the triple point of water as 0.0098°C, although the Ninth General Conference on Weights and Measures (1948) adopted [Science 109, 100 (1949)] and the Tenth Conference (1954) reaffirmed [Science 120, 1008 (1954)] the triple point of water as 0.0100° Celsius (centigrade), thus displacing, as a fundamental fixed point, the inexact and ill-defined freezing point of water under atmospheric conditions.

Brey has fine gifts for direct, clear, and interesting exposition. Except in those chapters where the condensation has caused a sense of breathless haste, the treatment moves forward smoothly, faultless language serving to establish the degree of comprehension that the writer has deemed reasonable. The purposes of the book are good, and minor technical imperfections of fact or treatment do not seriously detract or keep it from presenting physical chemistry in a way that will specifically suit the early student of biological science.

CHARLES PROFFER SAYLOR Chemistry Division,

National Bureau of Standards

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(Inquiries concerning these publications should be addressed, not to Science, but to the publisher or agency sponsoring the publication.)

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