base equilibria, buffers; amphoteric substances. A brief introduction to laboratory techniques follows the theoretical section.

The qualitative scheme covers only a limited selection of common metallic ions and anions. The group separations for the metallic elements are familiar ones, essentially along lines of the Noyes and Bray scheme. The anion scheme utilizes group tests, for example, for oxidizing or reducing properties, and for precipitation by barium, silver, or calcium ions, to narrow the field of specific tests for the limited selection of anions that is provided for in the scheme.

The appendix includes lists of apparatus and reagents, density-molarity tables for the common acids and ammonia, preparation of test solutions, and mathematical operations and problems thereon. Further tables are for ionization constants of weak acids and bases, solubility products, dissociation constants of complexes, and a very extensive table (32 pages) of properties of substances that may be formed by combinations of the various anions and cations that are provided for in the qualitative schemes.

The subject index is rather brief, but probably adequate. Tables of four-place logarmithms and a set of answers to problems follow the index. A table of 1953 atomic weights is inside the front cover, and a periodic chart, including transuranic elements, is inside the back cover. The typography and the figures are excellent.

N. H. FURMAN Department of Chemistry, Princeton University

The Properties of Glass. George W. Morey. ACS Monograph Series No. 124. William A. Hamor, Ed. Reinhold, New York, ed. 2, 1954. 591 pp. Illus. \$16.50.

Morey's definitive book on glass properties follows closely the format and organization of the 1938 edition and it remains the best available book on the subject. During the past 15 years this book has achieved first rank in providing reliable information to glass technologists and scientists interested in glass. The emphasis on reliability of property data in relation to chemical composition, especially Morey's own pioneering and continuing work on phase equilibriums, has contributed greatly to the development of systematic research in glass. While some of its shortcomings may be minor, they nevertheless deserve to be considered here.

The first three chapters cover the chronological development and characteristics of glass, crystallization studies of glass systems and rates of crystal growth, and the requirements of commercial glasses and the development of new compositions. The presentation would have been improved if glass history and statistics had been brought up to date and a more complete coverage of devitrification rate studies had been ineluded.

Chapter 4 covers the chemical resistance characteristics important in glass usage. It is believed that the four new references do not adequately cover the chemical durability studies made since 1938.

The next 16 chapters are devoted to specific glass properties important in glass fabrication and end usage. Some 225 references are made to new material in these chapters, and many new data have been added. Discussions on new material appear to have been added in a manner that least disturbs the original format. This has resulted in overcondensation and, in a few instances, near exclusion of some accounts of new property measurements.

The last chapter discusses the constitution and structure of glass, principally on the basis of x-ray diffraction studies. Many investigators will not agree with the author's implication that x-ray diffraction studies, notably the excellent work of Warren and his coworkers, give a satisfactory picture of glass structure. In fact, in the author's reference No. 57 to this work, Warren states (p. 258):

... the X-ray diffraction study of a glass gives information only on average quantities; it tells nothing about the fine details of the structure.... The X-ray studies of glass might be said to establish the first order approximation to a picture of the structure, and the fine details must be filled in with other kinds of measurements.

Although future research may show that this average picture is the best that can be achieved, I would point out that it is too indefinite to be of much use except in the most simple problems of glass technology. A complete account should have included the continuing efforts of many investigators to apply Raman and infrared spectra, heat capacity, neutron and electron diffraction, electron microscope, and other types of data toward a more definite and usable picture of glass structure.

C. L. BABCOCK Glass Technology Section, General Research Division Owens-Illinois Glass Company, Toledo, Ohio

The Steel Skeleton. vol. I. Elastic Behaviour and Design. J. F. Baker. Cambridge Univ. Press, New York, 1954. xi + 206 pp. Illus. + plates. \$8.50.

In 1929 the British steel industry and the Department of Scientific and Industrial Research helped form the Steel Structures Research Committee and embarked upon an intensive investigation of various design procedures applicable to steel building frames. It was believed by many engineers that the ductile properties of steel were not fully exploited for structural purposes and that existing building codes were irrational and too restrictive. It was hoped that certain advantages would exist in a design procedure based on the theory of continuous frames and on the ductile properties of steel.

Volume I of *The Steel Skeleton*, by J. F. Baker of Cambridge University, is a review of the analytic and experimental investigations that were conducted by the Steel Structures Research Committee on the elastic behavior of steel building frames. Volume II