophan and suppress serotonin synthesis) in normal animals produces hyperalgesia within 1 hour after injection (2). It appears unlikely that these compounds would also produce pellagra in rats within 1 hour after injection.

6) Brain lesions that destroy serotonin axons and terminals in the telencephalon also produce hyperalgesia in rats (3). Here, too, we know of no relation between these manipulations and nicotinic acid.

In summary, although the consumption of tryptophan-deficient diets causes numerous biochemical changes besides decreasing brain serotonin, we believe that the most parsimonious explanation for our findings on pain sensitivity is that diet-induced reductions in brain serotonin underlie this effect.

> LOY D. LYTLE, RITA B. MESSING LAUREL FISHER, LEE PHEBUS

Department of Nutrition and Food Science, Massachusetts Institute of Technology, Cambridge 02139

References

- L. D. Lytle, R. B. Messing, L. Fisher, L. Phebus, *Pharmacologist* 17, 167 (1975).
 R. B. Messing, L. Fisher, L. Phebus, L. D. Lytle, in
- 3.
- R. B. Messing, L. Fisher, E. Fileous, E. D. Eyler, ... preparation.
 J. A. Harvey and C. E. Lints, *Science* 148, 250 (1965); C. E. Lints and J. A. Harvey, *J. Comp. Physiol. Psychol.* 67, 23 (1969); S. A. Lorens, J. P. Sorensen, J. A. Harvey, *ibid.* 73, 284 (1970); J. A. Harvey and C. E. Lints, *ibid.* 74, 28 (1971); L. M. Yunger and J. A. Harvey, *ibid.* 83, 173 (1973).

Good Scents?

The discovery and research of truly effective malodor counteractants (Research News, 28 Nov., p. 870) is undoubtedly welcomed by investigators concerned with olfactory systems, and probably by enterprising business, as market testing is reportedly in progress. The "other side of the coin" is not, however, mentioned in the article.

Would indiscriminant use of these chemicals as "air fresheners" deprive one of sampling some of the simpler pleasures of life (the distinctive aroma of a favorite

cheese, the bouquet of a fine beverage)? Could not untimely or abusive use of malodor counteractants interfere with the detection of undesirable substances, such as the aromatics evolved from spoiling food or the mercaptans in escaping natural gas?

For a naive or unsuspecting consumer, would "good scents" mean good sense? It is to be hoped that public merchandising of these chemicals will not open Pandora's box.

L. M. KELLEY

Department of Microbiology and Medical Technology, University of Arizona, Tucson 85721

Biohazard: Virus-Contaminated Liquid Nitrogen

We wish to alert microbiologists to a potential source of laboratory infection which may be frequently overlooked. A variety of pathogenic and nonpathogenic microbes are commonly stored in glass ampules in liquid nitrogen storage tanks. This form of storage may present risks.

As a part of our routine laboratory duties, we store vesicular stomatitis virus (VSV), as well as other viruses, in flamesealed glass ampules in a 240-liter liquid nitrogen storage tank. Recently, while removing a vial containing VSV from the storage tank, we noticed that an ampule had shattered within a storage cane. Several more broken ampules were found during a more thorough search. A review of the literature as well as several phone calls revealed that apparently no information was available on the survival of viruses under such conditions, but it was known that erythrocytes remained viable after direct contact with liquid nitrogen (1). We thus sought to determine if the contents of the broken vials had found their way into the surrounding liquid nitrogen and, if so, whether the contaminating viruses had survived.

Six 250-milliliter samples of liquid nitrogen obtained in sterile containers were taken from the storage tank in which VSV ampules had broken, placed in a hood, and evaporated to dryness. Small volumes of saline were used to rinse the interior surfaces of the sample vessels, and these rinses were examined for VSV. As many as 160 infectious virus particles were detected in one sample. By adding virus directly to liquid nitrogen, we demonstrated that there is no significant loss of virus infectivity.

The recovery of an infectious virus of minor clinical importance portends the potential biohazard of storing more virulent viruses under similar conditions. One might also be concerned about the possible creation of aerosols by the evaporating liquid nitrogen. We are unaware of the stability of other infectious agents in liquid nitrogen, but appropriate safety precautions appear to be indicated if glass vials are used for their storage. Possibly, plastic vials (which are commercially available) could be investigated as an alternative storage container.

In the past 75 years, some 3500 cases of laboratory-acquired infections have resulted in more than 150 deaths (2). The sources of such infections were frequently difficult, if not impossible, to trace. These statistics, and the recent discussions of laboratory biohazards and potential biohazards (2, 3) underscore the need for constant vigilance in the microbiology laboratory.

THOMAS W. SCHAFER, JEFFREY EVERETT GERALD H. SILVER, PAUL E. CAME

Schering Corporation, Bloomfield, New Jersey 07003

References

- H. T. Meryman and E. Kafig, Proc. Soc. Exp. Biol. Med. 90, 587 (1955).
- 2.

N. Wade, Science 182, 566 (1973). C. Norman, Nature (London) 254, 6 (1975); R. Weiss, ibid. 255, 445 (1975). 3.

"Affinity" Chromatography

The terms "affinity chromatography column" and "affinity purification" are used inaccurately and often indiscriminately in many publications (1). For the last two decades it has generally been believed that the attachment of an appropriate substrate or inhibitor to a support matrix can produce a highly specific chromatographic column which can be designed to isolate a particular protein or enzyme. Recently, however, it has become evident that in many cases the column specificity is less than adequate. There are three major causes which may interfere with the specificity.

1) The matrix offered by the major commercial manufacturers of the activated gel used for the attachment of the ligand is Sepharose 4B, and this matrix retains its molecular sieving properties and will selectively retard proteins of low molecular weight.

2) In cases where the manufacturer offers a spacer attached to the matrix, the spacer contains a charged moiety to which the ligand is to be bound, and this can act as an ion exchanger if not properly blocked. Ion exchanging properties can also result from charged groups on the ligand itself.

3) In many instances, the strong adsorption of an enzyme on an affinity column has been shown to be a result of hydrophobic interactions between the spacer at-

Although a good resolution of proteins is obtained by "affinity" chromatography, ascribing the results to a real affinity binding process is often more speculative than proven, and the actual cause of the resolution is disregarded. We suggest that, unless affinity binding is specifically demonstrated, terminology such as "activated gel column" or "modified gel column" is preferable.

Amiram D. Landman E. THACKERAY PRITCHARD Department of Oral Biology, Faculty of Dentistry, University of Manitoba, Winnipeg, Canada R3E0W3

Notes

1. See, for example, P. O'Carra, *Methods Enzymol.* 34, 108 (1974).

Cross-Cultural Health Assessment

It may interest those who read Horatio Fabrega's article "The need for an ethnomedical science" (19 Sept., p. 969) to know that an interdisciplinary group at the University of Washington is developing the Sickness Impact Profile (SIP), by which self-perceived changes in usual daily activities related to health can be measured (1). The rationale for developing the SIP closely parallels Faberga's concern: the need for a culture-free measure that would permit evaluation of treatment and assessment of need both within and across sociocultural groups. The developmental study design includes examination of the relationship between the SIP and clinician measures of disease and dysfunction, between the SIP and a subject's self-assessment of sickness and dysfunction, and between the SIP and diagnostic measures (2).

The SIP is being used in England and in Alabama, and a translation into a Chicano dialect of Spanish has just been completed. Such efforts should permit study of the cross-cultural issues Fabrega mentions.

MARILYN BERGNER

BETTY S. GILSON, RUTH A. BOBBITT DIANE K. MARTIN, WILLIAM CARTER Sickness Impact Profile Project, Department of Health Services, School of Public Health and Community Medicine, University of Washington, Seattle 98195

References and Notes

- 1. This research has been supported by the Bureau of Community Health Services and the Bureau of Health Services Research, Health Resources Administration, Department of Health, Education, and Welfare.
- and Weltare.
 B. S. Gilson, J. S. Gilson, M. Bergner, R. A. Bobbitt, S. Kressel, W. E. Pollard, M. Vesselago, *Am. J. Public Health* 65, 1304 (1975); M. Bergner, R. A. Bobbitt, W. E. Pollard, D. K. Martin, B. S. Gilson, *Med. Care* 14, 57 (1976); W. E. Pollard, R. A. Bobbitt, M. Bergner, D. K. Martin, B. S. Gilson, *ibid.* in press Gilson, ibid., in press.



Checked your spectrophotometer lately?

Do you know the exact temperature of your spectrophotometer cuvette in terms of absolute accuracy? Here's an easy way to find out.

The YSI Model 45CU cuvette thermometer is a laboratory temperature standard specifically designed to check cuvette temperature.

The instrument has four high accuracy ranges that let you measure 25°, 30°, 32° and 37°C with a readability, reproducibility and precision of $\pm .01$ °C and an absolute accuracy of $\pm .05^{\circ}C$ traceable to NBS.

There is also a survey range of 20° to 40°C providing continuous coverage of the four high accuracy ranges.

This portable, battery powered temperature standard has many additional clinical laboratory applications, including calibration and accuracy verification of temperature control systems, laboratory thermometers and other temperature related equipment.

If knowledge of true temperature is critical to your measurements, check with the YSI Model 45CU cuvette thermometer.

SCIENTIFIC DIVISION YELLOW SPRINGS INSTRUMENT CO., INC.