mixture (mass action.) Pure palladium, even in thin leaves, is not easily soluble in nitric acid, whilst all the other platinum metals are perfectly insoluble if in a moderately compact condition. Palladium cannot be isolated by agitation with mercury from a solution which, along with the platinum metals contain base metals, such as copper, lead, &c., since the mercury precipitates, not merely the palladium, but all the other platinum metals, forming probably amalgams. From the platinum metals thus precipitated by mercury, metal free from mercury cannot be obtained by distillation and subsequent ignition, since a part of the mercury forms a stable compound with the platinoids. —THEODOR WILM.

GLYCERIN.—Notwithstanding the low price which prevails for almost every description of raw produce and manufactured goods, there are a few articles which form notable exceptions. Perhaps one of the most remarkable of these is refined glycerin, which, within the last two years, has advanced from about £30 to £130 per ton avoirdupois for 30° B. This enormous advance is due partly to increased consumption, diminished production and the influence of speculation working on a market devoid of stocks. In view of the present position of the article and the prospect of a continuance of high prices for a considerable time to come, the attention of soapmakers is now being turned to the utilization of their waste "leys," and various new processes for recovering the glycerin contained in these liquors have lately been tried with more or less successful results. Apart from minor impurities, waste soap "leys" are generally found to contain glycerin, carbonate of soda or caustic soda, chloride of sodium, gelatin and albumen. One of the processes for recovering the glycerin which promises to be the most economical and the most successful begins with concentrating the liquor until the salts con-

tained therein begin to crystallize. The liquid is then cooled and filtered to rid it of gelatin and albumen. It is afterwards made to absorb carbonic acid, which precipitates bi-carbonate of soda, and which is separated from the liquor in the usual way. After undergoing this process the liquor is then made to absorb gaseous hydrochloric acid until what remains of carbonate of soda has been converted into chloride, and further, until all, or almost all, the chloride of sodium has been precipitated and separated from the liquor contains water, glycerin and hydrochloric acid. The acid is then evaporated entirely and absorbed in water for using afresh. The dilute glycerin remaining can be purified by filtering it through animal charcoal or by concentrating and distilling it in the usual way.

AN INDUSTRIAL AND TECHNOLOGICAL MUSEUM. —An Industrial and Technological Museum of a very comprehensive character is in course of organization at Sydney. It is to include animal, vegetable and mineral produce in the crude and in the manufactured states: waste products, of whatsoever origin, foods with their constituents, and that necessary shadow side of the picture, their adulterations; educational appliances; sanitary apparatus and systems, models, plans, machinery, etc., for mining; agricultural machinery and manures; models, drawings, and descriptions of patents; a department of economic entomology; ethnological specimens, etc. One remark in the prospectus may call up a smile. The museum is intended to occupy a similar position to the South Kensington Museum. This might be construed to mean that it is to occupy a site as far out of the way of merchants, manufacturers, patentees, etc., as possible. We need scarcely say that the project has our best wishes.

METEOROLOGICAL REPORT FOR NEW YORK CITY FOR THE WEEK ENDING JULY 9, 1881.

Latitude 40° 45' 58"; Longitude 73° 57' 58"; height from ground, 53 feet; above the sea. 97 feet; by self-recording instruments.

	BAROMETER.							THERMOMETERS.													
	MEAN FOR THE DAY.	MAXII	MAXIMUM. MINIMUM.			́ ме	MEAN. MAXIMUM.						MINI	MAXI'N							
JULY.	to	to	Time.	Reduced to Freezing.	Time.	Dry Bulb,	Wet Bulb.	Dry Bulb.	Time.	Wet Bulb.	Time.	Dry Bulb.	Time.	Wet Bulb.	Time.	In Sur					
Sunday, 3 Monday, 4 Tuesday, 5 Wednesday, 6 Thursday, 7 Friday, 8 Saturday, 9	29.974 29.861 29.850 29.828 29.983 29.927 30.059	30.100 29,898 29,906 29,902 30,002 29,998 30,090	o a. m. o a. m. g a. m. 12 p. m. g a. m. 12 p. m. 12 p. m.	29.800 29.790 29.750 29.892 29.836	12 p. m. 5 p. m. 7 p. m. 4 a. m. 12 p. m. 6 a. m. o a. m.	71.6 77.3 82.7	67.3 67.6 70.7 75.0 69.3 65.0 66.6	87 85 85 88 83 71 80	5 p. m. 1 p. m. 5 p. m. 4 p. m. 3 p. m. 7 a. m. 4 p. m.	72 75 79 71 69	6 p. m. 1 p. m. 5 p. m. 7 p. m. 3 p. m. 7 a. m. 4 p. m.	70 68 74 70 64	5 a. m. 5 a. m. 3 a. m. 5 a. m. 5 a. m. 2 p. m. 6 a. m.		5 a. m. 5 a. m. 3 a. m. 5 a. m. 5 a. m. 2 p. m. 6 a. m.	136. 138. 147.					

grees _____68.8 degrees.

at 7 pm 6th, 79. at 5 am 3rd, 60.

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WIND.						HYGROMETER.						CLOUDS.			RAIN AND SNOW.				
	DIRECTION.		N.	VELOCITY IN MILES. SQR. FEET.		FORCE OF VAPOR.			RELATIVE HUMIDITY.			CLEAR, O OVFRCAST, IO			DEPTH OF RAIN AND SNOW IN INCHES,				
JULY.	7 a. m.	2 p. m.	9 p. m.	Distance for the Day.	Max.	Tıme.	7 a.m.	2 p. m.	9 P. m.	7 a.m.	2 p. m.	9 p. m.	7 a.m	2 p. m.	9 p. m.	Time of Begin- ing.	01	Dura- tion. h. m.	Amount of water
Ionday, 4 Juesday, 5 Vednesday, 6 Jhursday, 7 Jriday, 8	w. n.e.	e. n. e. s. s. e. n. n. e. s. e. e. n. e.	w. s.w. n.n.w. s.e. n.n.e.	104 94 141 180	$2\frac{1}{2}$ 4 $2\frac{1}{2}$ 3 $6\frac{3}{4}$ 1	1.20 pm 6.30 pm 8.30 pm 2.40 pm 2.00 pm 11.40 am 4.00 pm	.581 .622 .690 .641 .682	.644 .650 .836	.585 .654 .717 .773 .631 .536 .622	69 72 85 70 76 90 89	49 85 59 65 56 94 73	55 85 70 71 80 84 85	3 cir. s. 9 cu. 2 cir. 10	2 cir. cu. 9 cir. cu. 3 cir. cu. 4 cir. cu. 3 cir. cu. 10 7 cir. cu.	9 cu. 5 cir. cu. 2 cir. cu. 10 10		5.15 pm	5.45 0.15 5.00	.80 .04 06

DANIEL DRAPER, PH. D.

Director Meteorological Observatory of the Department of Public Parks, New York.