## Plasma 3S $\gamma_1$ -Globulin: Identity with Erythrocyte Carbonic Anhydrase B

Abstract. The 3S  $\gamma_1$ , 2S  $\gamma_2$ , and 0.5S  $\gamma_{2}$  fractions of human plasma are heterogeneous in protein composition. Although each fraction contained a relatively small amount of protein antigenically related to the immunoglobulin light chains, most of the proteins were unrelated to immunoglobulin G or its light chains. Of the 3S  $\gamma_1$ -globulins the greater part was immunochemically identical to carbonic anhydrase B and had carbonic anhydrase activity. These findings explain earlier reports of an immunochemical similarity between 3S  $\gamma_1$ -globulins and immunoglobulin light chains in spite of marked differences in amino acid and peptide composition between the two. Apparently not all plasma  $\gamma$ -globulins are necessarily immunoglobulins.

There are at least three different classes of  $\gamma$ -globulins in normal human plasma which have molecular weights of 25,000 or less; these have been designated 3S  $\gamma_1$ , 2S  $\gamma_2$ , and 0.5S  $\gamma_2$ , respectively, on the basis of their electrophoretic mobilities at pH 8.6 and their sedimentation rates (1). Although the 3S  $\gamma_1$  class may contain proteins with antigenic determinants similar to those of the immunoglobulin light chains (2), the 3S  $\gamma_1$ -globulins and the immunoglobulin light chains are quite different from each other in peptide and amino acid composition (1). The nature of the 3S  $\gamma_1$ -globulins was explored further in our study.

The three classes of low-molecularweight  $\gamma$ -globulins were isolated from pooled human plasma by a combination of ethanol fractionation, DEAEcellulose (diethylaminoethyl) chromatography, and Sephadex G-100 filtration; the physicochemical properties of these preparations have been described (1). Human immunoglobulin G ( $\gamma$ G) was isolated from pooled plasma by ethanol fractionation (3) and was used for the preparation of  $\gamma$ G light chains

Table 1. Percentage of  $I^{131}$ -labeled protein precipitated by specific antiserum.

Labeled protein	Antiserum to	
	γG	0.3S y1
γG	98.6	< 0.05
$\gamma G$ light chains	99.2	<0.05
3S Y1	3.0	<b>90.5</b>
$2S \gamma_2$	8.7	19.0
$0.5S^{\prime}\gamma_2$	0.9	15.3

(4). Antiserums against  $\gamma G$  and against 3S  $\gamma_1$  were prepared in rabbits. Portions of  $\gamma G$ ,  $\gamma G$  light chains, and each of the low-molecular-weight  $\gamma$ -globulins were labeled with I<sup>131</sup> by a nitrous acid method (5).

Mixtures of a given I131-labeled protein, the same protein unlabeled as a carrier, and unlabeled  $\gamma G$  were precipitated with antiserum to  $\gamma G$  and antiserum to 3S  $\gamma_1$  in the zone of antibody excess (37°C for 1 hour and overnight at 4°C); the precipitates were separated by centrifugation, and washed twice with 0.15M NaCl. Precipitates and supernatants were assayed for radioactivity with a well-type crystal (Table 1). Antiserum to  $\gamma G$  precipitated more than 98 percent of the radioactivity of the labeled  $\gamma G$  and  $\gamma G$  light-chain preparations but only 3 percent of the labeled  $\gamma_1$ -globulin; several different 3S  $\gamma_1$ -globulin preparations gave similar results. On the other hand, antiserum to 3S  $\gamma_1$  precipitated 90.5 percent of the  $3S_{\gamma_1}$ -globulin but less than 0.05 percent of the labeled  $\gamma G$  or  $\gamma G$  light chains. Thus, the 3S  $\gamma_1$  preparation contained several different proteins: (i) A protein which represented a small fraction of the total 3S  $\gamma_1$ -globulins had antigenic determinants which reacted with antiserum to  $\gamma G$ ; this reaction was not due simply to contamination of the 3S  $\gamma_1$ preparation with  $\gamma G$ , since it has been shown that the reacting protein lacks  $\gamma G$  heavy-chain determinants (2). (ii) A protein which constituted the bulk of the 3S  $\gamma_1$  preparation was precipitable with antiserum to 3S  $\gamma_1$ , and was antigenically different from  $\gamma G$ ; in support of this, the 3S  $\gamma_1$  preparation gave a single precipitation line with antiserum to 3S  $\gamma_1$  in Ouchterlony plates (6), and  $\gamma G$  gave none (Fig. 1a). (iii) From Table 1, some of the 3S  $\gamma_1$ -globulins and most of the 2S  $\gamma_2$ - and 0.5S  $\gamma_1$ -globulins did not precipitate with either antiserum to  $\gamma G$  or antiserum to 3S  $\gamma_1$ .

The concentration of the globulin precipitable by antiserum to  $3S_{\gamma_1}$  in 74 normal adult serums, as determined by radial immunodiffusion (7), ranged from less than 0.014 mg per 100 ml of serum, the lower limit of detection, to approximately 0.56 mg. However, serums with detectable hemolysis gave extraordinarily high values, and the concentration of this protein in erythrocytes averaged 300 mg per 100 ml of blood in 13 adults, 23 mg in 26 newborns, and less than 3 mg in three fetuses of 3 to 5 months' gestation. Since the change in erythrocyte concentration of the pro-



Fig. 1. (a) Ouchterlony plate; antiserum to 3S  $\gamma_1$  in central well. S indicates 0.1M NaCl; G,  $\gamma$ G; 3, 3S  $\gamma_1$ ; 2, 2S  $\gamma_2$ ; and 0.5, 0.5S  $\gamma_2$ . (b) Ouchterlony plate; antiserum to 3S  $\gamma_1$  in central well. S indicates 0.1M NaCl; C, erythrocyte catalase; B, carbonic anhydrase B; 3, 3S  $\gamma_1$ . (c) Immunoelectrophoresis of carbonic anhydrase B (b) and 3S  $\gamma_1$  (3); antiserum to 3S  $\gamma_1$  diffused into agar from above and below the figure. Anode was to the right.

tein with age resembled the pattern of development of erythrocyte carbonic anhydrase, the low-molecular-weight  $\gamma$ -globulins were tested for carbonic anhydrase activity by an electrometric method (8) (Table 2). The carbonic anhydrase activity of the 3S  $\gamma_1$ -globulins was approximately 44 percent of that exhibited by erythrocyte lysates containing equivalent concentrations of protein precipitable with antiserum to  $3S \gamma_1$ ; this is compatible with the fact that erythrocytes have several carbonic anhydrases (9). The carbonic anhydrase activities of the 2S  $\gamma_2$  and 0.5S  $\gamma_2$ preparations were 22 and 3.7 percent, respectively, of that for the 3S  $\gamma_1$  preparation; 19 and 15 percent of the total proteins in 2S  $\gamma_2$  and 0.5S  $\gamma_2$  preparations, respectively, were precipitable with antiserum to 3S  $\gamma_1$ . Hence the 0.5S  $\gamma_2$ -globulin precipitable with antiserum to 3S  $\gamma_1$  must have had little carbonic anhydrase activity, and may represent degradation products of the antigenically related 3S  $\gamma_1$ -globulin. The  $\gamma G$  preparations had no detectable carbonic anhydrase activity.

In Ouchterlony plates, the precipitation line formed between antiserum to

Table 2. Carbonic anhydrase activities per microgram of protein.

Preparation	Enzyme units (mean $\pm$ S.D.)
$\begin{array}{c} 3S  \gamma_1 \\ 2S  \gamma_2 \\ 0.5S  \gamma_2 \end{array}$	$\begin{array}{c} 1.87 \pm 0.21 \\ 0.414 \pm 0.067 \\ 0.069 \pm 0.010 \\ 4.055 \pm 0.055 \end{array}$
erythrocytes	4.23 ± 0.03+

\* Per microgram of protein precipitable with antiserum to  $3S \gamma_1$ . 3S  $\gamma_1$  and the 3S  $\gamma_1$ -globulins fused completely with that formed between antiserum and purified erythrocyte carbonic anhydrase B (Fig. 1b). On immunoelectrophoresis (10) with antiserum to 3S  $\gamma_1$ , the mobility of the reacting 3S  $\gamma_1$ -globulin was identical to that for carbonic anhydrase B (Fig. 1c).

Thus, although the plasma 3S  $\gamma_1$ globulins included a small amount of a protein antigenically related to the immunoglobulins as reported earlier (2), most of the  $\gamma$ -globulin in this class of proteins was not related to  $\gamma G$  or to  $\gamma G$  light chains but was identical or very similar to carbonic anhydrase B, thus explaining the difference in peptide composition noted (2) between  $3S \gamma_1$ plasma fractions and the immunoglobulins as well as the similarity between an amino-terminal tetrapeptide obtained from the 3S  $\gamma_1$ -globulins and that of carbonic anhydrase B (11). In accord with this, the amino acid composition of whole 3S  $\gamma_1$ -globulin preparations is quite unlike that of the immunoglobulin light chains (1), but is identical, within the limits of error of the methods, with that reported for carbonic anhydrase B by others (9).

DAVID GITLIN, MARY BOESMAN KARL SCHMID, PEKKA VUOPIO

University of Pittsburgh School of Medicine, Pittsburgh, Pennsylvania, and Boston University School of Medicine, Boston, Massachusetts

## **References and Notes**

- 1. T. Ikenaka, D. Gitlin, K. Schmid, J. Biol. Chem. 240, 2868 (1965); T. Iwasaki and K. Schmid, *ibid.* 242, 2356 (1967); R. B. Nim-K. Schmid, Experientia Larsen, berg, L. Larse: 23, 520 (1967).
- 23, 520 (1967).
  R. C. Williams, Jr., and K. Schmid, J. Immunol. 99, 406 (1967); P. Fireman, E. Hershgold, F. Cordoba, K. Schmid, D. Gitlin, Nature 203, 78 (1964).
  J. L. Oncley, M. Melin, D. A. Richert, J. W. Cameron, P. M. Gross, Jr., J. Amer. Chem. Soc. 71, 541 (1949).
- 4. S. Cohen, Biochem. J. 89, 334 (1963); J. B. Fleischman, R. H. Pain, R. R. Porter, Arch. Biochem. Biophys. Suppl. 1, 174 (1962).
- 5. D. Pressman and H. N. Eisen, J. Immunol. 64, 273 (1950); D. Gitlin, J. Kumate, J. Urrusti, C. Morales, J. Clin. Invest. 43, 1938 (1964).
- Ochterlony, Acta Pathol. Microbiol.
  Scand. 32, 231 (1953); A. J. Crowle, J. Lab.
  Clin. Med. 52, 784 (1958). 6. Ö.
- 7. G. Mancini, A. O. Carbonara, J. F. mans, Immunochemistry 2, 235 (1965). F. Here-
- K. M. Wilbur and N. G. Anderson, J. Biol. Chem. 176, 147 (1948); H. Mattenheimer and H. de Bruin, Anal. Biochem. 4, 222 (1962).
- 9. J. M. Armstrong, D. V. Myers, J. A. Verpoorte, J. T. Edsall, J. Biol. Chem. 241, 5137 (1966)
- 10. J. J. Scheidegger, Int. Arch. Allergy 7, 103 (1955).
- 11. T. Ikenaka and K. Schmid, Nature 215, 66 (1967).
- Supported by grants HD-00652 and GM-10374 from NIH. We thank Dr. P. O. Nyman of the University of Göteberg for the purified ervthrocyte carbonic anhydrase B.
- 7 February 1968

3 MAY 1968

## Dihydroorotic Acid Dehydrogenase Activity of Human

## **Diploid Cell Strains**

Abstract. A gene affecting the final two enzymes of uridylic acid biosynthesis does not affect a third, metabolically adjacent enzyme. Similarly, compounds that increase cellular activity for the affected enzymes do not increase activity for the third enzyme. The pyrimidine pathway can be subdivided into groups of concurrently responding enzymes. These groups may be smaller in human cells than they are in microbial cells.

Dihydroorotic acid dehydrogenase (DAD, E.C. 1.3.3.1), one of the enzymes in the uridine biosynthetic pathway, catalyzes the reversible conversion of dihydroorotic acid to orotic acid (Fig. 1). This enzyme is present in circulating human leukocytes but absent from red blood cells (1). Both erythrocytes (2-4) and cultured diploid cell strains (5, 6) from patients with orotic aciduria, a rare autosomal recessive disorder, are deficient in two sequentially acting enzymes in the biosynthetic pathway (Fig. 1) leading to uridine-5'monophosphate (UMP). These enzymes are orotidine-5'-monophosphate (OMP) pyrophosphorylase (E.C. 2.-4.2.10) and OMP decarboxylase (E.C. 4.1.1.23); diploid cell strains from mutant homozygous donors show 1 percent of normal activity for both enzymes. When mutant homozygous cultures are grown in a medium containing either 5-azaorotic acid (a competitive inhibitor of OMP pyrophosphorylase) or 6-azauridine (whose ribotide is a competitive inhibitor of OMP decarboxylase), the cells develop nearly normal activity for both enzymes (7, 8). Barbituric acid, which is not an inhibitor of either enzyme, also causes mutant homozygous cells to develop nearly normal activity (8). The response of normal and heterozygous cells to these agents is similar to that of the mutant homozygous strains, but in both enzymes the proportionate increase is smaller. The other two enzymes peculiar to this pathway (Fig. 1), aspartate transcarbamylase (E.C. 2.1.3.2) and dihydroorotase (E.C. 3.5.2.3), which catalyze consecutive reactions immediately preceding the one which DAD catalyzes, are not deficient in mutant cells (2, 3, 6). Moreover, cellular dihydroorotase activity does not rise in response to at least one of the agents which increases OMP pyrophosphorylase and OMP decarboxylase activity (8). The effect of the gene for orotic aciduria, or of the compounds mentioned above, on the activity of DAD has not been reported. We have therefore sought answers to the following questions. (i) Does the gene for orotic aciduria affect DAD activity? (ii) Do human cells increase their DAD activity in response to the compounds that



Fig. 1. The catalytic sequence used by human diploid cell strains to synthesize uridine-5'-monophosphate.