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THE NEW SOIL SCIENCE¹

By Dr. P. E. BROWN

PROFESSOR OF SOILS, IOWA STATE COLLEGE

Soil science or pedology, as it is now coming to be called, is not new. It is centuries old, as may be readily deduced from an examination of many old books and records. Recently, however, the study of soils has taken on a different aspect and we now have a new concept of soil science. It is this modern, rather recent development of the subject which is referred to as "the new soil science."

As is the case with many of our present-day sciences, the beginnings of soil science lie buried in the dim mists of antiquity. Just when and where it might be said to have originated can not be determined. Perhaps the first observations were made in the Garden of Eden. The writings of Moses indicate that agriculture is as old as man. Isaac and Jacob certainly knew how to grow good crops and they probably

¹ Address of the retiring president of the Iowa State Chapter of Sigma Xi.

gathered many facts about soils. Observations on soils certainly have been made from the time man began to learn the art of using them for the production of crops.

According to Herbert Spencer, use is the underlying cause of the development of all science. In the early stages, the practical phases always receive the most attention, because of the struggle by man toward the utilization of nature. This has been particularly true, and naturally so, in the case of the various agricultural sciences.

But there is another force which is perhaps even more powerful than use in bringing about the evolution of sciences. It is the "unconscious struggle of our natures for the acquisition of abstract knowledge or for the discovery of the laws of phenomena." In any branch of science, this force soon becomes the more significant, and the present status of knowledge

This relationship is shown in the following partial formulas:

In all other respects the two aglucones are structurally identical. These conclusions have been reached from the study of the following series of substances.

Gitoxigenin on isomerization by alkali is converted into isogitoxigenin.¹ On saponification of the lactone group of the latter substance, the salt of isogitoxigeninic acid results, which can be oxidized by hypobromite to isogitoxigenic acid. When the latter is treated with concentrated hydrochloric acid, the remaining tertiary hydroxyl is replaced by chlorine with the formation of chloroisogitoxigenic acid. Simultaneously stereo-isomerization occurs under the influence of the reagent on some center of asymmetry in the molecule. The chlorine atom in this acid can be removed under certain conditions as hydrochloric acid with the production of anhydro isogitoxigenic acid.

It was hoped that on catalytic hydrogenation this anhydroacid (as the ester) would absorb one mol of hydrogen with the formation of a substance identical, or at least isomeric, with the previously described isodigitoxigenic acid obtained from isodigitoxigenin. The reaction, however, took an abnormal course in that two mols of hydrogen were consumed. Investigation showed that not only was the double bond hydrogenated but the lactone group was cleaved with the formation of a saturated acid in accordance with the following scheme:

On saponification this half ester readily yielded the dibasic acid.

Following a number of unsuccessful attempts the identical dibasic acid was obtained also from digitoxigenin through the following steps. Isodigitoxigenin² after saponification was oxidized by hypobromite to isodigitoxigenic acid. On treatment with

¹ W. A. Jacobs and E. L. Gustus, J. Biol. Chem., 1928, 79: 553; 1929, 82: 403.

² W. A. Jacobs and E. L. Gustus, J. Biol. Chem., 1928, 78: 573.

concentrated hydrochloric acid the latter was isomerized to γ-isodigitoxigenic acid. When treated with acetic anhydride and acetyl chloride a reaction occurred, which involved cleavage of the lactone group and formation of a substituted succinic anhydride while simultaneously the newly uncovered hydroxyl group was removed as water.³ The secondary hydroxyl group elsewhere in the molecule was also acetylated. When this anhydro anhydride acetate (acetate of anhydro-γ-digitoxenoldiacid)

was treated with methyl alcohol containing one per cent. of hydrochloric acid, the succinic anhydride group was converted into the half ester. Catalytic hydrogenation of the resulting unsaturated half ester gave rise to the saturated substance, the acetate and half ester of γ -digitoxanoldiacid. On saponification γ -digitoxanoldiacid was produced and this substance proved to be identical with the above described dibasic acid obtained from gitoxigenin. This conclusion was substantiated by the comparison of the neutral dimethyl esters prepared from both acids as well as of the stable half esters which resulted on partial saponification of the latter.

WALTER A. JACOBS EDWIN L. GUSTUS

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BOOKS RECEIVED

The American Annual of Photography. 1930. Vol. XLIV. Edited by Frank R. Fraprie. Pp. 272. Illustrated. The American Photographic Publishing Company. \$1.50.

APPLETON, A. B. Laboratory Guide to Vertebrate Dissection. Pp. xix + 152. Cambridge University Press. 6s. Conklin, Edmund S. Psychology of Religious Adjustment. Pp. xiv + 340. Macmillan. \$2.00.

DAMPIER WHETHAM, W. C. D. A History of Science.
Pp. xxi + 514. Cambridge University Press. 18s.

DANIELS, FARRINGTON, J. HOWARD MATHEWS and JOHN WARREN WILLIAMS. Experimental Physical Chemistry. Pp. xvi+475. 132 figs. McGraw-Hill. \$3.50. DICKSON, LEONARD E. Introduction to the Theory of

DICKSON, LEONARD E. Introduction to the Theory of Numbers. Pp. viii + 183. University of Chicago Press. \$3.00.

Pollard, A. F. C. The Kinematical Design of Couplings in Instrument Mechanisms. Pp. 64. 25 figs. Adam Hilger, London. 4s. 6d.

Pool, RAYMOND J. Flowers and Flowering Plants. Pp. xx+378. 191 figs. McGraw-Hill. \$3.50.

RAMSAY, SIR WILLIAM, and F. G. DONNAN, Editors. The Effects of Moisture on Chemical and Physical Changes. Pp. xii + 235. 44 figs. 50 tables. Longmans, Green. \$5.00.

³ W. A. Jacobs and E. L. Gustus, J. Biol. Chem., 1929, 84: 183.

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