that three of the Australian colonies are scarcely in a position to undertake this new work: but South Australia is freer, and if it can establish a solar observatory, either at Adelaide or elswhere, a gap in longitude will be satisfactorily filled. The scheme is receiving substantial private support. In addition to the promise of a telescope (a 6-in. Grubb equatorial refractor) from the trustees of the estate of the late Lord Farnham, Dr. W. Geoffrey Duffield, of the Physical Laboratories, Manchester University, has received from Mr. Frank K. McClean the offer of £500 towards the purchase of a large spectroheliograph, on condition that an additional sum of £1000 be privately subscribed towards the same piece of apparatus.

UNIVERSITY AND EDUCATIONAL NEWS

THE University of Toronto is conducting a course in hygiene of twenty lectures. Dean Reeve of the medical faculty will take up the eye and ear; Dr. George R. McDonagh, the nose and throat; Dr. Charles Sheard, contagious and infectious diseases; Dr. Abbott, color blindness, and Dr. William Oldright, general sanitation.

DR. CHAS. C. ADAMS, of the University of Chicago, has been appointed associate in animal ecology at the University of Illinois.

In the faculty of engineering at University College, London, a new lectureship in electrical design has been instituted, to which Mr. Henry Metcalf Hobart has been appointed.

MR. W. JACKSON POPE, F.R.S., professor of chemistry in the University of Manchester, has been elected into the professorship of chemistry, at Cambridge, rendered vacant by the resignation of Professor G. D. Liveing, who has held the chair since 1861. Mr. Pope, who was born in London in 1870, was educated at Finsbury Technical College and the Central Technical College, London. Before going to Manchester he was head of the chemistry department of the Goldsmiths' Institute, London.

DR. HANS SPEMAN, of the University of Würzburg, has been appointed professor of zoology at the University of Rostock.

DISCUSSION AND CORRESPONDENCE THE CHEMICAL FORMULA OF THE MINERAL BENITOITE

In a recent issue of SCIENCE (May 1, 1908) Mr. Edward H. Kraus discusses the recently described mineral benitoite, and suggests a formula which differs somewhat from the one proposed by Professor Louderback in his original paper. The two analyses already reported, and a third which will be published shortly, all show that the empirical formula of the mineral is BaTiSi_sO_s and the most reasonable assumption is that it is made up of the three oxides BaO, TiO₂ and SiO₂. Mr. Louderback's proposal is that the mineral is a very acid titano-silicate of barium, whereas Mr. Kraus suggests that the titanium here plays the part of a base and that therefore the mineral is a double metasilicate of barium and Though it must be admitted that titanium. absolute proof of the correctness of either of these two suggestions is at present scarcely possible it seems to me that the arguments advanced by Mr. Kraus are entirely unsatisfactory and I submit the following objections.

The main argument upon which Mr. Kraus bases his formula is the alleged isomorphism of benitoite with beryl, which mineral is usually regarded as a salt of metasilicic acid. If the most favorable values are chosen the ratios of the a axis to the c axis for the two minerals are 1.4989 and 1.4230, respectively. These figures show an actual difference of more than sixteen per cent. of the magnitude concerned, and even interpreting the law of isomorphism with that degree of looseness which is not uncommon among mineralogists, furnish no evidence upon which to base conclusions as to the molecular structure of the two compounds. The two minerals also differ widely as to form and habit; one is trigonal and the other holohedral. Further, many of the illustrations which Mr. Kraus cites as examples of isomorphism are open to serious question; galena and argentite most certainly can not be called isomorphous merely because they both crystallize in cubes.

The occurrence of benitoite in rock formations which are of a basic character, upon