gration of goals, some federal funding, administered possibly through the National Academy of Sciences, will be required to provide lists of cooperating industries and to circulate grant proposals to relevant industries. Also, the possibility of cost-sharing (say, onethird federal funding) should be considered.

DONALD F. PARSONS Roswell Park Memorial Institute, Buffalo, New York 14203

La Porte Precipitation Fallacy

I was pleased to note that Landsberg in his admirable survey "Man-made climatic changes" (18 Dec, p. 1265) recognized the controversial nature of the La Porte precipitation anomaly which too many meteorologists, without close scrutiny of the record, have accepted as a proven case of man-made climatic change. The La Porte rainfall record is a celebrated but specious example of man's inadvertent modification of climate.

During the years 1928-1963 the precipitation for La Porte, Indiana, located some 30 miles east of the Chicago industrial complex, was higher than that recorded at surrounding stations. Meteorologists have been puzzled by this precipitation anomaly. Changnon suggested that the increased precipitation might be due to inadvertent man-made cloud modification resulting from nucleation debris ejected by the industrial activities to the west (1). Air pollution in any form is bad, but to blame the Chicago industrial complex for the change of climate at La Porte is most unfair.

The precipitation record at La Porte can be shown (2) to be spurious, statistically invalid, and physically unacceptable. (A most ludicrous outgrowth of the belief that the La Porte anomaly is true was the award of a government research contract to a fine university to study the presumed ecological changes in the environs of La Porte because of fictitious excesses of rainfall.)

The rainfall anomaly began in 1927 when a new cooperative climatic observer was appointed and ended in 1964 when an automatic rain gauge installed at La Porte replaced him. Prior to 1927 many of the surrounding stations reported more annual rainfall than La Porte and after 1964 many of the nearby stations also reported more rain than La Porte. However, in the intervening 37 years except for one instance, La Porte always reported the highest rainfall amounts. . .

A study was also made of many synoptic weather situations, especially in those years which had unusually high rainfall amounts. Trajectory analyses demonstrated that when a west-northwest flow brings effluent nucleation debris to La Porte, either of two events occurs: (i) if it is raining at La Porte, the rain stops, and (ii), if it is cloudy at La Porte, the weather clears. These easily explainable occurrences are hardly conducive to causing excess precipitation.

B. G. HOLZMAN 11305 Maryvale Road, Upper Marlboro, Maryland 20870

References

 S. A. Changnon, Jr., Bull. Amer. Meteorol. Soc. 59, 4 (1968); *ibid.*, 50, 411 (1969).
B. G. Holzman and H. C. S. Thom, *ibid.* 51, and the second seco 335 (1970).

Last Word on Yogurt-Making

Being an Armenian by ethnic origin and also a microbiologist, I should perhaps be in a more advantageous position to comment on the making of yogurt than the previous letter writers (Segal, 31 July; Goodman, 9 Oct.; and Bagdikian, 6 Nov.).

Yogurt (yoghurt in Turkish; mādzun in Armenian; leben in Arabic) is curdled or coagulated milk resulting from the fermentation of sugars in the milk by two microorganisms, Lactobacillus bulgaris and Streptococcus thermophilus (1). Both these microorganisms are capable of multiplying at temperatures between 20° and 50°C (2) and, with acid formation, ferment a number of sugars including lactose which is the main sugar present in milk. As a result of acid formation the proteins in the milk are curdled or coagulated and the milk attains the yogurt consistency.

Any kind of milk can be used to make yogurt but whole fresh cow's milk often gives the best results. The optimum temperature for the growth of yogurt-forming microorganisms lies within the range of 40° to 50°C. At this temperature range growth is rapid and yogurt formation takes only a few hours. If the temperature falls below 40°C the growth continues but at a slower rate and consequently yogurt formation delays. Multiplication of







Nalgene Beakers. Unbreakable, non-wetting, and dripless... all the way down.

Why buy replacement beakers that will just break again? Specify the permanent replacements-unbreakable Nalgene . . . the original dripless beakers. Available in a wide range of sizes and materials.

TPX: Transparent. Excellent heat and chemical resistance. 8 sizes, 30-1000 ml. (Cat. #1203)

Teflon* FEP: Indestructible, autoclavable by any method, unaffected by virtually any chemical. 8 sizes, 30-1000 ml. (Cat. #1500)

Polypropylene: Translucent, economical. 10 sizes, 30-4000 ml. (Cat. #1201) Order from your Laboratory Supply Dealer. Ask for our catalog or write Dept. 4103, Nalgene Labware Division, Rochester, N. Y. 14602. *DuPont Registered Trademark

Nalgene[®]Labware ... b etter all the time

