

thymus that difficulty in differentiating the two tissues does not arise; often one can grossly distinguish the surface lymphoid follicles in the nodes.

This work has been supported by a research grant (C-2344) from the National Institutes of Health, Department of Health, Education, and Welfare.

VAUGHAN P. SIMMONS

*Department of Pharmacology, Marquette University
School of Medicine, Milwaukee, Wisconsin*

29 July 1954.

Plastic Chamber for Inert Atmospheric Work

The determination of physical constants of materials that are sensitive to the atmosphere requires expensive and elaborate equipment. We have designed an inexpensive dry-box that can be used to determine the refractive indices of materials that are oxidized by atmospheric oxygen. To make these determinations, an Abbé refractometer must be enclosed in an atmosphere-tight container having the visibility necessary for efficient operation of the instrument. Commercially available dry-boxes are unsatisfactory for this specialized purpose because of size, cost, visibility, and weight. Polystyrene was used to construct the dry-box shown in Fig. 1 because it would provide the desired properties of the container and retain ease of construction.

A cement made of polystyrene dissolved in trichloroethylene was used to join the component parts, $\frac{1}{2}$ in. polystyrene base, 12.5×18 in., $\frac{1}{4}$ in., polystyrene sides, 9×12 in. and 9×18 in., and 0.01 in. polystyrene top. The water, electric, and nitrogen inlets were sealed with the same cement. The glove ports were made by bending with heat a 2×17.5 in. strip of $\frac{1}{4}$ -in. polystyrene around a pipe. To make bends in the thin polystyrene to form the top, the seams were moistened with trichloroethylene to soften the plastic. Any leaks that may be detected are readily sealed by application of the cement.

The use of this box was found to be quite satisfactory. The refractometer scale could be read with ease through the top, and the box was sufficiently light to



Fig. 1. An apparatus for determining refractive indices in an oxygen-free atmosphere.

be moved even with the refractometer inside. Some leakage occurred at the glove ports, but this was not critical, if the pressure inside the box was kept greater than atmospheric pressure. A sheet of aluminum foil on the bottom of the box prevented spilled organic liquids from softening the plastic. For the removal of traces of oxygen remaining after sweeping the dry-box with an inert gas, a weighing bottle containing a glass-wool wick saturated with tri-*n*-butylborane, a substance that is readily oxidized, was opened. Materials were introduced into and removed from the dry-box through the glove ports.

With slight adaptations the box could be used for containing other pieces of apparatus or for work in an anhydrous atmosphere.

I wish to express appreciation for the technical assistance of Helmut Haendler and Walter Eldredge and for advice received from the Plax Corporation of Hartford, Connecticut.

ROBERT E. LYLE

*Department of Chemistry,
University of New Hampshire, Durham*

24 September 1954.

On Column Chromatography of Sugars

The use of carbon (1) and cellulose (2) columns has become widespread for the separation of carbohydrates of comparatively low molecular weight. It is therefore desirable to point out several factors of importance in the general use of such columns.

When using carbon columns made of any of a variety of charcoals, we have found it expedient to give the column a preliminary wash with dilute hydrochloric acid solution in order to assure the removal of basic ash, which might otherwise cause some isomerization of the sugars applied later. A 1 percent hydrochloric acid solution is sufficient. The acid is then removed from the column by washing with distilled water. Celite (3) is usually mixed with finely ground charcoals (4) to increase flow rate. However, celite sometimes dissolves in the developing solutions and is obtained as a flocculent precipitate in the concentrated effluence. Celite can be removed from the concentrated effluence by filtration through a bacterial filter or by evaporation of the solution to dryness and redissolution of the carbohydrate in water. To avoid this inconvenience we often use columns composed entirely of charcoal. The charcoal selected is that which passes a 40- or 60-mesh screen but is retained on an 80-mesh screen. This produces a column composed entirely of charcoal and consequently increases sorptive capacity of the column.

Often in the use of cellulose columns, carbohydrates other than those placed on the columns are observed in the eluates. These extraneous carbohydrates arise from the cellulose or disintegrated filter paper employed to pack the column. The cellulose used is, of course, not chemically pure but represents a purified pulp that still contains a small amount of hemicellu-