

cluding chemists. A second group consists of units which normally fill necessary civilian needs which continue during war and which are shared by the armed forces. Such are the groups producing food. Their problems are multiplied and intensified because of the war. This group of industries also needs more chemists than in peacetime. The third group consists of those producing goods which are essential neither to the armed forces nor to the civilian production army. Such units have already largely shifted over to war work. This shift is continuing at an increasing rate. Here again the need for technical help, especially of a chemical nature, is increased many fold. For instance, a peacetime company making ten-cent automatic pencils probably needs a minimum of chemical help. On the other hand, when it shifts over to making machine gun parts it certainly can not get along with less help of this kind.

Whenever we do find a chemist doing "business as usual" it is not something which we have to "tolerate" but something for which we can be thankful. This is because of the fact that, as a nation, we are hardly more than ten per cent. into total war. As we get fully into the war effort we shall need more chemists than can possibly be found or produced. Thus, the few chemists who are not yet fully in the war effort constitute our only chemical reserve. It is indeed too bad that this reserve is so small.

F. C. WHITMORE

SCHOOL OF CHEMISTRY AND PHYSICS,  
THE PENNSYLVANIA STATE COLLEGE

## d-TUBOCURARINE CHLORIDE AND CHOLINE ESTERASE

MANY have reported inhibition of choline-esterase by crude curare preparations, and recently Harris and Harris<sup>1</sup> have found that 0.016 mg of a partially purified curare preparation<sup>2</sup> will inhibit 85 per cent. of the choline-esterase activity of 0.5 cc of human serum. All preparations of impure curare, including "Intocostrin," examined in this laboratory were found to possess this choline-esterase inhibitory property. However, our recent experiments have shown that the chemically pure substance d-tubocurarine chloride is devoid of inhibitory action upon the choline-esterase activity of dog serum. It was found that 0.866 mg of d-tubocurarine chloride injected into the femoral artery of a barbital anesthetized 10 kilogram dog caused the complete curarization of the skeletal musculature. There was no change in blood pressure. *In vitro* experiments using concentrations of d-tubocurarine chloride up to approximately 29,000 times the calculated concentrations used in the *in vivo* experiments were entirely without inhibitory activity upon choline-esterase activity of dog serum. From these experiments it appears that the curare-action and the effects upon choline-esterase of the impure preparations of curare are not necessarily related.

A. R. MCINTYRE

RAY E. KING

DEPARTMENT OF PHYSIOLOGY AND PHARMACOLOGY,  
UNIVERSITY OF NEBRASKA COLLEGE OF  
MEDICINE, OMAHA

## SPECIAL CORRESPONDENCE

### ODE ON NEWTON'S THEORY OF GRAVITATION BY EDMOND HALLEY

WHEN Newton opened the printed copy of the first edition of the "Principia" in 1687, he found prefixed to the text a poem dedicated to his work and signed "Edm. Halley." It was at Halley's urging that the book had been written, and Halley had seen it through the press and paid the expenses of publication. He was then thirty years of age, Newton forty-four. The poem was in Latin like the *Principia* itself. It consisted of forty-eight hexameters; apart from its dedication, it bore no title.

The verses were reprinted in the second edition of the "Principia" (1713), which was edited by Roger Cotes, the mathematician, and seen through the press by Richard Bentley, the classical scholar. Bentley, without Halley's consent, altered some of the lines and omitted others. In the third edition (1726), edited by Henry Pemberton, a physician and scientist, Halley's original text was restored in most places; but a few of Bentley's changes were retained and some

additional ones introduced. This text, like that of the second edition, did not meet with Halley's approval.

No further edition of the "Principia" appeared during the lifetime of Newton, who died in 1727. Both Halley and Bentley died in 1742; so that the two hundredth anniversaries of their deaths coincide approximately with the three hundredth of the birth of Newton and the death of Galileo, and the four hundredth of the death of Copernicus.

The text of all three editions of Halley's poem is printed in Stephen Peter Rigaud's "Historical Essay on the First Publication of Sir Isaac Newton's *Principia*," Oxford, 1838 (pages 57-59); in Sir David Brewster's "Memoirs of the Life, Writings and Discoveries of Sir Isaac Newton," two volumes, Edinburgh, 1855 (volume 1, pages 457-459); and in Eugene Fairfield MacPike's "Correspondence and Papers of Edmond Halley," Oxford, 1932 (pages 203-206).

<sup>1</sup> *Proc. Soc. Exp. Biol. and Med.*, 46: 619, 1941.

<sup>2</sup> Intocostrin. There is no evidence that the choline-esterase inhibitory activity of Intocostrin constitutes a clinical hazard.