

tween quiz sections was manifested, and the different sections came as units to boost their representatives. The students had had three or four weeks in which to prepare for the contest, and nearly all of them had been working hard for it. Our instructors are all agreed that the students participating derived great benefit from this match.

In addition to these benefits, the contest brought out the fact that our chemical nomenclature is not yet above reproach. A few instances of ambiguity might be cited: Sodium thiosulphate, $\text{Na}_2\text{S}_2\text{O}_3$, is sometimes named sodium hyposulphite and so labelled by a few manufacturers of chemicals. The latter name, however, is represented by the formula $\text{Na}_2\text{S}_2\text{O}_4$. Potassium fluosilicate and potassium silicofluoride are both used to represent the same substance. Potassium sulphocyanide and potassium thiocyanate are two names in use for KCNS. Then, again, we say hydronitric acid, or triazotic acid, or azoamide, when we mean a substance with the composition N_3H .

If these spelling bees were to be adopted by a considerable number of educational institutions it would doubtless tend to unify chemical nomenclature so that finally we should have one name only to represent a chemical compound having a definite composition. Spelling matches of this sort could also be profitably arranged between classes in organic chemistry, mineralogy and perhaps other departments of science. The contests appeal to students because they combine the elements of sport and competition. The benefits derived therefrom are incalculable, and we are now planning to make the chemical spelling match an annual event at the West Virginia University.

C. A. JACOBSON

MORGANTOWN, WEST VIRGINIA

SCIENTIFIC BOOKS

Proteins and the Theory of Colloidal Behavior.

By DR. JACQUES LOEB, member of the Rockefeller Institute for Medical Research. New York: McGraw-Hill Book Co., 285 pp. 1922.

In this volume the author has collected the results of his extensive investigations upon the

properties of protein solutions and has attempted to found upon them a general theory of colloids. The book falls naturally into two sections. The main argument in the first half is that proteins are amphoteric electrolytes and that consequently, when hydrogen ion concentrations are duly measured and considered, proteins are found to combine with acids and alkalies according to the stoichiometrical laws of classical chemistry. This argument is illustrated and supported by numerous tables and diagrams. In the second part of the book the conclusion is established that all of the experimental results recorded can be logically explained upon the basis of Donnan's theory of membrane equilibria.

The far-reaching significance of the author's contentions may be summarized in the statement that, if justified, they dispose of colloid chemistry as a special branch of the science, with laws different from those of general chemistry. This does not, as is pointed out in the preface, detract from the importance of colloidal behavior for physiological and technical problems, but it completely changes the theoretical treatment of the subject.

A revolution in our current conceptions of colloidal solutions is hereby threatened, equal in importance to that brought about by van't Hoff and Arrhenius a generation ago in the field of crystalloidal solutions, and it seems probable, from certain reviews that have already appeared, that the battle between the new and the old points of view will be waged with equal bitterness. It is interesting to note in this connection that the veteran fighter Armstrong, now president of the Society of Chemical Industry in England, went out of his way in his recent Messel Memorial Lecture at Glasgow to refer to Loeb's "praiseworthy efforts to raise the character of the proteins from mere indeterminate lumps of jelly to a status of definite materials behaving in a simple and definite, orderly manner, if only put under comparable conditions." Since, however, he indulged in the course of the same address in his customary diatribes against the Scandinavian Ikon Arrhenius and his High Priest Ostwald, remarking that "hydrogen ion concentration is pure gibberish," his conversion to Loeb's

theory is obviously incomplete. As he so ap-
positely puts the case himself: "We all have
partially permeable intellects."

At this stage, indeed, it is altogether pre-
mature to express an opinion as to the outcome
of the struggle. What is certain is that Loeb
has made, in this volume, a brilliant thrust
which his adversaries will find it difficult to
counter. There are many points of detail
in his experimental work which will curdle
the blood of any analytical chemist, yet it ap-
pears on close examination that the errors in-
troduced are, after all, insufficient to affect
the main issue. The opponents of Loeb's views,
in any case, cannot restrict themselves to at-
tacking the weak points of his presentation;
he has already succeeded so far as to put them
definitely on the defensive. To quote from
his own preface: "Any rival theory (of col-
loidal behavior) which is intended to replace
the Donnan theory must be able to accomplish
at least as much as the Donnan theory, *i. e.*,
it must give a quantitative, mathematical and
rationalistic explanation of the curves express-
ing the influence of hydrogen ion concentra-
tion, valency of ions, and concentration of
electrolytes on colloidal behavior; and it must
explain these curves not for one property
alone but for all the properties, electrical
charges, osmotic pressure, swelling, viscosity,
and stability of solution, since all these
properties are affected by electrolytes in a
similar way."

This quotation may be supplemented by an-
other, from the final page of the book, in-
dicating the importance of Loeb's work out-
side of chemistry. "If Donnan's theory of
membrane equilibria furnishes the mathematic-
al and quantitative basis for a theory of col-
loidal behavior of the proteins, as the writer
believes it does, it may be predicted that this
theory will become one of the foundations
upon which modern physiology will have to
rest."

Every so-called colloidal chemist will evi-
dently be forced to read Loeb's book in self-
defense. Those also who are only indirectly
interested in colloidal phenomena cannot fail
to find it stimulating.

JAMES KENDALL

SPECIAL ARTICLES

MOSAIC CROSS-INOCULATION AND INSECT TRANSMISSION STUDIES

WHETHER or not the plant disease known as
mosaic is transmissible to plants of different
orders, and the rôle of insects as agents in
such transmission, are questions of funda-
mental importance. It is generally held that
mosaic of the Cucurbitaceæ, Solanaceæ and
Leguminaceæ are all quite specific and with few
exceptions transmissible only to species within
the same family. Certain mosaic diseases have
been described indicating that even among spe-
cies within the same family there may be two
or more types of the disease. Allard¹ in 1916
described a specific mosaic disease on *Nicotiana
viscosum* distinct from the mosaic disease of
Nicotiana tabacum. Jagger²⁻³ in 1917 and
1918 reports three specific mosaic diseases of
the cucurbits. The tendency has thus been to
divide mosaic into types which are distinct in
their host range.

As opposed to the evidence indicating that
there are a number of types of mosaic which
are specific to a narrow host range, we have
evidence showing that mosaic will cross to
species belonging to other families and orders.
Jagger⁴ in 1918 published results of cross-
inoculation studies where he succeeded in trans-
ferring mosaic from the Cucurbitaceæ to spe-
cies of two other families of the Order Campa-
nulales. Doolittle⁵ has shown that mosaic of
cucumber is transmissible to *Martynia louisiana*,
a species belonging to the Order Polemoniales.

Cross-inoculation experiments by the writer
have shown that the mosaic diseases of the
Cucurbitaceæ, Solanaceæ and Leguminaceæ are
inter-transmissible. Four petunia plants inocu-
lated with mosaic from crookneck squash be-
came infected while an equal number of checks
remained healthy. The inoculations were made
by inserting mosaic tissue into the stem with a
sterile scalpel. An experiment in which juice
from mosaic plants was inoculated hypo-

¹ *Journ. Agr. Research*, 7: 481-486, 1916.

² *Phytopathology*, 7: 61, 1917.

³ *Phytopathology*, 8: 74-75, 1918.

⁴ *Phytopathology*, 8: 32-33, 1918.

⁵ *U. S. D. A. Bull.* 879, 1-69, 1920.