# SCIENCE

Vol. 98	FRIDAY, SEPTE	EMBER 10, 1943	No. 2541
Discovery in Eastern Washington of a the Pleistocene Continental Glacier WILLIAM H. HOBBS	: Professor	FESSOR E. G. D. MURRAY. In Analysis: Dr. D. P. SMITH	
Branches of the Academy of Sciences of	of the USSR.	Special Articles:	no of Hoonemidia on
II: Dr. P. Kolesnikov		The Influence of Methyl Chalcon the Toxicity of Mapharsen in I	Rabbits: Dr. David
Obituary:  Karl Landsteiner: Dr. Michael F. C. Stuart Gager: Dr. William J. Robi and Memorials	BINS. Deaths	H. GOLDSTEIN, DR. ABRAHAM ARTHUR E. GOLDFARB. The A of Enzymatic Xanthine Oxidati MANN and CORA R. OWEN	Intibacterial Effect ion: Dr. Fritz Lip-
Scientific Events:	*	Scientific Apparatus and Laborato	ory Methods:
Scientific Research in Great Britan nadian Aircraft Standards Technical	l Committee;	A Simple Three-Color Mixer Use Dr. Walter F. Grether	ing Filtered Colors: 248
The Lewis Cass Ledyard, Jr., Fello Society of the New York Hospital Research in Applied Psychology; I plex' Award of Mead Johnson and C Nutrition Foundation; The Fiftieth of Field Museum of Natural History	; Grants for The ''B-Com- ompany; The Anniversary	SCIENCE: A Weekly Journal ment of Science, edited by J. M	devoted to the Advance
Scientific Notes and News	238	CATTELL, assistant editor. Pub	
Discussion:		THE SCIENCE	PRESS
Scientific Research by the Departmen ture: T. Swann Harding. A Tea P.	repared from	Lancaster, Penns	<b>y</b> lvania
Needles of Pine Trees Against Scurv Schick. Vitaman C in Evergreen-T Dr. Charles Macnamara	ree Needles:	Annual Subscription, \$6.00	Single Copies, 15 Cts
Scientific Books: Psychology for the Fighting Man: CARMICHAEL. Fundamentals of Immu	Dr. Leonard	SCIENCE is the official organ of tion for the Advancement of Sciening membership in the Association the office of the permanent secre Institution Building, Washington, D	ice. Information regard n may be secured from tary in the Smithsonian

# DISCOVERY IN EASTERN WASHINGTON OF A NEW LOBE OF THE PLEISTOCENE CONTINENTAL GLACIER

By Professor WILLIAM H. HOBBS

UNIVERSITY OF MICHIGAN

Studies made during the past field season<sup>1</sup> in eastern Washington have disclosed the presence there in Late Pleistocene time of a hitherto unsuspected lobe of the Cordilleran continental glacier. This lobe covered an area of nearly four thousand square miles and is to be known as the Scablands Glacier Lobe. It blocked the Spokane River of that time to impound the waters in Glacial Lake Spokane. This lake included the basins of Lakes Coeur d'Alene and Pend Oreille and extended across Idaho into Montana (see

<sup>1</sup> With the aid of grants from the American Philosophical Society and the Geological Society of America. This preliminary notice is printed by permission.

Map I). Its area was thirty-one hundred square miles. When the glacier had evacuated the Scablands, Lake Spokane expanded a distance of one hundred miles down the canyons of the Spokane and Columbia rivers to the site of Coulee Dam as Glacial Lake Leverett.

The lava plain which lies to the southwest of Spokane has a surface development almost unique, and has long been known to its inhabitants as the "scablands." To geologists of the late decades it is known as the "channeled scablands" from the apt description by Professor J Harlen Bretz, who has dego Of Dr. Bretz's many papers treating of the area the

gone conclusion that hydrogen peroxide is inhibitory to certain bacteria, mostly anaerobic and facultative anaerobic types devoid of catalase. Many of the pathogenic cocci are of this type. There may, in certain surgical cases, be merits to the slow but steady generation of hydrogen peroxide as brought about by an enzyme system. The difficulties of preparation,

however, would outweigh any, probably rather limited usefulness.

FRITZ LIPMANN
CORA R. OWEN

BIOCHEMICAL RESEARCH LABORATORY,
AND SURGICAL RESEARCH LABORATORIES,
HARVARD MEDICAL SCHOOL, MASSACHUSETTS
GENERAL HOSPITAL, BOSTON

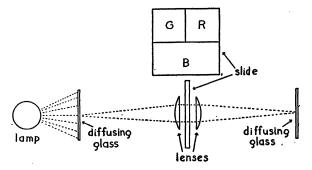
### SCIENTIFIC APPARATUS AND LABORATORY METHODS

## A SIMPLE THREE-COLOR MIXER USING FILTERED COLORS

The method of color mixing most commonly used in psychological laboratories is the rotating disc or color wheel. Though this method is excellent for many purposes, it has certain disadvantages inherent to the use of surface pigments: (1) mixtures are relatively low in saturation (particularly in the case of yellow produced by a mixture of red and green); (2) complementary mixtures are gray rather than white; (3) brightness can not be controlled independently of room illumination without sacrifice of saturation. These disadvantages can be avoided with mixtures produced with filtered colors.

A device which the author recently built uses an extremely simple principle to provide mixtures of three filtered colors. The general design would seem to be of potential value for such uses as (1) measurement of hue (and saturation) discrimination thresholds in both human beings and animals, (2) detection and analysis of color blindness, (3) specification of pigments in terms of three-color components by matching with three color mixtures, (4) classroom demonstrations of color mixing.

The optical system for accomplishing the trichromatic mixture is shown in Fig. 1. Gelatin color filters



Optical System of Color Mixer
Fig. 1

are mounted between glass, forming a square slide. One upper quadrant of the slide is red, the other quadrant is green, and the bottom half is blue. The filters used in the author's apparatus are Wratten

<sup>15</sup> J. W. MacLeod and J. Gordon, Jour. Path. Bact., 25: 139, 1922. gelatin filters of numbers 29 (red), 47 (green) and 61 (blue).

This color mixing slide is supported on a two-way sliding mount between two lenses. Light from one diffusing screen is focused by the lenses upon another diffusing screen. Both screens are of flashed opal glass. Between the two lenses the light rays are parallel, and whatever colors intercept the light beam at this point become uniformly mixed on the second diffusing screen.

Two adjustments of the color-mixing slide are provided by means of levers moving along graduated scales. One lever moves the slide horizontally, thus changing the red-green proportion in the mixture. The other lever moves the slide vertically, thus adjusting the blue component. More specifically, by the two directions of movement, the slide can be adjusted to transmit any one of the three colors singly, or combinations of the three colors in any proportions. On the stimulus screen one can, therefore, obtain any color around or within the color circle. Due to the lack of homogeneity of the filtered colors, however, the resulting mixtures do not quite equal spectral mixtures in saturation. On the other hand, the mixtures far exceed in saturation those obtained on the color wheel.

That the color-mixing principle described here is successful was adequately demonstrated by a single unit built by the author while at the University of Missouri. Color mixtures of good saturation, moderately high brightness and uniform distribution were obtained.

The apparatus is particularly recommended for the demonstration of color-mixing principles. If intended for this purpose, the apparatus should be built so that the stimulus screen can be removed easily and the separate colors being mixed thus exposed to view.

WALTER F. GRETHER

SAN ANTONIO AVIATION CADET CENTER, SAN ANTONIO, TEXAS

#### **BOOKS RECEIVED**

BLACKWOOD, OSWALD. General Physics. Illustrated. Pp. viii + 622. John Wiley and Sons. \$3.75. SOPER, FRED L. and D. BRUCE WILSON. Anopheles Gambiae in Brazil. Illustrated. Pp. xviii + 262. The Rockefeller Foundation.

WILEY

## WILEY BOOKS

for your V-12 and AST courses

## INTRODUCTORY COLLEGE CHEMISTRY

By H. G. DEMING, Professor of Chemistry, University of Nebraska. Second Edition Completely Revised in Collaboration with B. CLIFFORD HENDRICKS, Professor of Chemistry, University of Nebraska.

A course for beginners. In this book the simplest of the important principles of chemistry are interwoven in an interesting way with the story of how chemistry grew and developed and a vivid description of the part that it plays in modern life and industry. The final chapter deals with the chemistry of food and nutrition. Simpler and more concise than either the "General Chemistry" or the "Fundamental Chemistry."

Second Edition: 521 pages; 6 by 9; \$3.00

W

### FUNDAMENTAL CHEMISTRY

By HORACE G. DEMING.

An elementary book for college classes. It reveals the spirit of modern chemistry and gives a brief survey of its achievements as a science. Its purpose is to show how a chemist thinks about chemical problems of everyday life. It is remarkable for the care that is taken in defining and explaining words, thus leading the student gradually to acquire an extensive and accurate technical vocabulary.

756 pages; 6 by 9; \$3.50

### GENERAL CHEMISTRY

By HORACE G. DEMING.

Substantiality and exactness together with its popular manner of presentation have made this textbook one of the few permanently outstanding general chemistries. It may be used equally well for beginners and by students with previous knowledge of chemistry, as arrangement of material permits somewhat different sequence of topics in the two cases. Briefer and simpler than the author's "Fundamental Chemistry," and gives more attention to the applications of chemistry in modern industry.

Fourth Edition: 769 pages; 51 by 81; \$3.50

# LABORATORY MANUAL OF COLLEGE CHEMISTRY: Elementary Course

By HORACE G. DEMING.

This manual is designed for use in the elementary course in college chemistry. Each of the forty-eight experiments consists of several parts, from among which a choice may be made according to the length of the laboratory period. At frequent intervals, from the very first, laboratory problems are included for which the student must devise his own experimental procedure, based on experience gained in the exercises just preceding.

268 pages;  $8\frac{1}{2}$  by 11; \$1.75

# EXERCISES IN GENERAL CHEMISTRY AND QUALITATIVE ANALYSIS

By HORACE G. DEMING and SAUL B. ARENSON, Professor of Inorganic Chemistry, University of Cincinnati. Offers a wide range of experiments, with accurate and clearly written directions and comments. The first part of the book presents exercises including inorganic preparations, verifications of the quantitative laws in chemistry, fundamental concepts such as oxidation and reduction, physico-chemical measurements, photography and the sensitivity of qualitative tests, etc. The latter part deals with systematic qualitative analysis of inorganic mixtures.

Fourth Edition: 326 pages; 5½ by 8½; \$1.80

JOHN WILEY & SONS, Inc., 440-4th Ave., New York 16, N. Y.