As already reported by Seggel, fluorescence of blood cells is an evanescent phenomenon, and its duration is related to the original brightness. In the case of the longest period of fluorescence of a fluorocyte observed by us, it was found that 120 seconds elapsed before the fluorocyte was too faint to be seen. In contrast, about the shortest time involved in a low-order fluorescence was only 4 seconds. While these data are considerably more extreme or of a wider time range than those reported by Seggel in the case of a rabbit to which phenylhydrazine had been administered, it cannot be concluded that, in our experiments, more coproporphyrin was present in the fluorocytes. It is probable that the conditions of observation were different in two particulars from those under which Seggel worked: (1) intensity of illumination and (2) level of dark adaptation.

Samples of blood from a patient having pernicious anemia were examined under optimal conditions and with the benefit of previous experiences in this type of investigation. Although fluorescence was observed, it was of a yellowish-green hue similar to that of willemite and not the red fluorescence characteristic of coproporphyrin. Although the significance of this yellowish-green fluorescence is not known to us, the presence of a brightly fluorescing yellowish-green pigment in a reticulocyte undoubtedly indicates the presence of some other modification of the normal hemoglobin of red blood cells.

The number of fluorocytes per unit volume of blood cannot be correlated with the amount of fluorescing pigment in the blood. This statement is deduced from the probability that the

total time of visibility for fluorescence per unit area under constant intensity of illumination is roughly propertional to the fluorescing pigment present. This being the case, if the population is one of fluorocytes of uniform size, the total statistical time of fluorescence should be proportional to the amount of pigment per unit volume. These considerations would explain the lack of correlation between the fluorocyte count and other clinical findings in the evidence submitted by Seggel and others and would offer a basis for the criticism of Seggel's work by Chytrek. These considerations are especially pertinent, since in no case is it possible to determine the correct total time of fluorescence. The number of reticulocytes carrying the reddish fluorescent pigment must be in excess of those observed, since some cells will contain so little of this pigment that fluorescence will be below the level of visibility or will last for too short a period to permit recognition. In general, however, the number of fluorocytes which can be observed may be taken to indicate the severity or the state of the disease.

## References

- 1. HAMLY, D. H. Amer. Ann. Photog., 1943. Pp. 58-70.
- 2. HAMLY, D. H., and SHEARD, CHARLES. J. opt. Soc. Amer., 1947, 37, 316.
- KELLER, C. J., and SEGGEL, K. A. Folia haemat., 1934, 52, 241; SEGGEL, K. A. Folia haemat., 1934, 52, 250.
- SEGGEL, K. A. Ergebn. inn. Med. Kinderh., 1940, 58, 582; UNGRICHT, MAGDALENA. Folia haemat., 1938, 60, 145; CHYTREK, E. Klin. Wschr., 1940, 19, 1321; WATSON, C. J., GRINSTEIN, MOISES, and HAWKINSON, VIOLET. J. clin. Invest., 1944, 23, 69.

## IN THE LABORATORY

## Improved and Improvised Sterilizing Containers for Bacteriological Petri Dishes and Pipettes

## IVAN C. HALL Dublin, Georgia

In spite of the close of the war it is still difficult to obtain such simple and relatively inexpensive items as sterilizing containers for bacteriological Petri dishes and pipettes, and some of those available have a common defect which this writer has observed for several years and now wishes to record so that manufacturers may correct it if they wish to do so.

For example, one type of round Petri dish container, 10 inches tall, has a lid  $2\frac{1}{2}$  inches high which slides over  $1\frac{1}{2}$  inches of the container. The lid fits so closely that its removal is often extremely difficult. A lid  $\frac{1}{2}$  inch high would be just as effective and would save much effort in handling. There is on the market an aluminum container, without a rack, with a shallow lid for four dishes, which is quite ideal. It is cheap, durable, and altogether satisfactory.

Some commercial containers for pipettes are open to the same kind of criticism as that just mentioned for Petri dishes. One sample has a cover 6 inches long, of which 2 inches fit over the container. Both container and cover are finely made of copper, are seamless, and are almost as closely fitted as a piston in a cylinder. It generally requires considerable effort to close or open one of these, and the upper third of the pipettes is exposed to aerial contamination when the lid is off, if they do not actually fall out.

Another container, made of stainless steel, not seamless, is 17 inches long, with a lid  $1\frac{1}{2}$  inches long and a sliding area of 1 inch. This is much easier to handle and exposes only  $\frac{1}{2}$ inch of the pipettes to the air. A loosely fitting lid  $\frac{1}{2}$  inch long with a short screw would be even better.

The writer has often used as containers for pipettes the discarded cartons in which adhesive tape or surgical plasters have been purchased. Some of these have shallow screw tops, and they will stand repeated sterilization in the dry air sterilizer very well, but not in the autoclave.

They are particularly useful for sterilizing Pasteur pipettes, but are too short for standard commercial pipettes. This difficulty has been met by lengthening them with an additional section fastened with Scotch tape, which also withstands repeated dry air sterilization remarkably well.

Coffee cans make very satisfactory substitutes for the commercial containers for Petri dishes, but they should first be boiled in a strong solution of lye to remove the lacquer.